

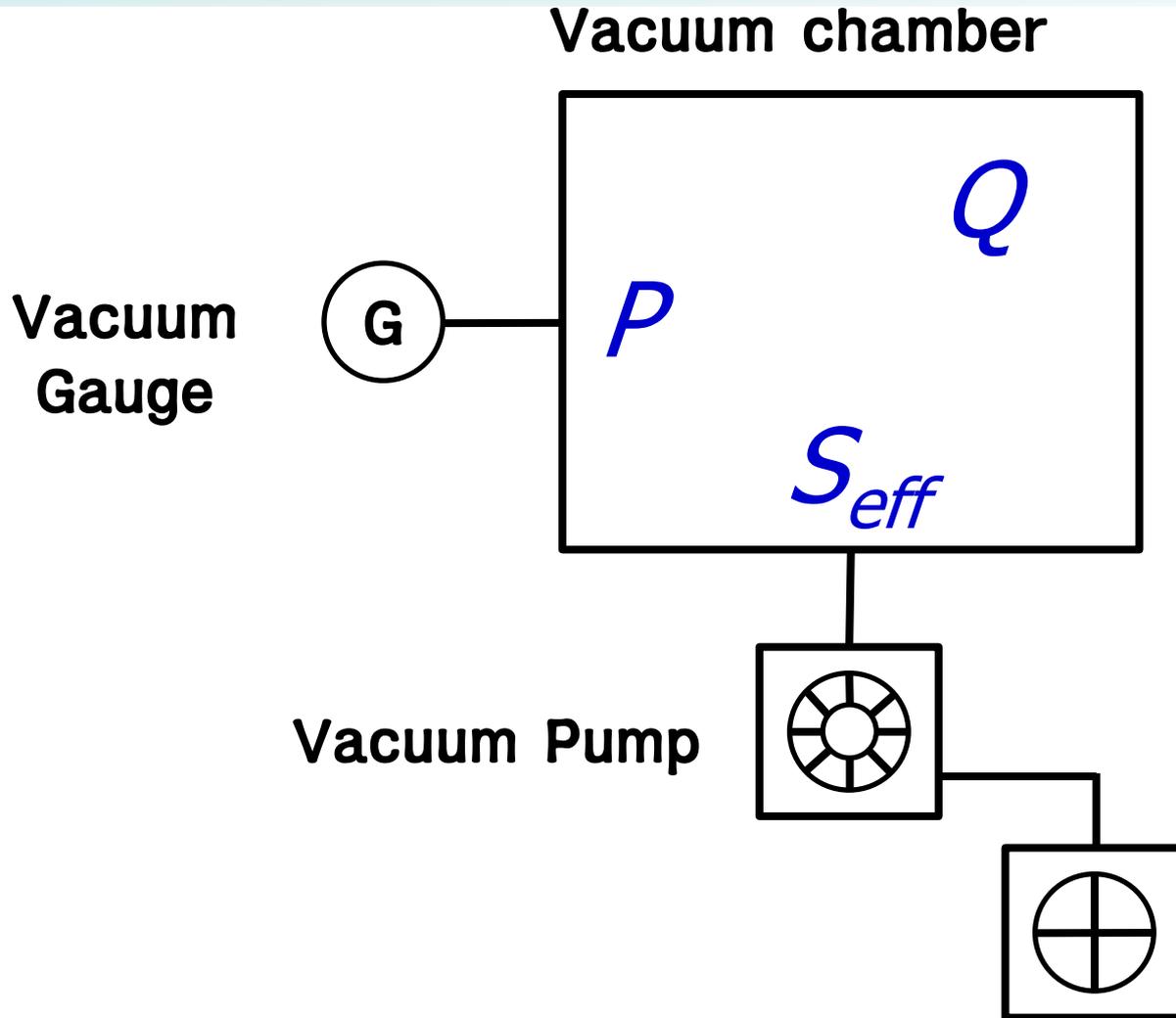
진공재료의 종류와 부품의 특성 및 제조 관리

2019.06.21

하태균
포항가속기연구소

1. 진공 재료의 특성 이해

2. 진공 재료의 성능 향상



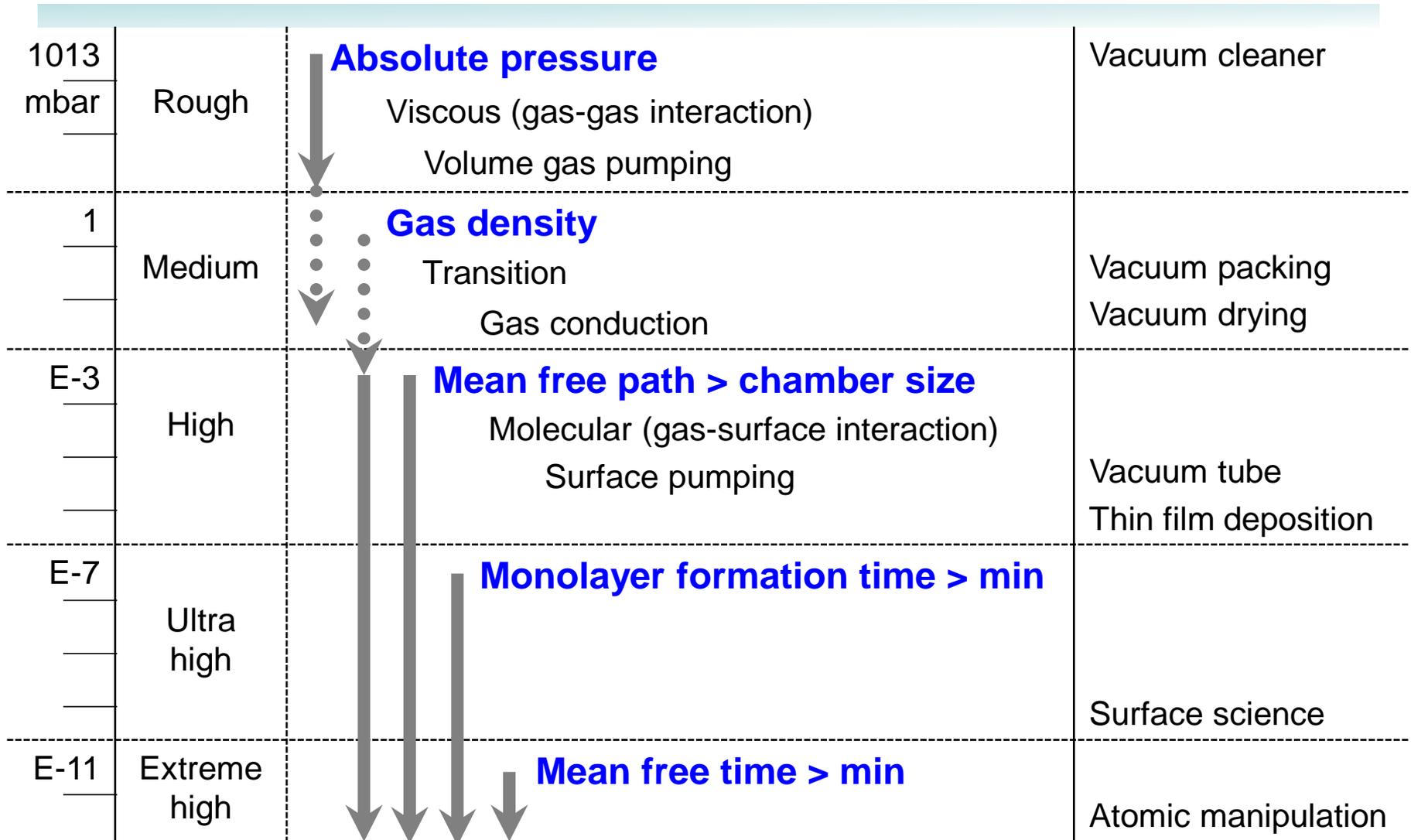
왜 재료가 중요한가?

System base pressure

$$P \text{ (mbar)} = \frac{Q}{S_{eff}} \frac{\text{(mbar } \ell/\text{s)}}{(\ell/\text{s})}$$


$$\frac{1}{S_{eff}} = \frac{1}{S} + \frac{1}{C}$$

목표진공도에 따라 재료가 달라짐



A photograph of a laboratory or industrial setting. A person in a blue shirt is crouching and working on a large, complex piece of equipment, likely a vacuum furnace or similar high-temperature processing system. The equipment is mounted on a white base and has various components, including a large cylindrical chamber and various pipes and gauges. The background shows a clean, industrial environment with blue and white walls. The text "Materials for Vacuum technology" is overlaid in the center of the image in a bold, black font.

**Materials
for
Vacuum technology**

진공재료 선택

재료 선택시 고려할 내용

- 얻고자 하는 진공도는?

관련 항목

- 적절한 기체방출률
 - ✓ 전처리 필요여부
 - ✓ 세척 방법
- 고유 증기압
- 적절한 녹는점과 끓는점
- 재료의 누출률
- 적절한 투과율
- 불순물 기체 함유량
- 깨끗한 표면

진공재료 선택

재료 선택시 고려할 내용

- 강도는 충분한가?
- 제작은 용이한가?
- 기밀방법은?
- 사용환경과 수명은?

관련 항목

- 1 기압차 유지
 - ✓ 허용 응력과 허용 변형
- 기계가공성, 성형성
- 적절한 접합법
- 금속 또는 엘라스토머
- 내부식성(내화학적성)
- 내방사성
- 열변형(적절한 열팽창 거동)

진공재료 선택

재료 선택시 고려할 내용

- 사용온도와 수명은?
- 투자율은?
- 전기전도도는?
- 제작 비용은?
- 재료 수급성은?

관련 항목

- 금속 재료 또는 플라스틱?
- 극저온 또는 고온?
- 높은 내열피로성
- 전자기 특성에 영향
- 전도체 또는 부도체
- 국내 또는 국외 수입
- 구입 기간

Vacuum Materials

Steels

Stainless Steel

Aluminum (alloy)

Copper (alloy)

Other metals

Ceramics

Glass

Plastics

Steels



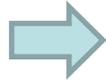
Steels

- Mild/structural steels
 - ✓ Carbon < 0.3 %
 - ✓ outgassing rate
 - ~~$q > (20 \sim 200) \times STS$~~
 - $q_{rd} > (20 \sim 200) \times STS, q_{H_2} \lesssim STS$
 - ~~HV compatible (10^{-6} mbar), endless emission of CO~~
 - UHV compatible, $RegGas_{ST} \sim RegGas_{STS}$
 - ✓ Weldable
 - ✓ Easy to corrode
 - ✓ Needs anti-corroding coating
 - ✓ Magnetic
 - ✓ Shielding material for magnetic field

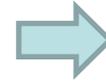
Steels

- Mild/structural steels
 - ✓ S235, S355, S20C
 - ✓ UHV compatible
 - plate, pipe, rod
 - $\sim 10^{-11}$ mbar
 - $\lesssim 5 \times 10^{-12}$ (mbar l/s cm²) after bake
 - ✓ MV, RV compatible
 - Cast parts; pump and valve housing
 - $\sim 10^{-3}$ mbar

Steels



- Anti-corrosion
 - ✓ ~300°C
 - ✓ Process



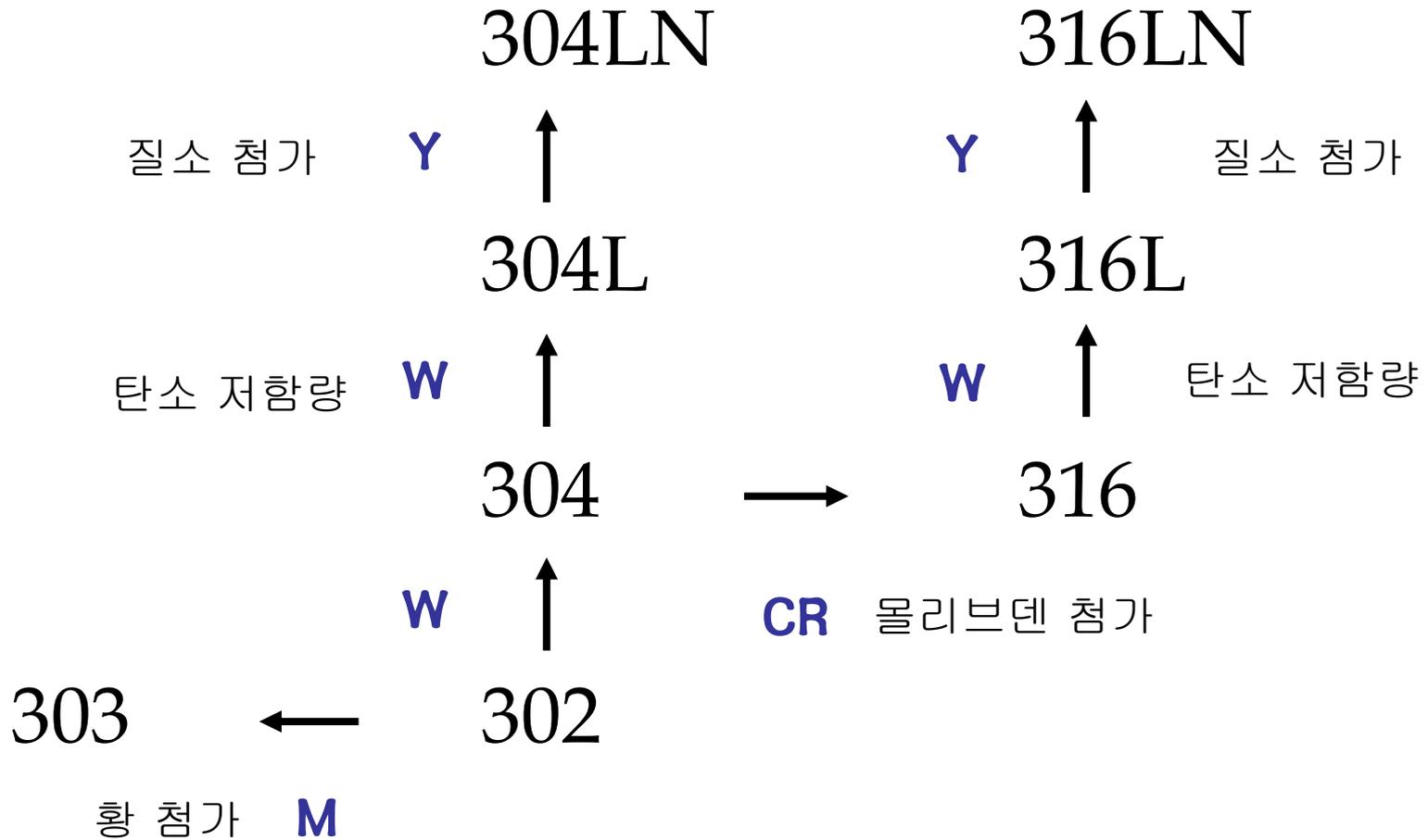
Stainless steels

- ~~• Low q~~
 - ~~— UHV~~
 - ~~— XHV~~

Stainless steel

- AISI 304/316
 - ✓ Austenitic
 - High strength
 - Non-magnetic (but, not entirely)
 - Good weldability
 - ✓ Corrosion resistance
 - During vacuum processing and bakeout ($\sim 300^{\circ}\text{C}$)
 - ✓ UHV/XHV compatible
 - $< 10^{-11}$ mbar
 - $(2\sim 6) \times 10^{-12}$ (mbar l/scm²) after bake
 - $< 1 \times 10^{-13}$ (mbar l/scm²) after special treatment

Stainless steel



18-8 Steel Family

Stainless steel

- Role of ingredients
 - Cr(10%) Resistance to oxidation
 - Ni(8%) Austenitic structure/ Anticorrosion
 - Mo Accelerates passivating film formation
 - W Mechanical resistance at high temperature
 - Ti During welding and cycles stabilizes the austenitic structure
 - N Mechanical characteristics

Stainless steel

- 303

- ✓ 19% Cr, 10% Ni, 0.15% C, 0.15% S

- ✓ A free machining stainless

- ✓ Not suitable for UHV applications

- ✓ Emission of sulfur at higher temperature

- ✓ (The q of 303 can be lowered to 10^{-13} mbar l/s·cm² by a combination of fabrication and post treatments such as bakeout.)

- ✓ Welding is a problem due to the evolution of sulfur during welding causing porosity.

Stainless steel

■ 304

- ✓ 18% Cr, 8% Ni, 0.08% C
- ✓ Most common materials used in vacuum technology

■ 304L

- ✓ 18% Cr, 8% Ni, 0.03% C
- ✓ One of common steels used in vacuum technology
- ✓ Less carbide precipitation
- ✓ Cleaner machining and better welds than 304
- ✓ Lower mechanical properties than 304

Stainless steel

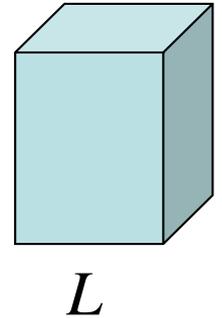
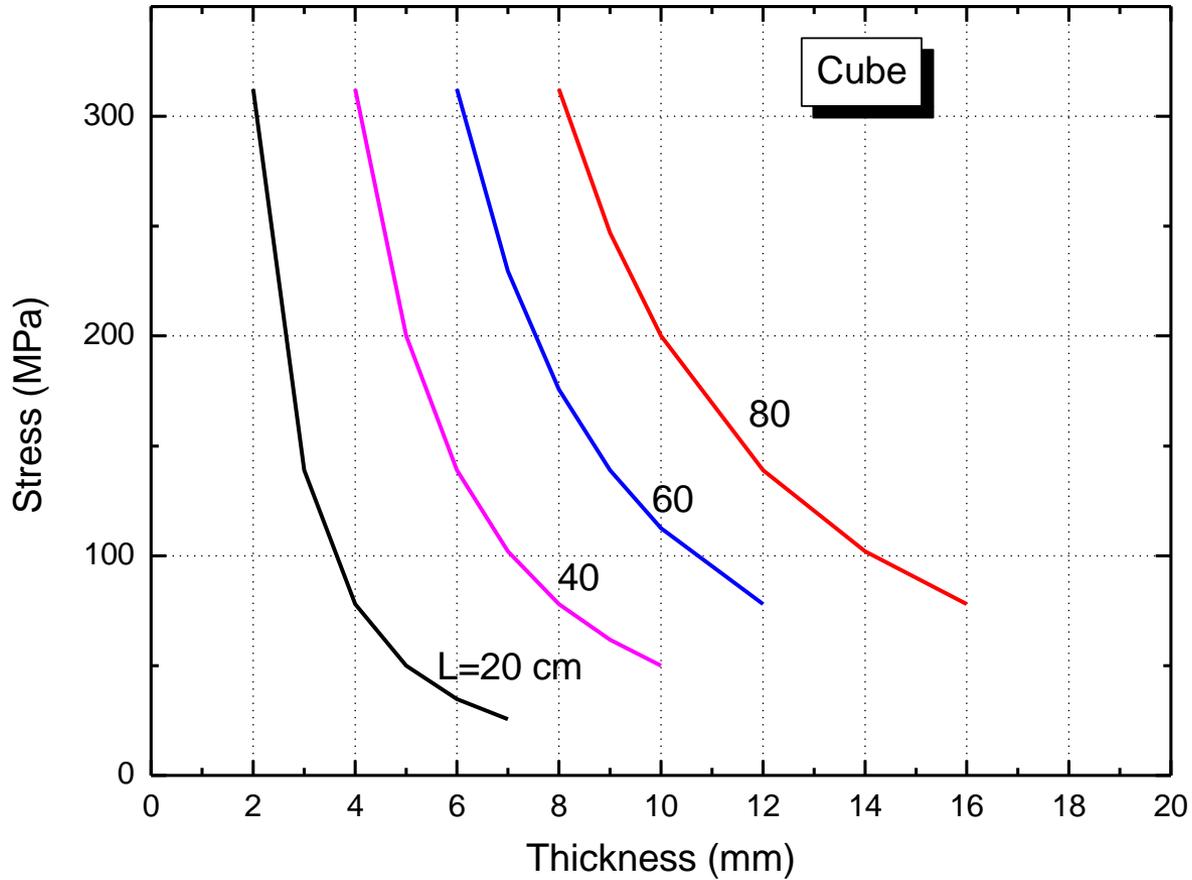
■ 316L

- ✓ 18% Cr, 14% Ni, 0.03% C, 3% Mo
- ✓ Stabilized with molybdenum to prevent carbide precipitation in the weld zone
- ✓ Lower q than 304(L)
 - $\sim 2 \times 10^{-12}$ (mbar l/scm²) after bake
 - $< 5 \times 10^{-14}$ (mbar l/scm²) after special treatments
- ✓ Used where chemical compatibility is a concern.
- ✓ Low-magnetic stainless steels ($\mu_r < 1.02$)
 - Suitable for analyzers, accelerators
 - Heat treatment; (750 ~ 1050°C)

Mechanical properties

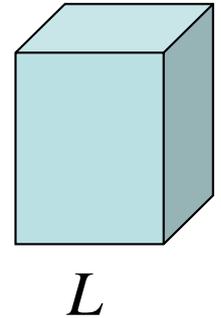
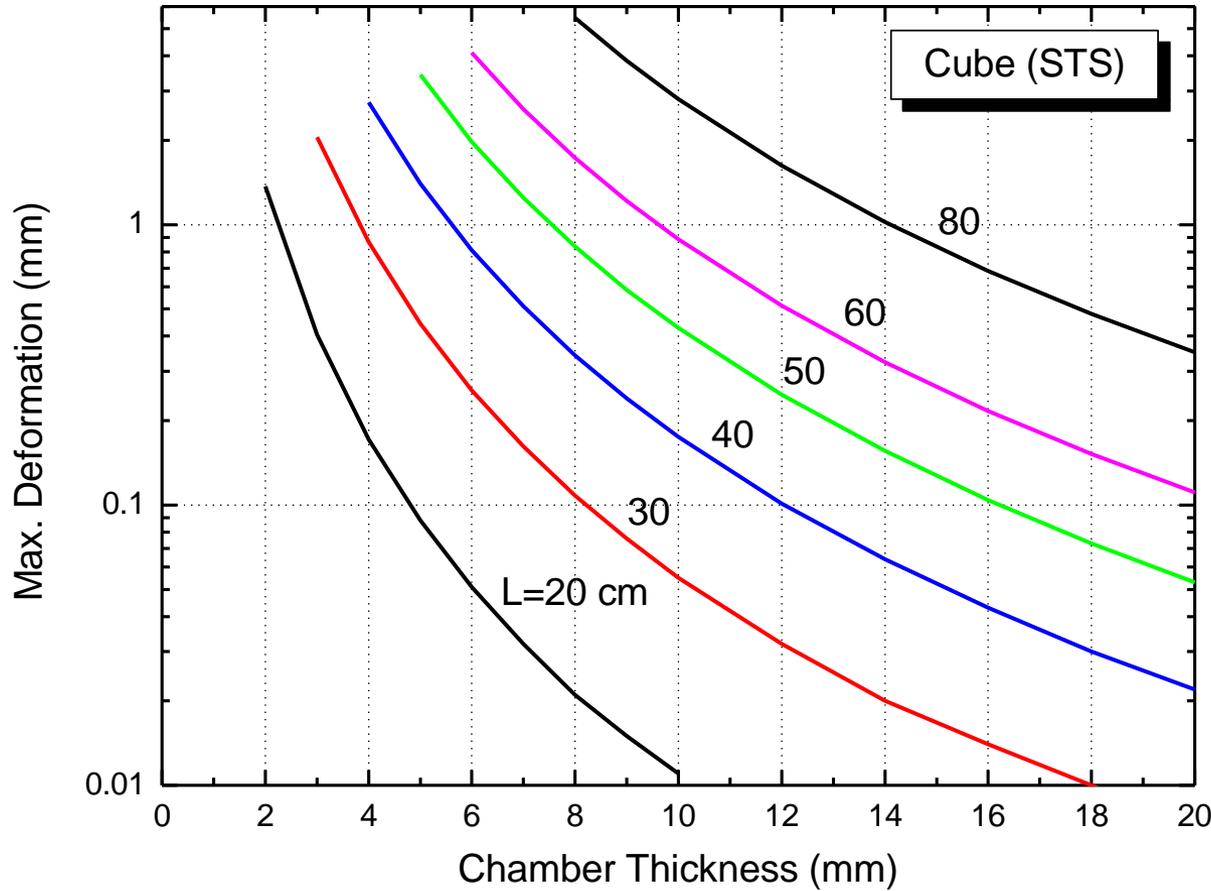
	Yield strength (0.2%)	Tensile strength
	MPa	MPa
316	206.8	517.1
316L	172.4	482.6
A6061-T6	241	289.4
A6063-T5 - T6	110	152

Stress : Cube



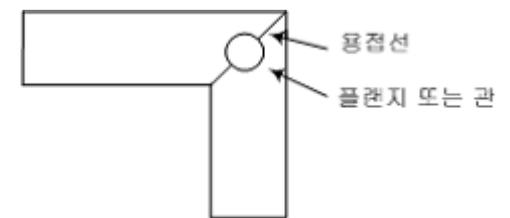
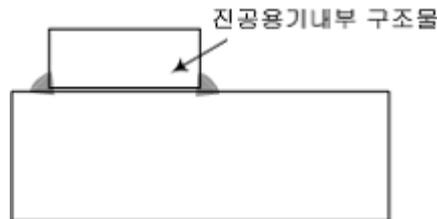
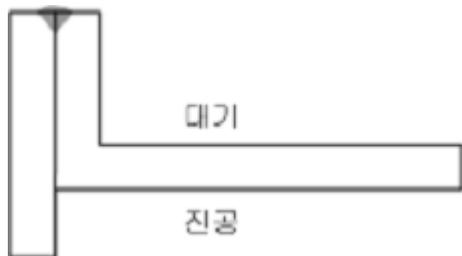
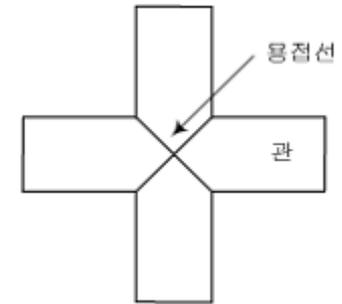
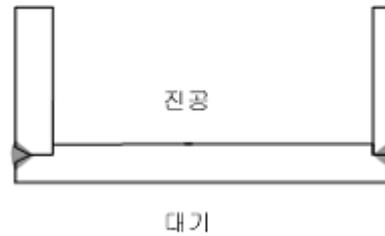
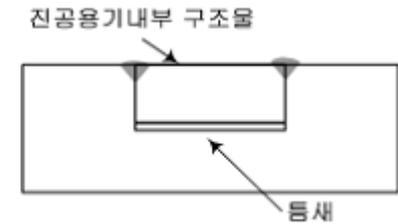
→ “진공공학(인상별 외)”

Deformation : Stainless steel



→ “진공공학(인상률 외)”

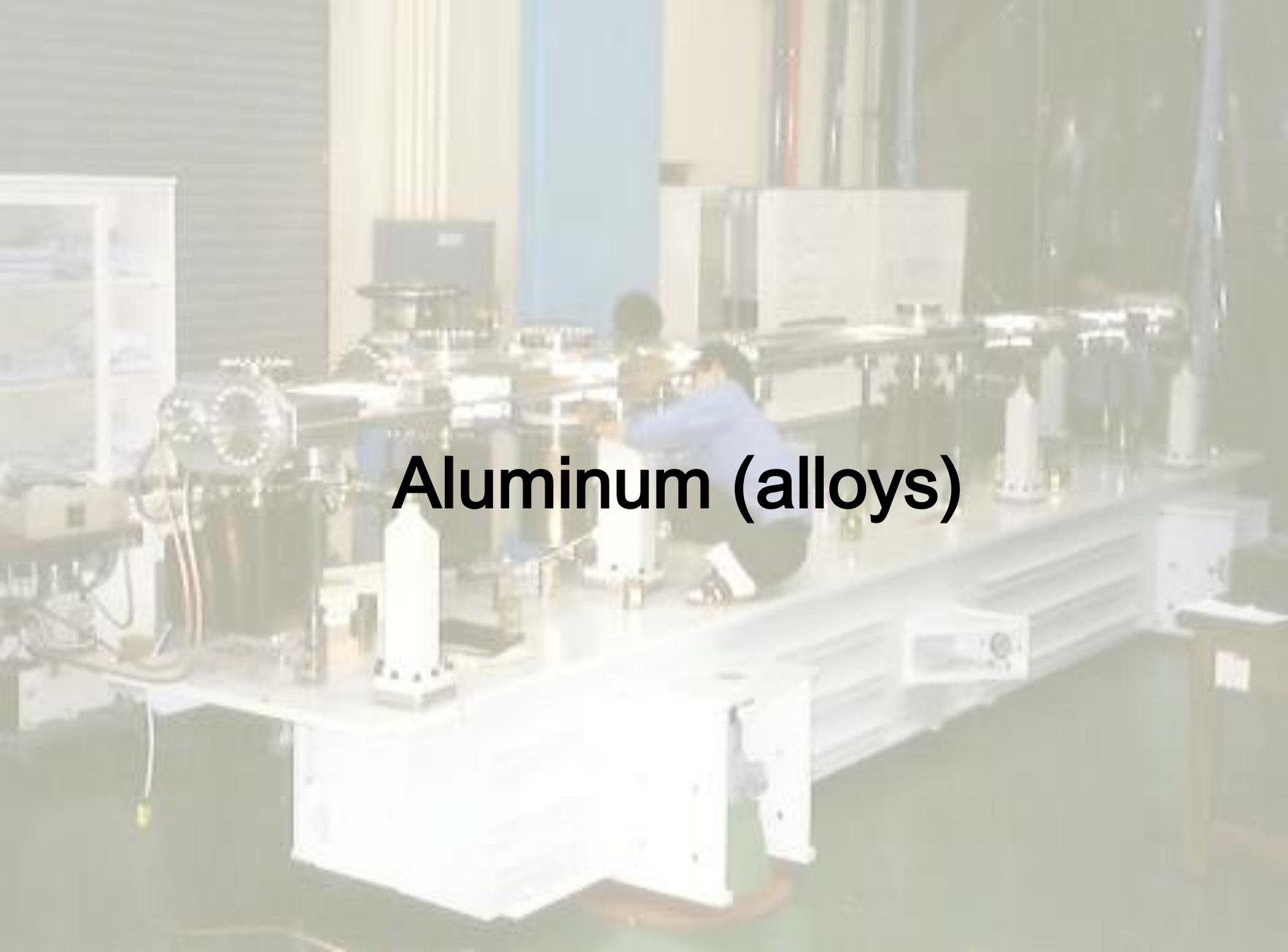
피해야 하는 용접



→ “초고진공 용접핸드북(KRISS)”

용접 품질



A photograph of a laboratory or industrial setting. A person in a blue shirt is crouching at a workstation, working with various pieces of equipment. The workstation is a long, white, L-shaped table. On the table, there are several large, cylindrical components, possibly part of a testing machine or a furnace. There are also some smaller instruments and containers. The background shows a large room with blue and white walls and some structural elements. The text "Aluminum (alloys)" is overlaid in the center of the image.

Aluminum (alloys)

Aluminum (alloys)

- 1xxx pure aluminum
 - ✓ > 99% Al by weight
 - ✓ **A1050**; suitable for metal gaskets
- 2xxx copper alloys
 - ✓ Duralumin; once the most common aerospace alloys (they were susceptible to stress corrosion cracking and are increasingly replaced by 7000 series in new designs.)
 - ✓ **A2219**; suitable for Conflat flange (weldable)

Aluminum (alloys)

- 3xxx manganese alloys
 - ✓ A3004; suitable for vacuum bellows
- 5xxx magnesium alloys
 - ✓ easy to machine, higher strength, good weldability
 - ✓ A5083; for a large scale chamber
- 6xxx magnesium and silicon alloys
 - ✓ Easy to machine and extrude
 - ✓ A6063; most common materials in vacuum technology
 - ✓ A6061; one of the most vacuum materials
 - ✓ A6060; extrusion

Aluminum (alloys)(진공용)

재 료	처 리	주요 합금성분 (%)		용 도
2219	T87, T852	Cu 5.8-6.8	Mn 0.2-0.4	플랜지
3004		Mn 1-1.5	Fe 0.25	벨로우즈
5052		Mg 2.2-2.8	Fe 0.4	진공용기, 벨로우즈
5083	H321	Mg 4-4.9	Mn 0.4-1	진공용기
6061	T5, T6	Mg 0.8-1.2	Si 0.4-0.8	진공용기, 벨로우즈
6063	T5, T6	Mg 0.45-0.9	Si 0.2-0.6	진공용기
6263		Mg	Si	진공부품 (티, 크로스,엘보)
6951	T6	Mg 0.4-0.8	Si 0.2-0.5	벨로우즈

Tempering

- -F As fabricated
- -H Strain hardened (cold worked) with or without thermal treatment
 - -H1 Strain hardened without thermal treatment
 - -H2 Strain hardened and partially annealed
 - -H3 Strain hardened and stabilized by low temperature heating
- Second digit A second digit denotes the degree of hardness
 - -HX2 = 1/4 hard
 - -HX4 = 1/2 hard
 - -HX6 = 3/4 hard
 - -HX8 = full hard
 - -HX9 = extra hard
- -O Full soft (annealed)

- -T Heat treated to produce stable tempers
 - T5 Cooled from hot working and artificially aged (at elevated temperature)
A6061 160°C 18 h
 - (ex) 160°C for 18 h -> ambient cooling
 - 170°C for 12 h -> ambient cooling
 - 175°C for 8 h -> ambient cooling
 - 205°C for 3 h -> ambient cooling (1.5 h)
 - T6 Solution heat treated and artificially aged
 - (ex) ~ 180°C - 220°C for 6 h - 1h -> water cooling
- -W Solution heat treated only.

Aluminum (alloys)

- **A6063-T5(T6)(T0)**
 - ✓ The most widely used aluminum alloy
 - ✓ The most common materials in vacuum technology
 - ✓ Low outgassing rates
 - $< 5 \times 10^{-13}$ (mbar l/scm²) after bake ($< 180^{\circ}\text{C}$)
 - ✓ Easy to forming using **extrusion**, machining and welding
 - ✓ Alloy retains its strength after welding
 - ✓ ConFlat[®] flanges are made from A2219 with knife edge coated with TiC (TiN, CrN).
 - ✓ With A1050 metal gaskets

- Mainly used in UHV and forelines
- Bakeout temperature
 - Max 180°C
 - In-general 100 - 150°C
- Melting point; 660°C ($P_{\text{vapor}}=10^{-8}$ mbar)
- Entirely non-magnetic
- High thermal and electrical conductivity

Aluminum (alloys)

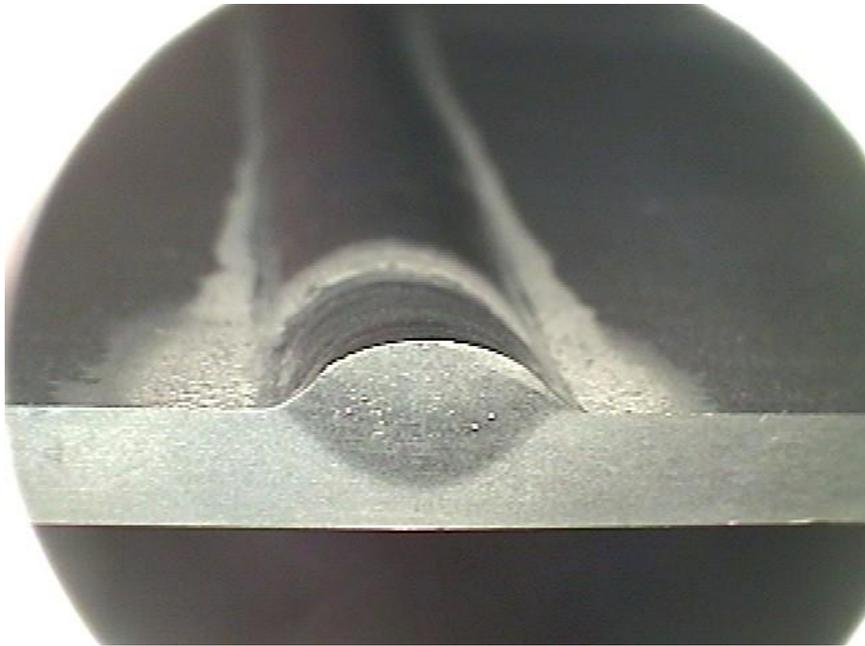
- **A6061-T5(T6)**
 - ✓ The most widely used aluminum alloy
 - ✓ Low outgassing rates
 - ✓ $< 5 \times 10^{-13}$ (mbar l/scm²) after bake ($< 180^{\circ}\text{C}$)
 - ✓ Easy to forming using *machining* and welding
 - ✓ Alloy retains its strength after welding
 - ✓ ConFlat[®] flanges are made from A2219 with knife edge coated with TiC (TiN, CrN).
 - ✓ Large chambers which would be difficult to heat treat in to T6 condition are often made from **5083**.

Aluminum (alloys)

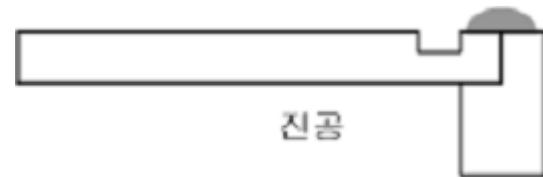
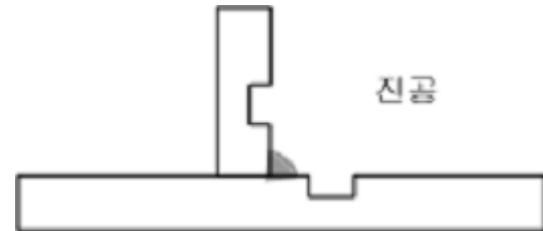
- Initial outgassing rate is higher($\sim 5x$) than that of SST.
 - ✓ The desorption rate of water vapor from the surface of aluminum is slower than stainless steels which gives it different initial pump down characteristics.
- **Anodizing**
 - ✓ a common surface treatment.
 - ✓ produces **hard inert** surface, but q becomes higher, $\sim x10$.
 - ✓ Not suitable for UHV applications
- ✓ Surface oxidation
 - ✓ Suitable for UHV and XHV applications

알루미늄 용접

- Al alloys require special attention to both weld design and weld technique.



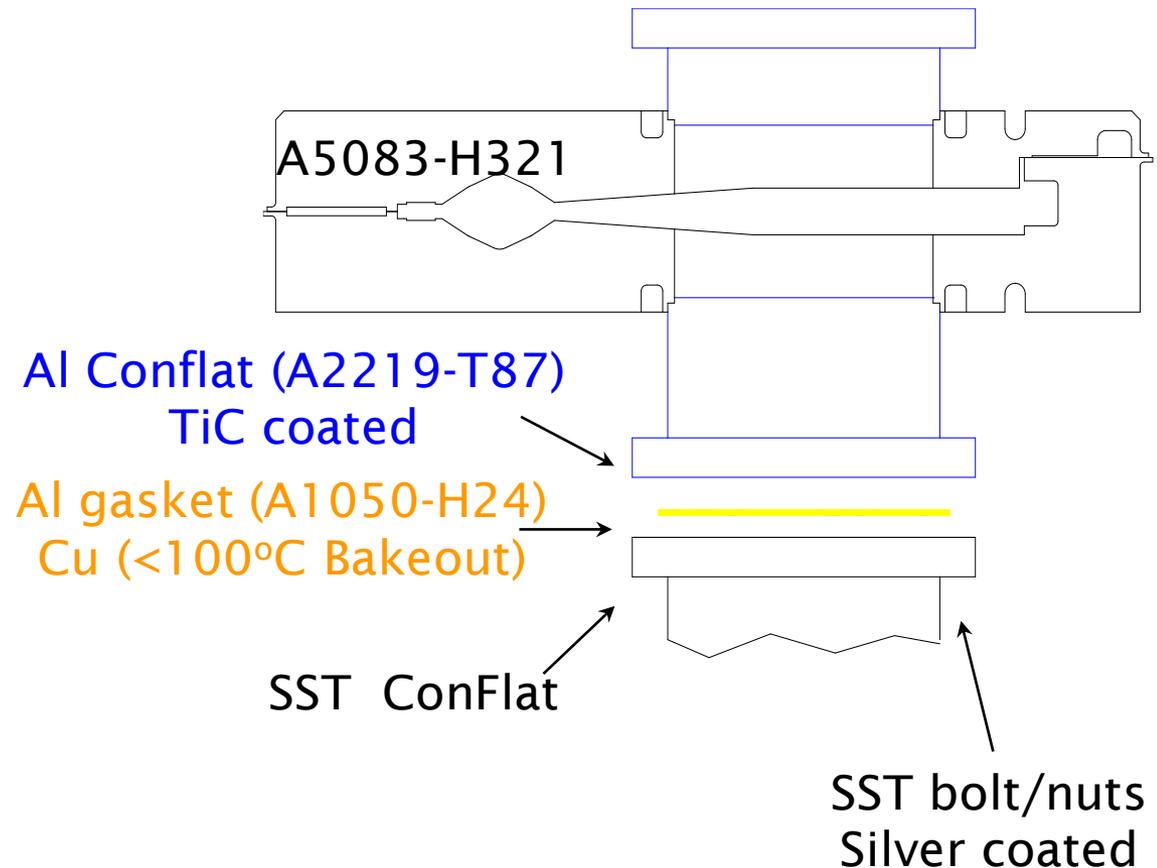
ac-TIG welding
(with filler metal)



→ “초고진공 용접핸드북
(KRISS)”

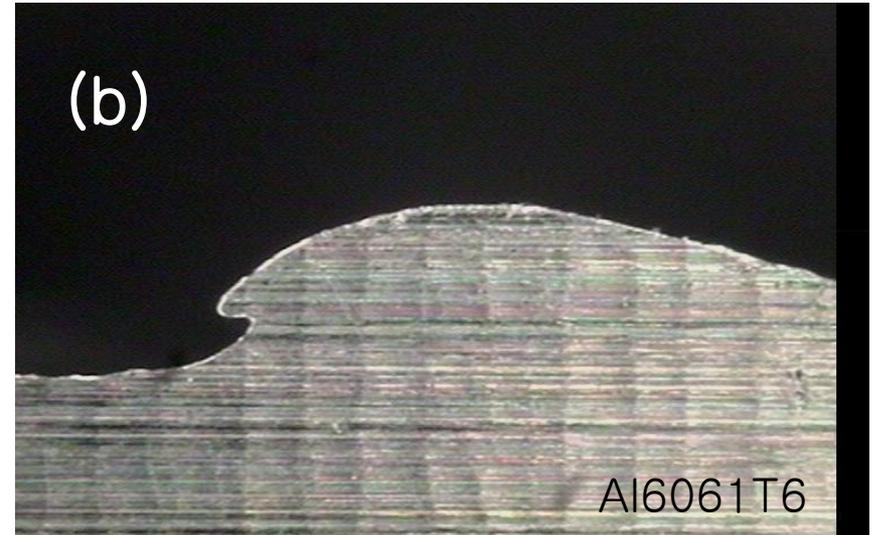
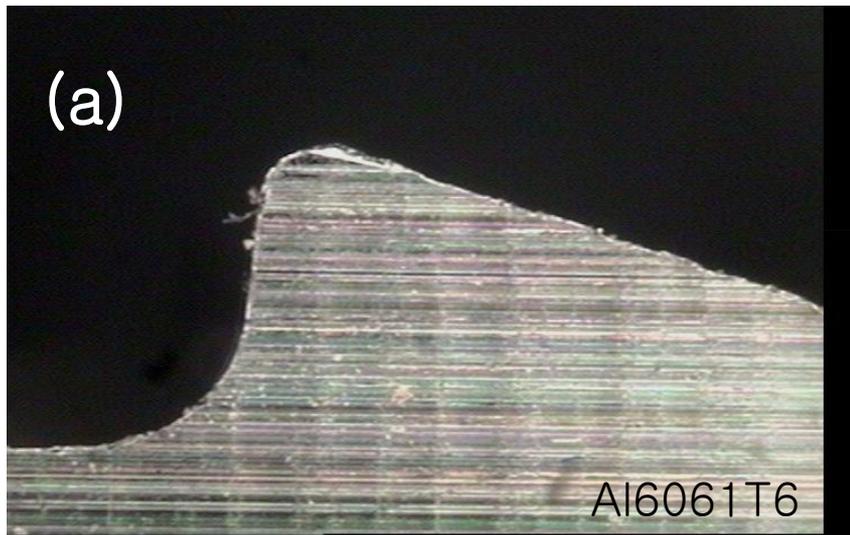
Vacuum Seals

- Al/SST hybrid ConFlat system



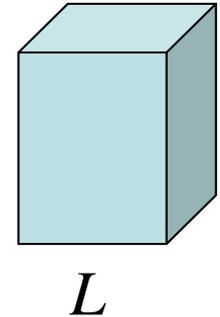
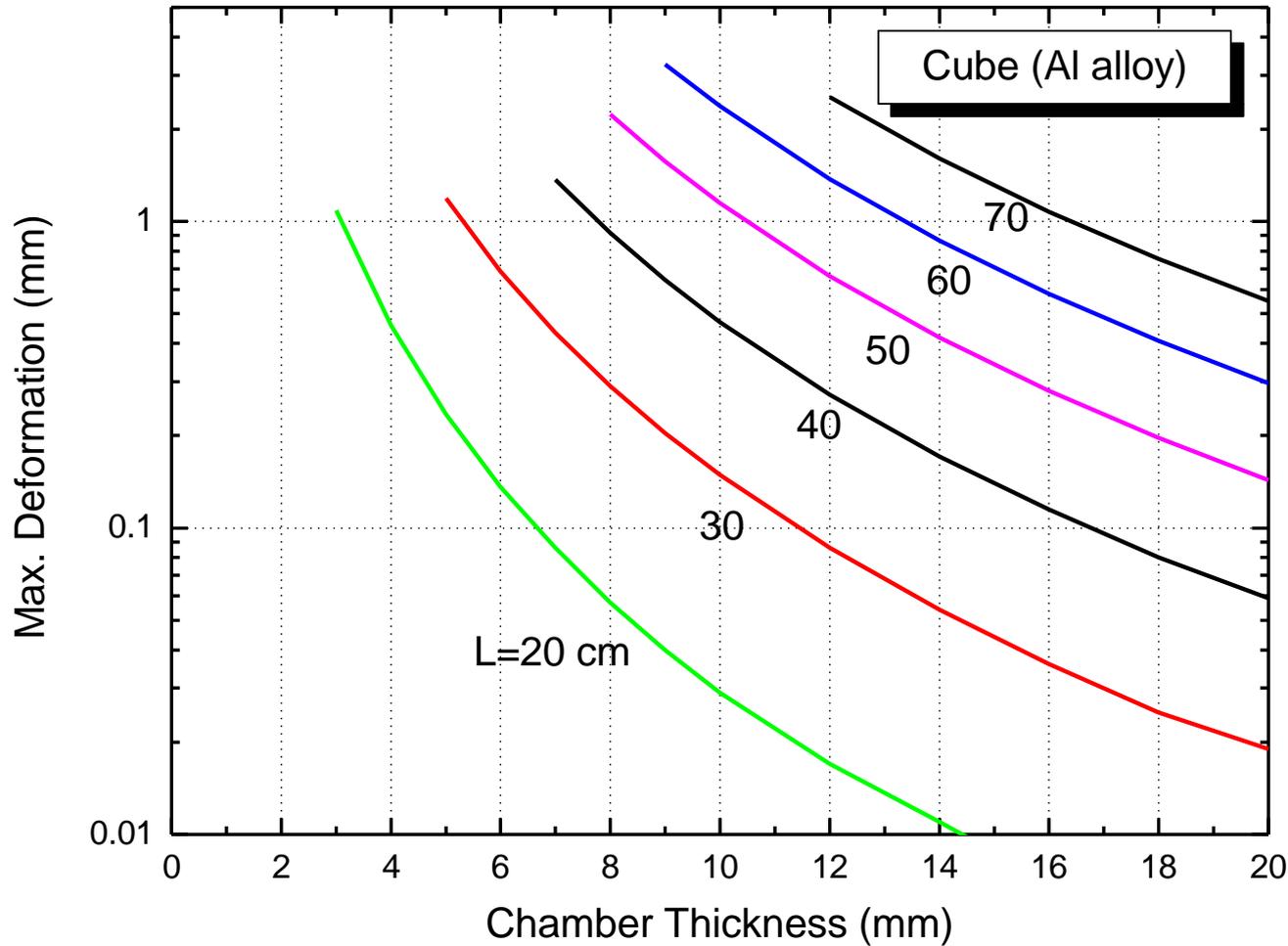
Vacuum Seals

- Al/SST hybrid ConFlat system



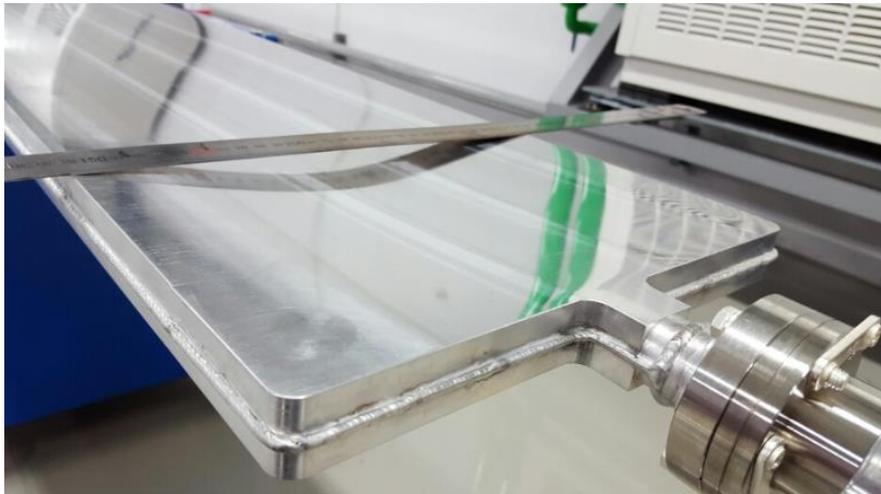
Knife edge of Al flange (before use (a) and after use (100 times) (b)]

Deformation : Al alloy



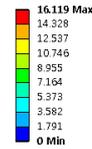
$t \ll L$
 $\delta < t$

Deformation : Al alloy

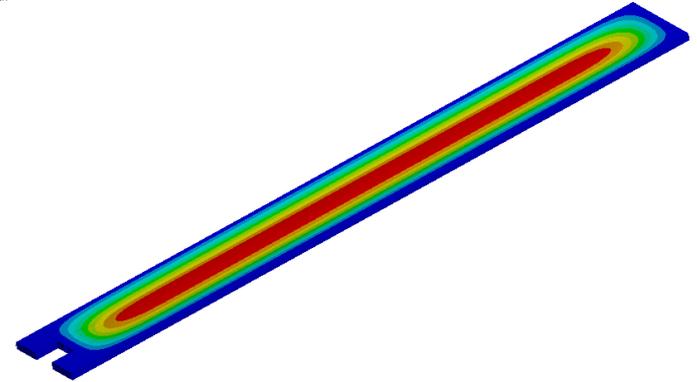


(Thickness=3 mm, width = 270 mm)

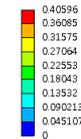
C: Static Structural
Total Deformation
Type: Total Deformation
Unit: mm
Time: 1



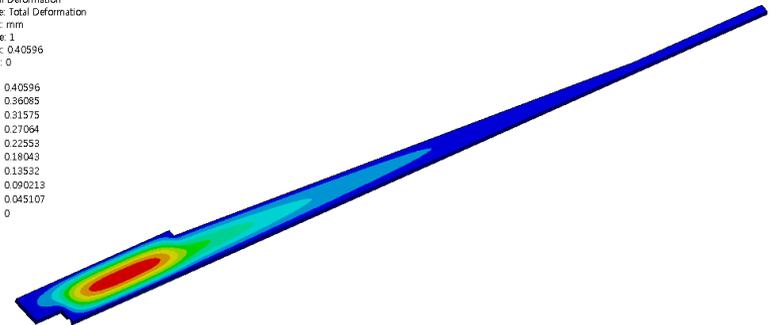
16 mm deformation (3T)



E: St
Total Deformation
Type: Total Deformation
Unit: mm
Time: 1
Max: 0.40596
Min: 0



0.4 mm deformation (4T)

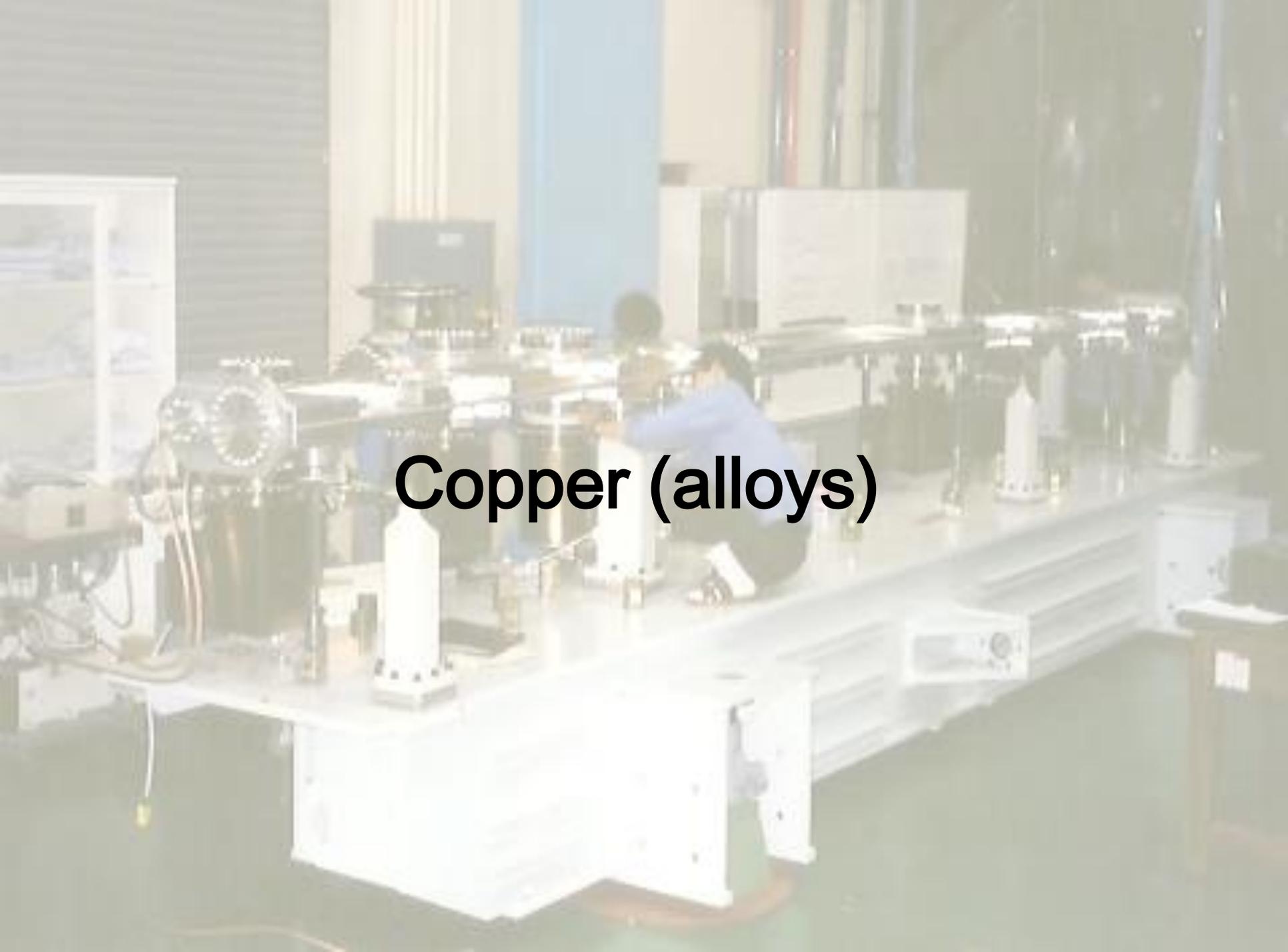




SST - Al 비교

		Stainless steel	Aluminum alloy
Vacuum characteristics	Outgassing rate	depends on surface treatments	
	Preinstallation bakeout	950℃	< 210℃
	<i>In situ</i> bakeout	150 - 450℃	< 150℃
Mechanical characteristics	Mechanical strength	higher (×1.5)	
	Thermal expansion coefficient		higher (×2)
	Thermal conductivity		higher (×15)
	Shaping by machining		easier
	Shaping by extrusion		easier
	Shaping by welding	easier	
	Quality of demountable seals	better	
Other characteristics	Magnetic property	not entirely non-magnetic	completely non-magnetic
	Residual radioactivity		lower

Copper (alloys)

A photograph of a laboratory or workshop. In the center, a person wearing a blue shirt and dark pants is crouching at a long white workbench. The workbench is cluttered with various pieces of scientific equipment, including what appears to be a large piece of machinery on the left, several glass vessels or containers, and other smaller instruments. The background shows a room with blue and white walls, possibly a cleanroom or a specialized lab. The overall lighting is somewhat dim, and the image has a slightly grainy quality.

Copper (alloy)

- High **thermal and electrical conductivity**
 - Suitable for electrical feedthrough
 - Suitable for thermal/radiation absorber
 - Suitable for cryogenic applications
- Hydrogen embrittlement
 - For HV and UHV, coppers(alloys) with oxygen free or reduced oxygen contents are required.
- Bakeout; Up to 300°C in vacuum
- Cold welding; **OFHC gasket** for ConFlat flanges
- Joining techniques; brazing, soldering, welding
- Outgassing rate; $\sim 10^{-9}$ mbar liter/sec cm²

Copper alloy

Brass(Zinc alloy) and **Bronze**(Tin alloy)

- High vapor pressure at high temperatures
(Used in vacuum systems where $T < 100^{\circ}\text{C}$)
 - Easy to machine
 - Cheap
- Commonly used in rough and high vacuum chambers and fixtures.
- Common joining techniques: Soldering
- Outgassing rate
 $\sim 10^{-7}$ mbar liter/sec cm^2

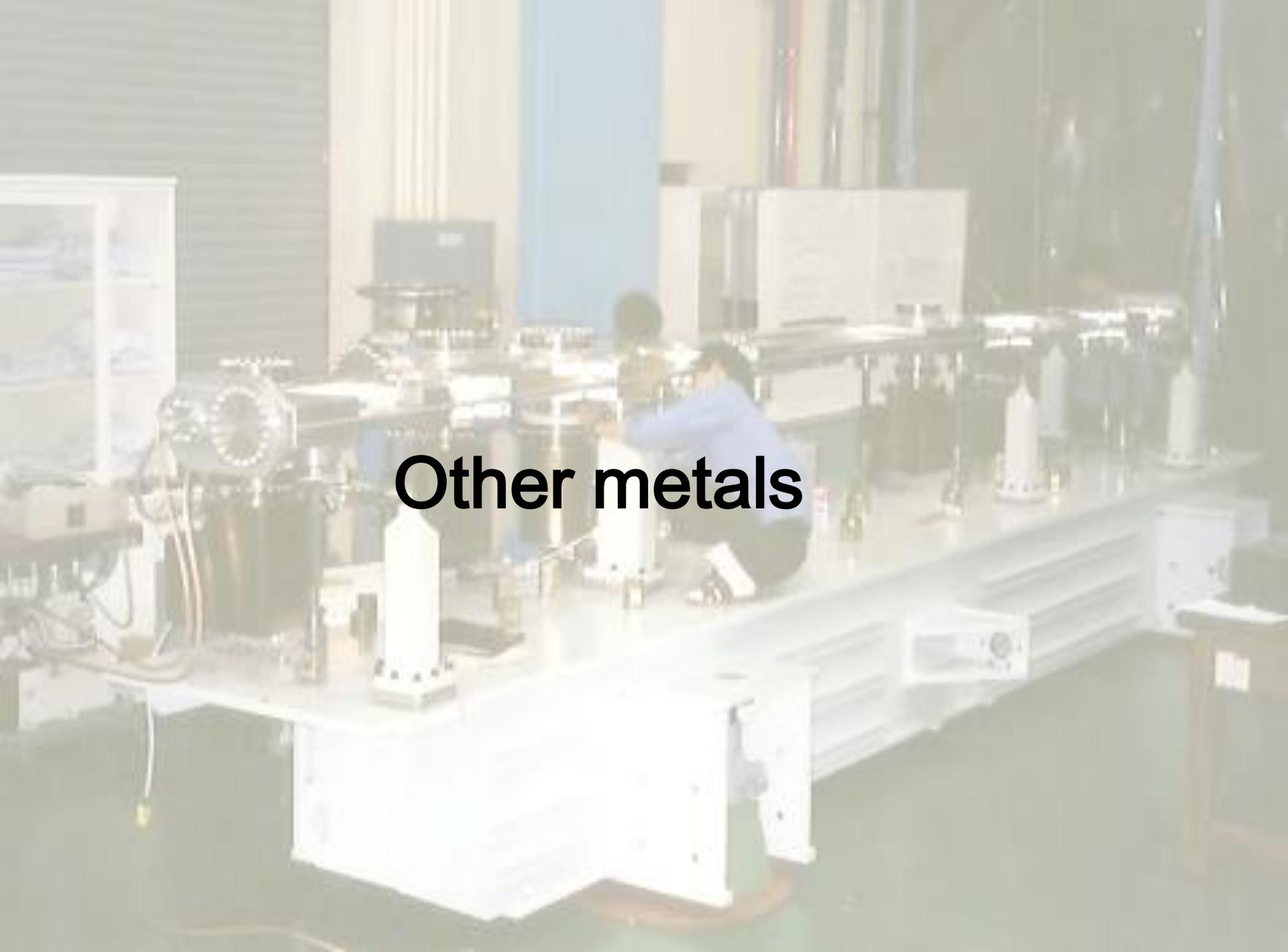
Copper alloy

- To increase its strength
 - OFC + Al_2O_3 (0.1-0.5%) GlidCop
 - Yield strength(at 0.2% offset) > 200 MPa
(OFC < 100 MPa)
 - OFC + Ag or $Au_{0.2\%}$ Expensive (x 4)
 - OFC + Zr High outgassing rate
 - OFC + Be Brazing(X), EBW(O)
 - OFC + Cr



Outgassing rates

q(mbar l/s cm ²)	@ 10 hour	Baked
Aluminum alloy	5×10^{-8}	$< 5 \times 10^{-13}$
Aluminum(anodized)	3×10^{-7}	5×10^{-10}
Stainless steel	1×10^{-8}	$(2-6) \times 10^{-12}$
Mild Steel	2×10^{-7}	$< 1 \times 10^{-12}$
Brass	6×10^{-7}	
Copper	5×10^{-9}	1×10^{-12}
Copper (OFHC)	2×10^{-9}	$< 1 \times 10^{-12}$

A photograph of a laboratory or workshop. A person in a blue shirt is kneeling at a long white workbench, working with various pieces of equipment. The bench is cluttered with glassware, including beakers and flasks, and other laboratory instruments. In the background, there are more pieces of equipment and a person standing. The overall scene is a busy laboratory environment.

Other metals

Gold and Silver

- Gold
 - ✓ Very low vapor pressure
 - ✓ Used as **metal gaskets**, surface seals in valves
as coating for electrical conductors
 - ✓ Used as brazing **filler** alloys
 - ✓ Cu/Au or Cu/Au/Pd
- Silver
 - ✓ Very low vapor pressure
 - ✓ Silver **plated** bolts/nuts to reduce friction/cold weld.
 - ✓ High oxygen permeation rate through silver at high temperature.

Titanium

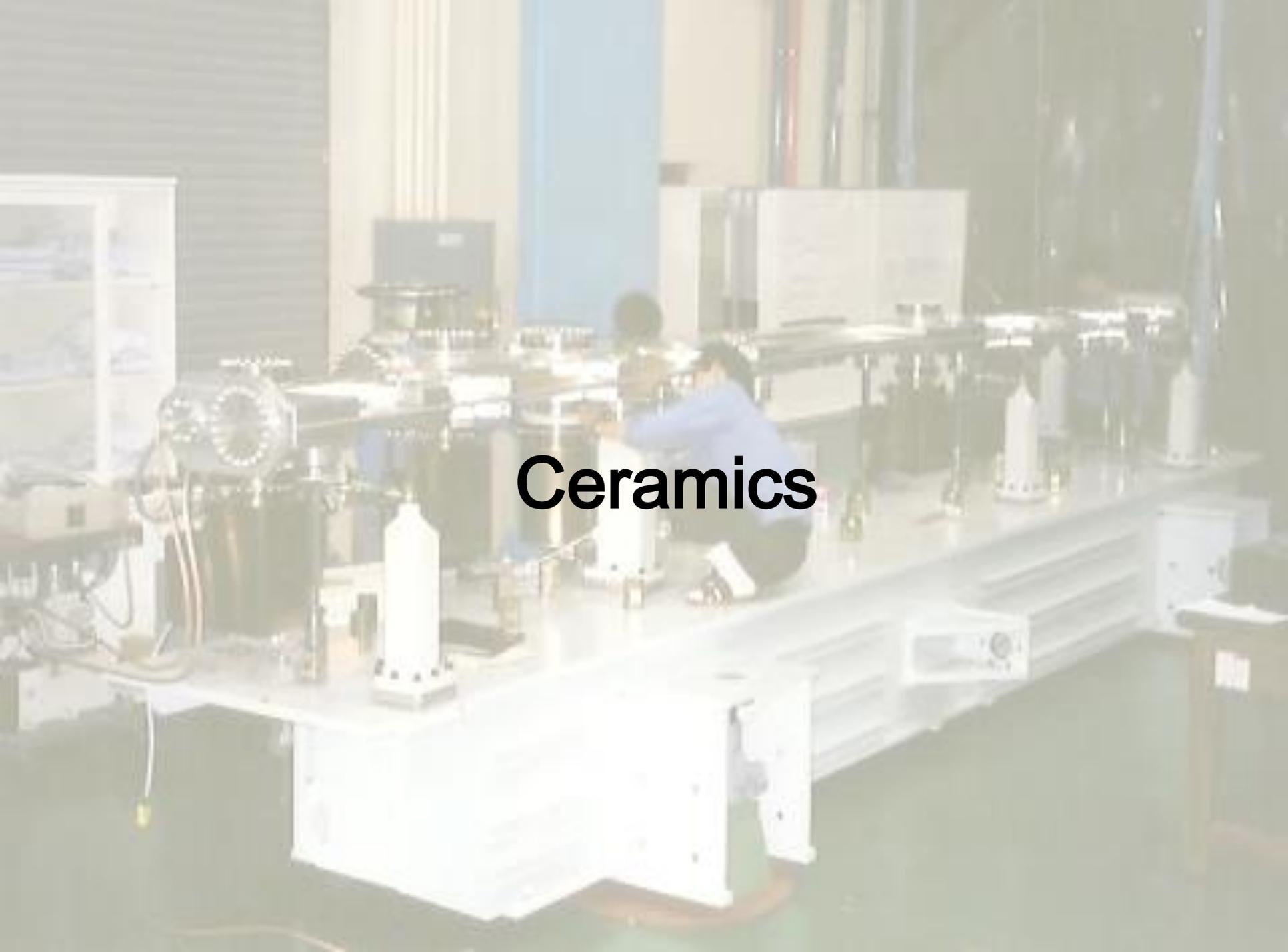
- Very active metal
- Easily react with O_2 , N_2 at $> 150^\circ C$
 - ✓ Weld should be done with inert gas environment
- Used as *metalizing materials* for brazing
- *TSP*(Titanium sublimation pump)
 - ✓ Sublimation at $\sim 1,350^\circ C$
 - ✓ $\sim 5 \mu m$ coating for 1 hr
- *Ion pump*: *Cathode* material
- *NEG alloy* (*Ti-Zr-Fe*)
- *Chamber materials*: Very low outgassing rate

Indium

- Melting point; 156°C
 - Not suitable for bakeable UHV applications
- Very low vapor pressure
- Very soft
- High thermal conductivity

- Thus indium may be used as
 - ✓ *vacuum seal* for UHV at cryogenic applications
 - ✓ *thermal conductors* between two different metals

Ceramics



Ceramics

- Ceramics
 - ✓ Non-metal, inorganic materials
 - ✓ Mainly used as **insulators** in vacuum technology
- Three types of ceramics
 - ✓ Pure-oxide ceramics
 - ✓ Silicate ceramics
 - ✓ Glass-ceramics

Ceramics

- Pure oxide ceramics
 - ✓ Alumina, Zirconia, Beryllium oxide,...
 - ✓ Alumina (Al_2O_3)
 - Mostly used ceramics
 - Max temperature; 1,800°C
 - > 92% in vacuum technology
 - Can be brazed
 - Mainly used as electrical *feedthroughs/insulator*
 - Bakable upto 350-550°C
 - Tensile strength 25 kpsi (96% density)
 - ✓ Sapphire (monocrystalline Al_2O_3)
 - UV and IR transparent
 - Used as *vacuum window*

Ceramics

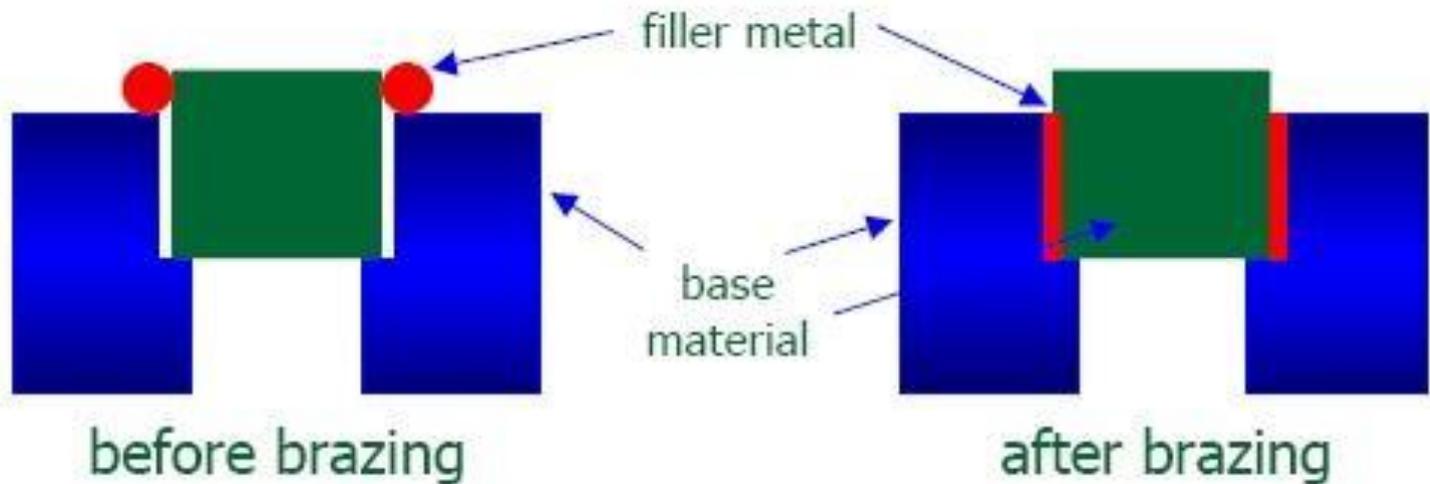
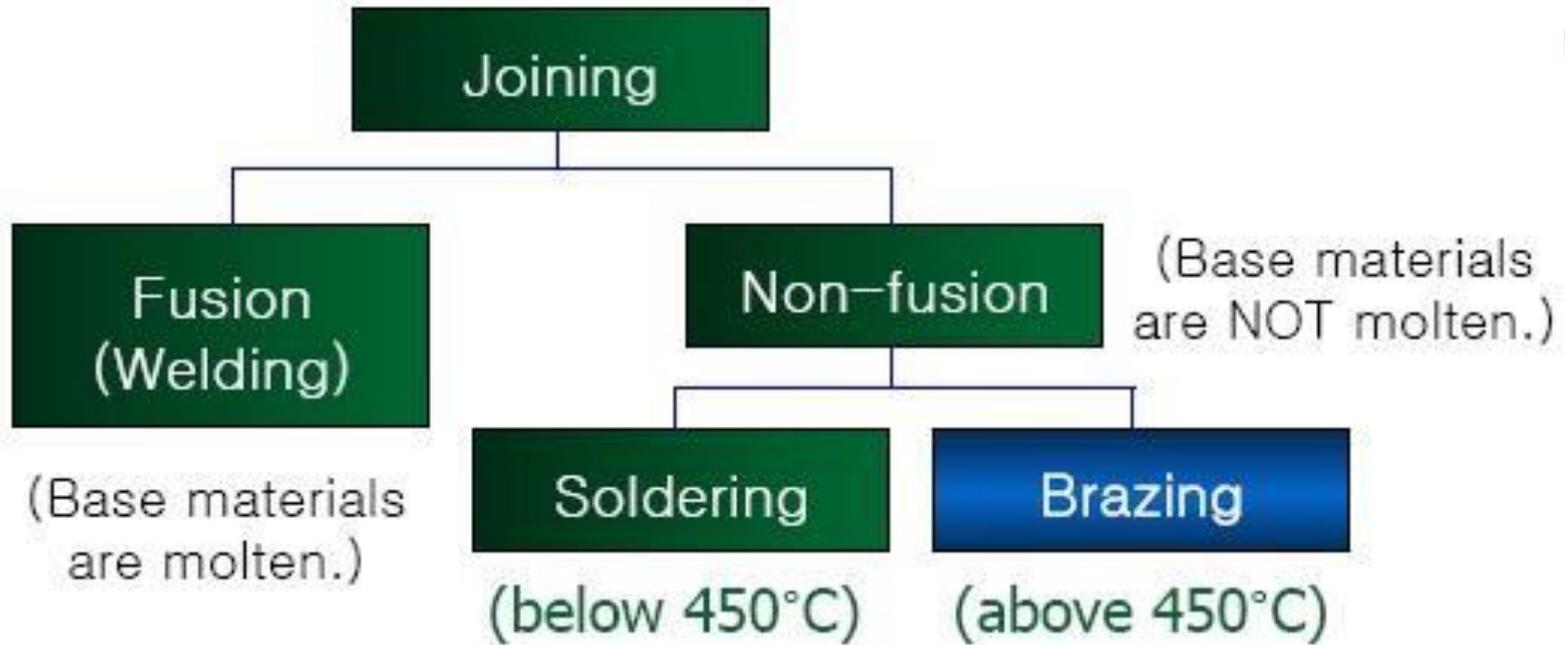
- Silicate ceramics
 - ✓ Steatite (MgO-SiO_2)
 - Max temperature $1,000^\circ\text{C}$
 - Tensile strength 15 kpsi
- Glass-ceramics
 - ✓ Crystalline ceramic
 - Can be **machined** with standard tools
 - Macor®, Corning 9658

Ceramics

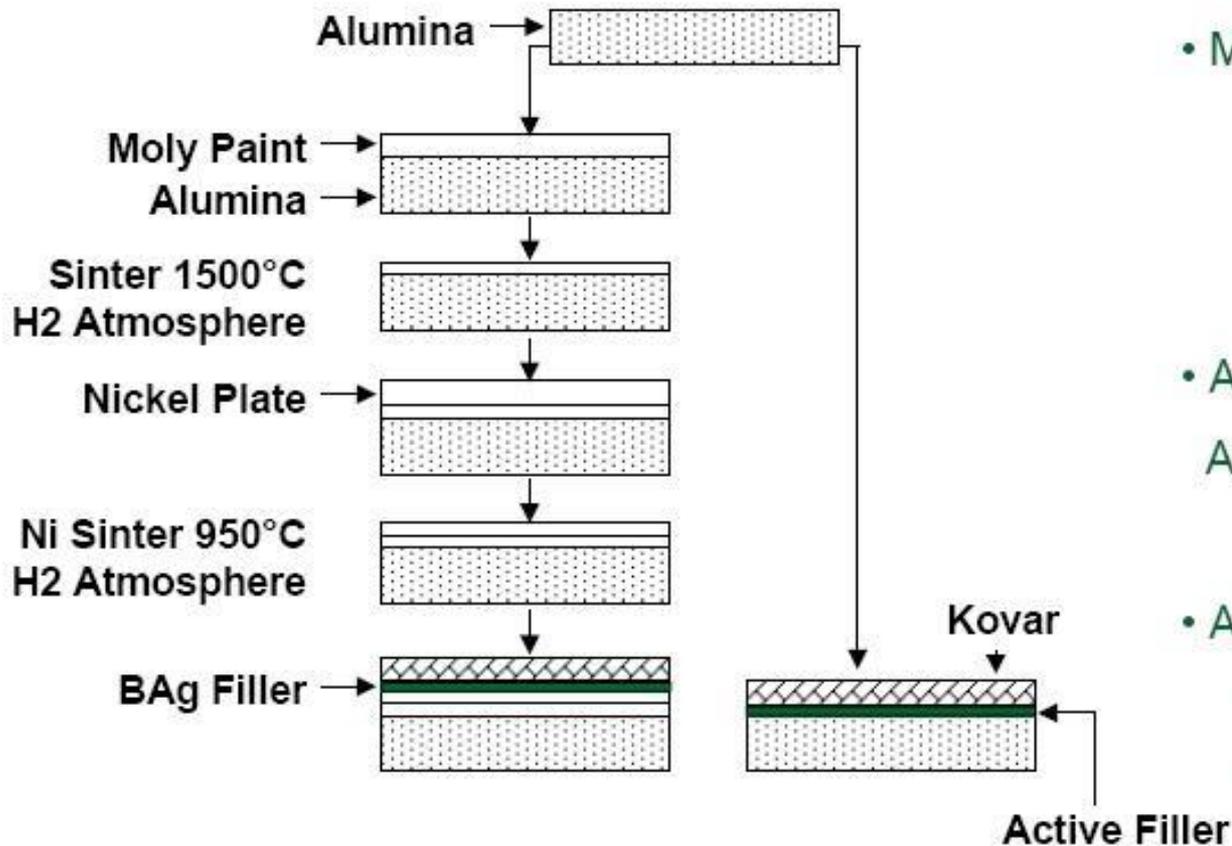
- Kovar

- ✓ Fe-Ni-Co alloy (thermal expansion ~ glass)
- ✓ Magnetic
- ✓ Intermediate material between ceramic and metal for brazing joint

Permanent vacuum joints



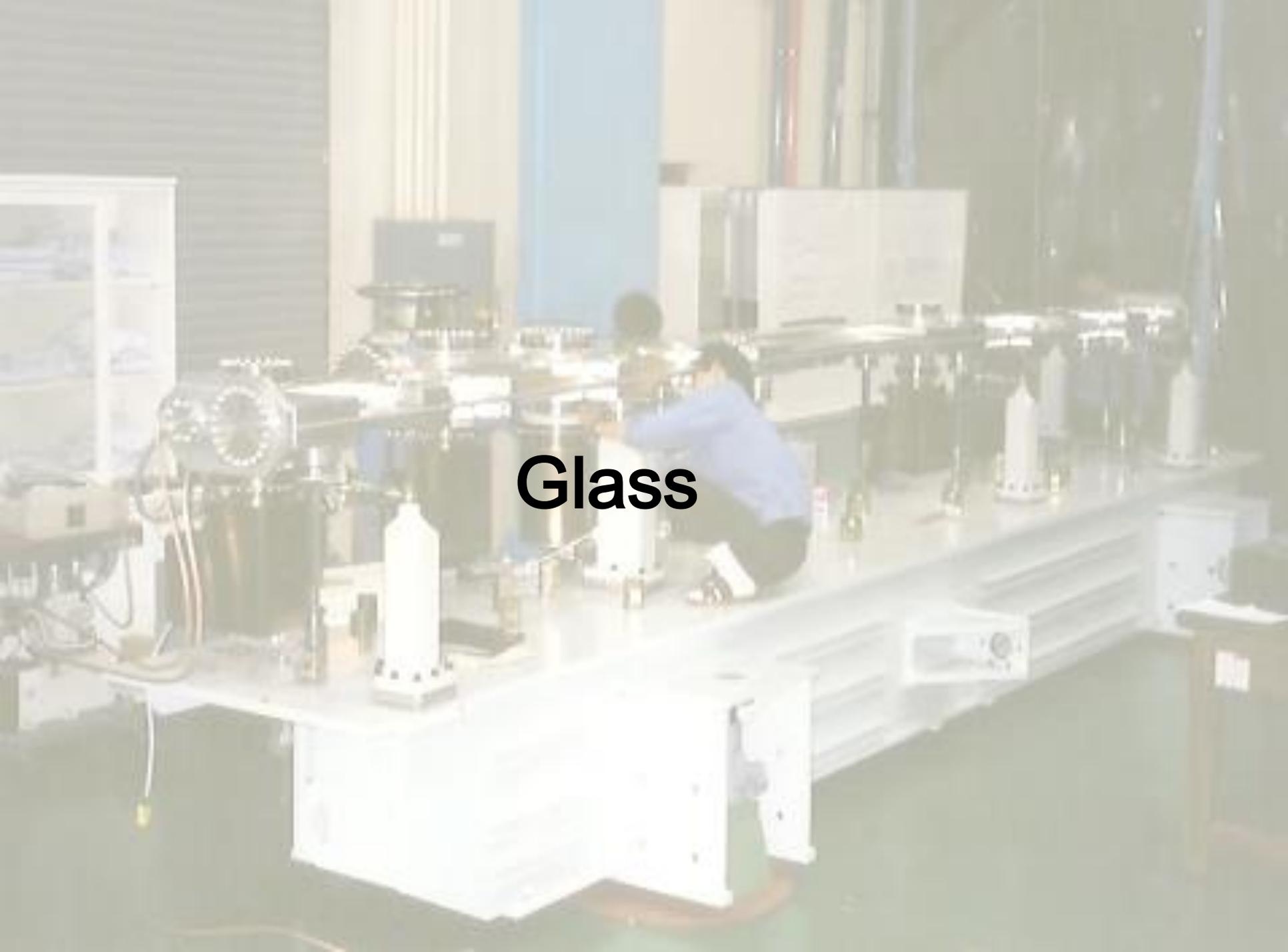
Brazing



- Metallization Method
 - Mo/Mn Paint Sintering
 - Vapor Deposition
- Active Brazing
$$\text{Al}_2\text{O}_3 + 3\text{Ti} \rightarrow 2\text{Al} + 3\text{TiO}$$
- Active Soldering
 - Brushing, Vibration required
 - No Spreading

Zeolite

- Very porous alumina silicates with alkali metals
- Mainly used as **drying** agents
- Mostly used as vapor adsorbing agents in vacuum
- *Sorption pump*
 - Liquid nitrogen cooled
 - Regeneration by heating

A photograph of a laboratory or industrial setting. A person in a blue shirt is crouching at a long white bench, working with various pieces of glassware and equipment. The bench is cluttered with items including a large glass vessel, several smaller glass bottles, and a piece of white equipment. The background shows a large window and some structural elements of the building. The word "Glass" is overlaid in the center of the image.

Glass

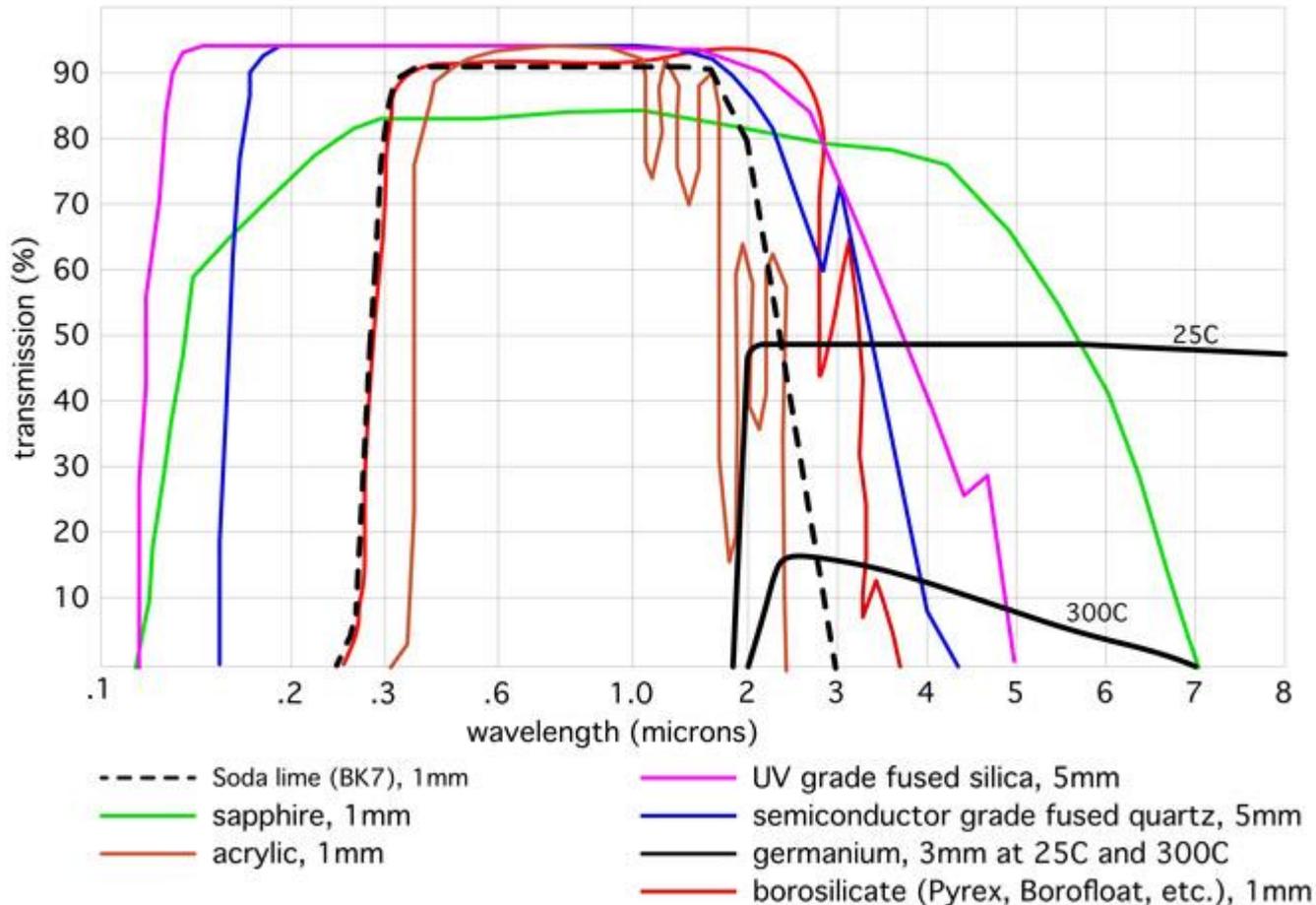
Glass

- Glasses
 - ✓ Non-metal, inorganic materials
 - ✓ Mainly used as *vacuum windows*
 - ✓ Also used in helium permeation leaks
- Three types of glasses
 - ✓ Soft glass: $60 \cdot 10^{-7} \text{ K}^{-1} - 120 \cdot 10^{-7} \text{ K}^{-1}$
 - ✓ **Hard glass:** $<50 \cdot 10^{-7} \text{ K}^{-1}$
 - ✓ Quartz glass: $\sim 5 \cdot 10^{-7} \text{ K}^{-1}$

Glass

- Soft glass(SiO_2 + Alkali)
 - 65–70% SiO_2 , 2.5–15% Na_2O , 5–15% CaO
- **Hard** glass ($\text{SiO}_2 > 70\%$ + Boron)
 - Corning 7056, Duran, Pyrex
 - Most common glass in vacuum tech.
- Quartz (SiO_2 100%)
 - Used as optical vacuum window

Optical transmission



Thickness affects: Making the material thicker only slightly affects transmission %, but can significantly affect how deep the transmission goes into the UV and IR

Plastics



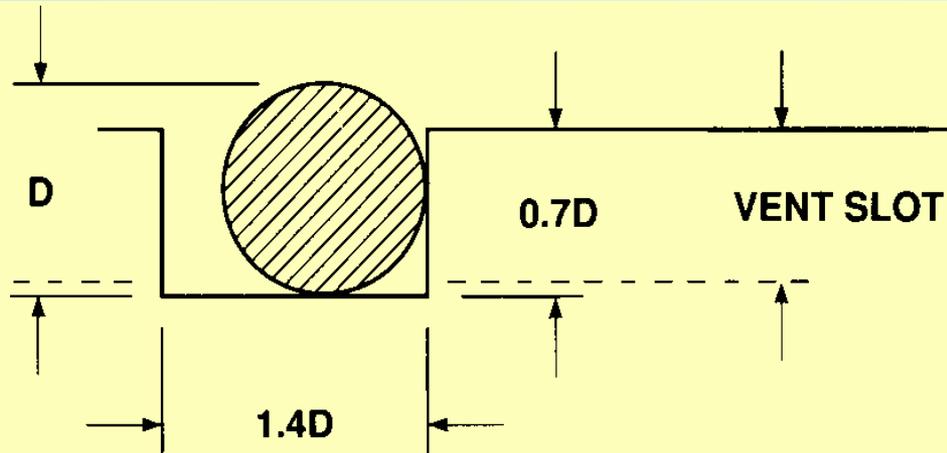
Plastics

- Elastomers
 - Elastic , Rubber-like materials
 - Used as reusable vacuum **gaskets** for RV, MV, HV, or (UHV)
 - Buna-N, Viton, Kalez, ...
- Thermoplastics
 - Thermally reversible
- Duroplastics
 - Thermally irreversible
 - **Epoxy**
 - Good adhesion with metals, glasses, ceramics

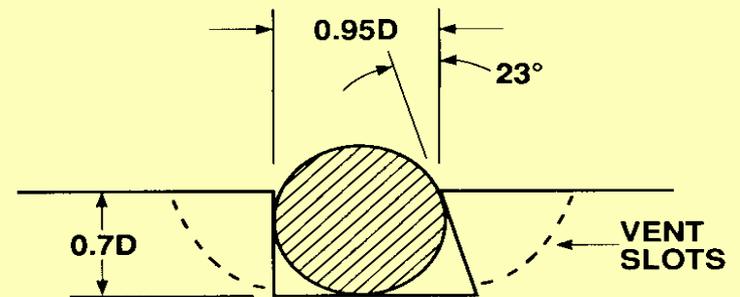
Outgassing rates

Outgassing rate (TorrLs⁻¹cm⁻²)	@ 10 h	Baked
Viton (FKM)	5×10^{-8}	5×10^{-10}
Buna N (NBR)	2×10^{-6}	4×10^{-8}
Epoxy (Shell Epon)	1×10^{-6}	8×10^{-8}
Teflon (poly'fluoro'lene/PTFE)	8×10^{-8}	8×10^{-9}
Nylon (polyamide)	3×10^{-7}	6×10^{-9}
PVC	3×10^{-7}	8×10^{-8}

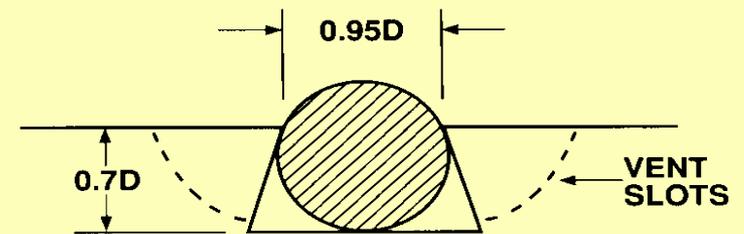
O-ring seal



Retaining the O-ring



A. DOVETAILED O-RING GROOVE



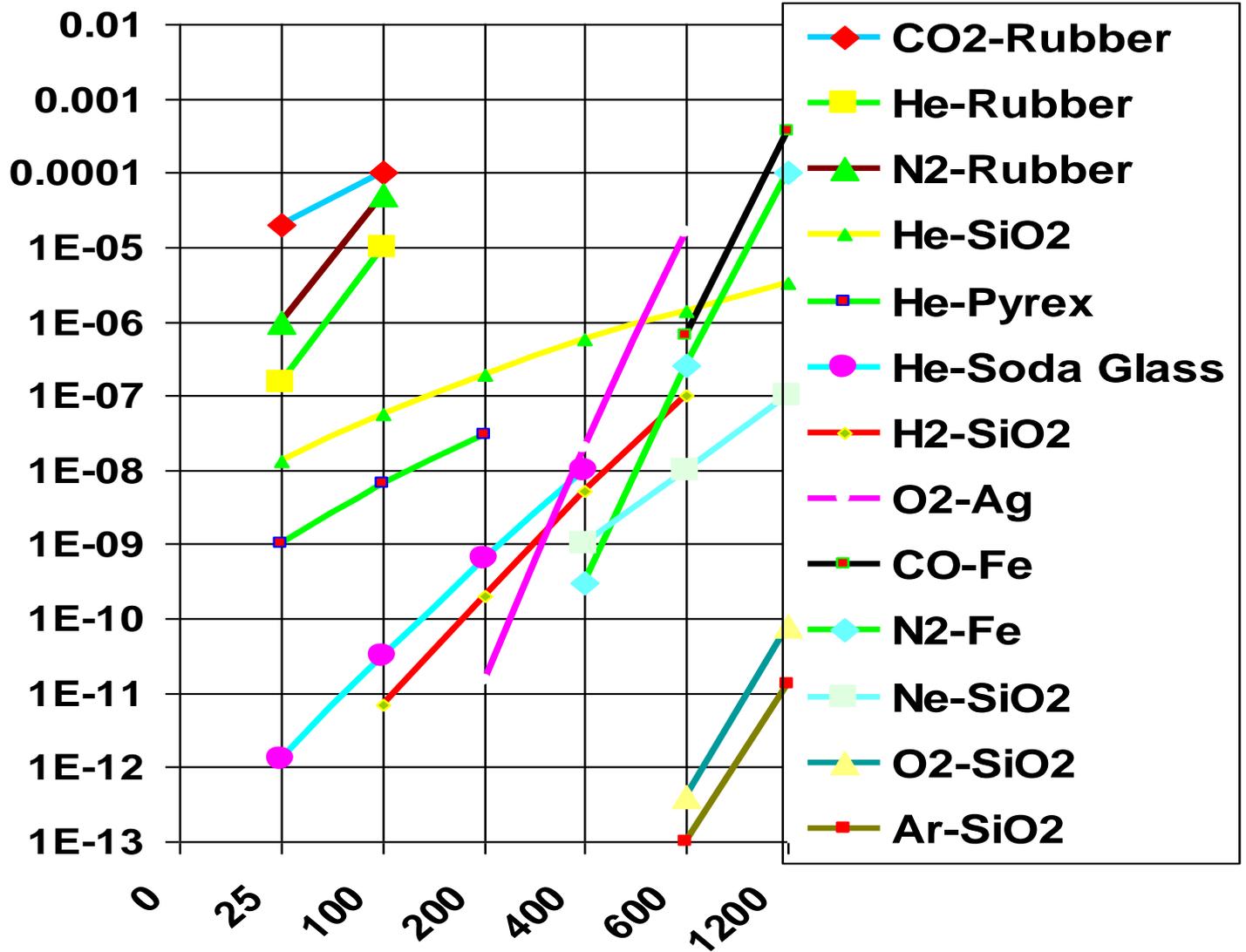
B. DOUBLE DOVETAILED O-RING GROOVE

DOVETAILING OF O-RING GROOVES

Permeation

- Permeation through solid materials involves
 - sorption, diffusion and desorption.
- Materials have permeation rates for different gases specific to that material.
 - steels have higher permeation rates with higher carbon content;
 - copper has low permeation for all gases;
 - aluminum has low permeation for hydrogen;
 - silver has high permeation rate for oxygen at high temperature
 - palladium has high permeation rate for hydrogen
 - **polymers are permeable to all gases.**

Permeation



Permeation

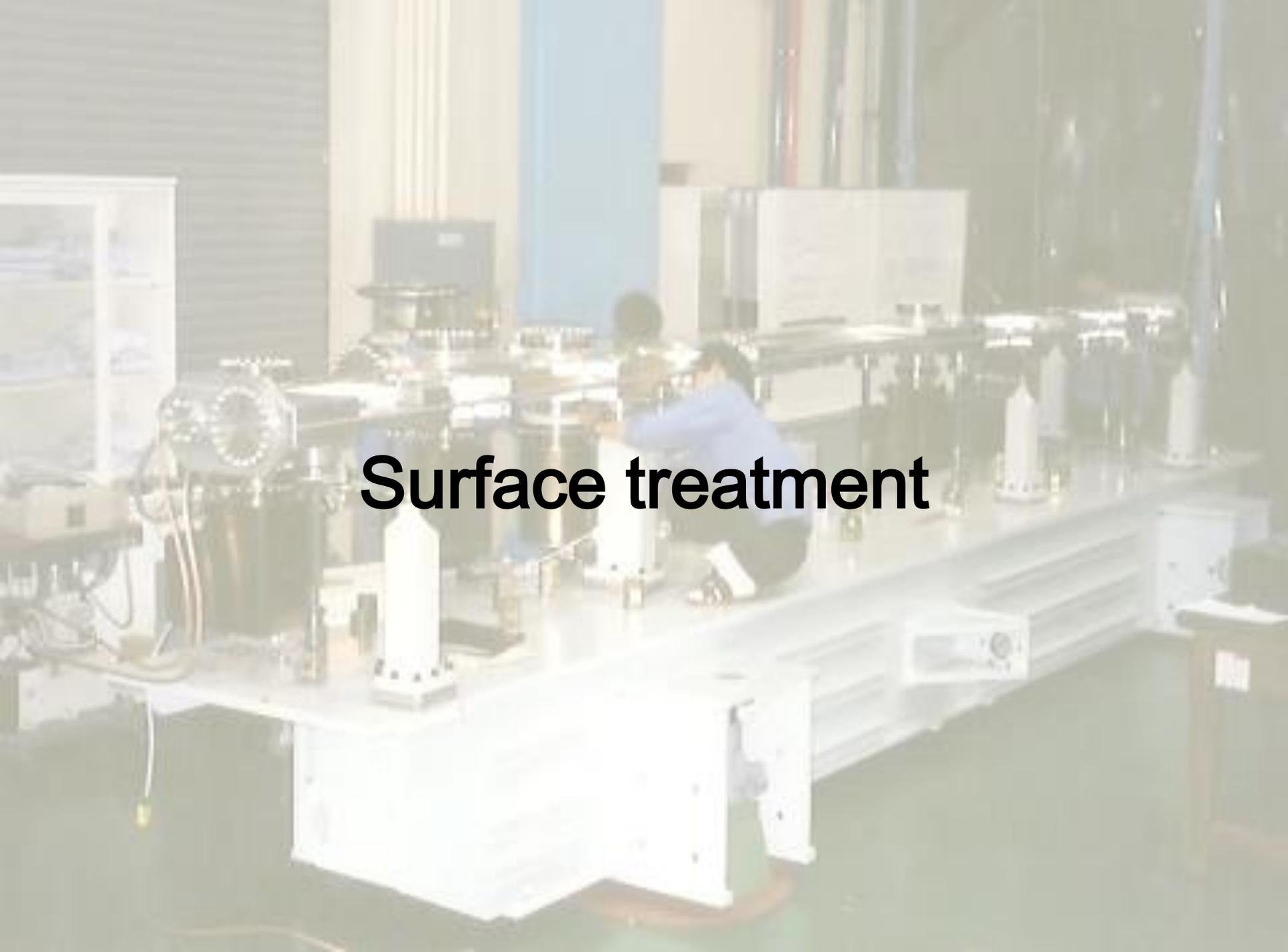
- Permeation is a strong function of temperature.
- Permeation gives additional gas loads.
 - Modifies the chamber environment(residual gases)
 - Affecting chemistry in vacuum process
 - Limits the ability to reach ultra high vacuum.
- Calibrated leaks
 - helium permeation through pyrex or quartz.

Helium permeation through elastomers

Polymer	Permeation rates	
	std-cc/s cm ² ($\times 10^{-7}$) at 1 atm	
	25°C	150°C
Viton	1.3	49
Buna N	0.8	25.2

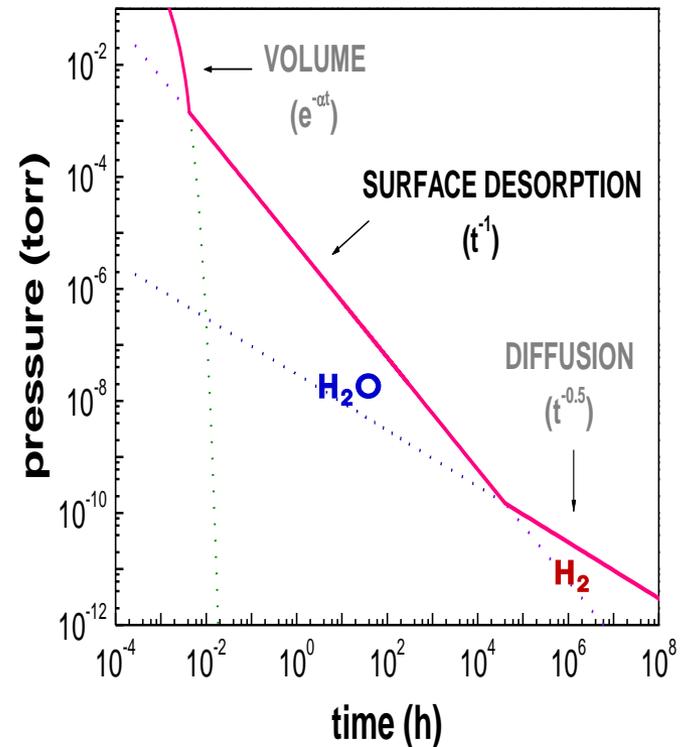
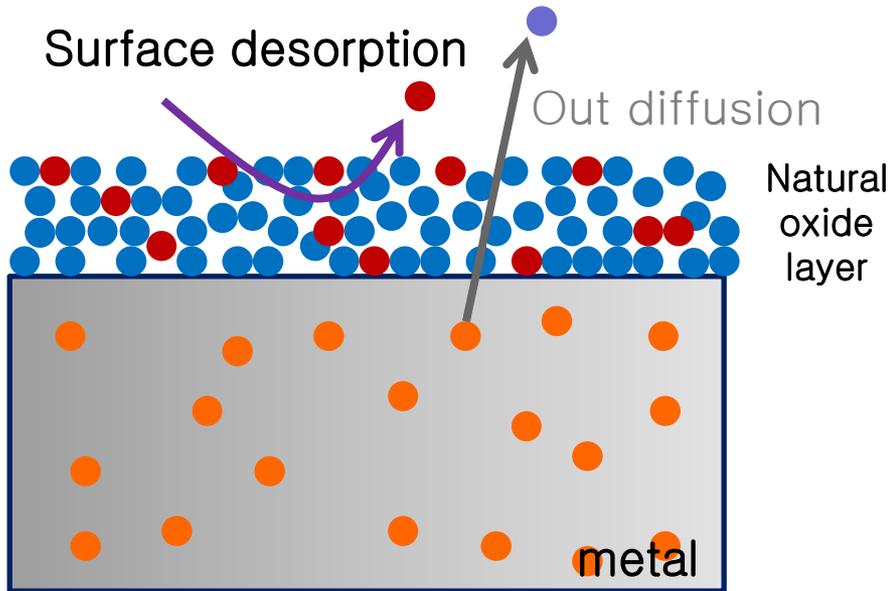
After several minutes helium will begin to permeate through the elastomer O-ring and show up as a leak.

Surface treatment

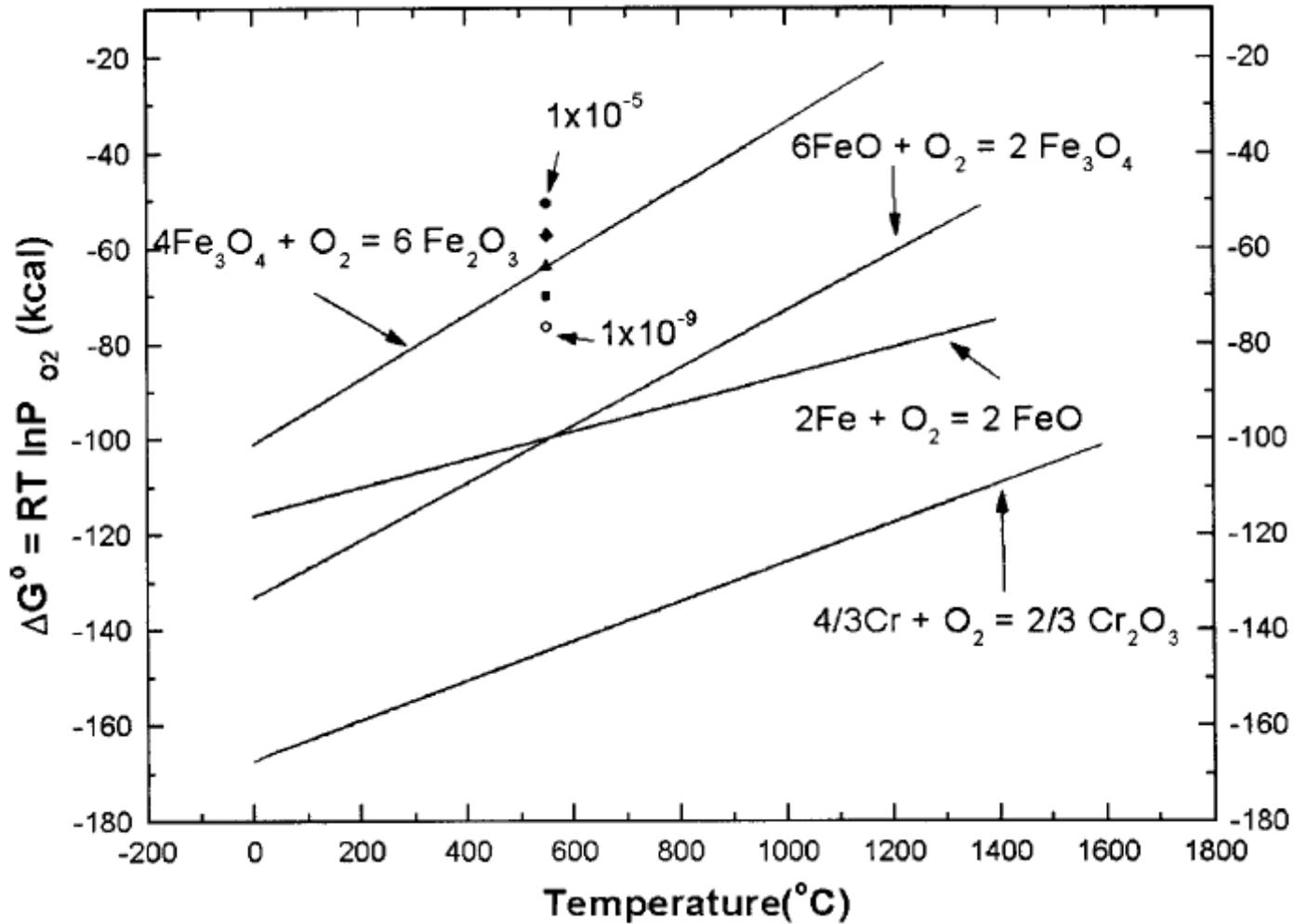
A photograph of a laboratory or industrial setting. In the center, a person wearing a blue long-sleeved shirt and dark pants is kneeling on a green floor, working on a piece of equipment. The equipment is a large, white, L-shaped structure with various components, including a large circular fan or motor on the left side. The background shows a blue wall and some other laboratory equipment. The text "Surface treatment" is overlaid in the center of the image.

일반 스테인리스강의 진공특성

- ✓ 일반 금속(스테인리스강)의 표면은 다공질의 산화막이 형성되어 있음.
- ✓ 공기 노출 시 **다공질 표면**에 물과 같이 흡착성이 강한 입자가 다량으로 흡착됨.
- ✓ 진공 배기하면 흡착된 **물 분자가 서서히 방출**되므로 압력은 시간의 역수에 비례하며 ($p \propto 1/t$) 느리게 떨어짐.

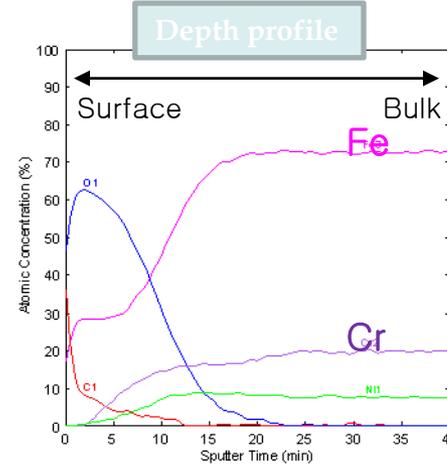
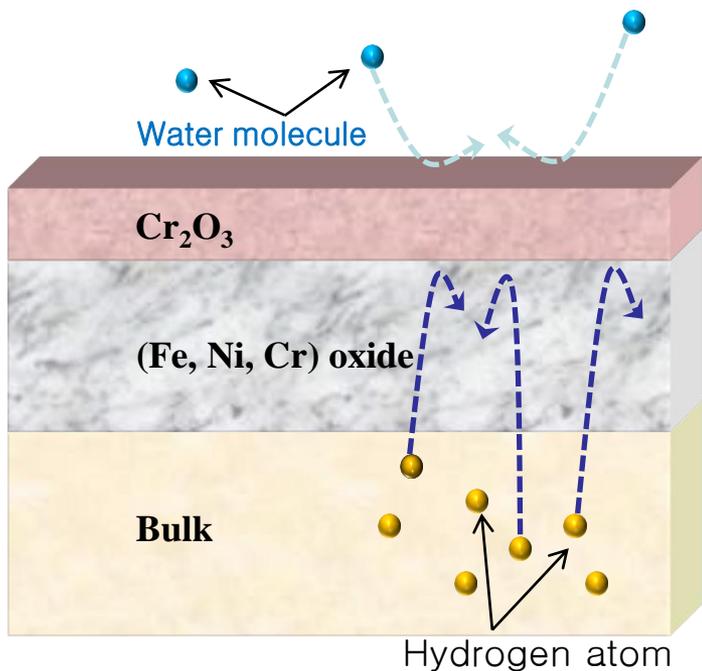


Ellingham Diagrams

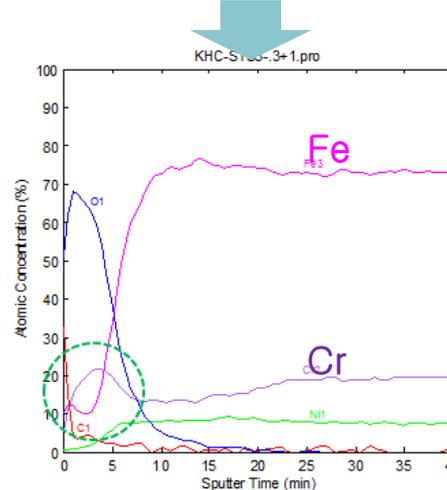


Vacuum Thermal Oxidation (VTO)

- ✓ VTO 표면 처리를 하면 다공질 산화막 위에 **순수한 크롬산화막이 형성됨.**
- ✓ 크롬산화막에 의해 공기 노출 시 물 분자의 흡착량이 크게 감소하고 흡착된 물 분자도 비교적 쉽게 떨어져나가므로 **급속 배기 가능함.**

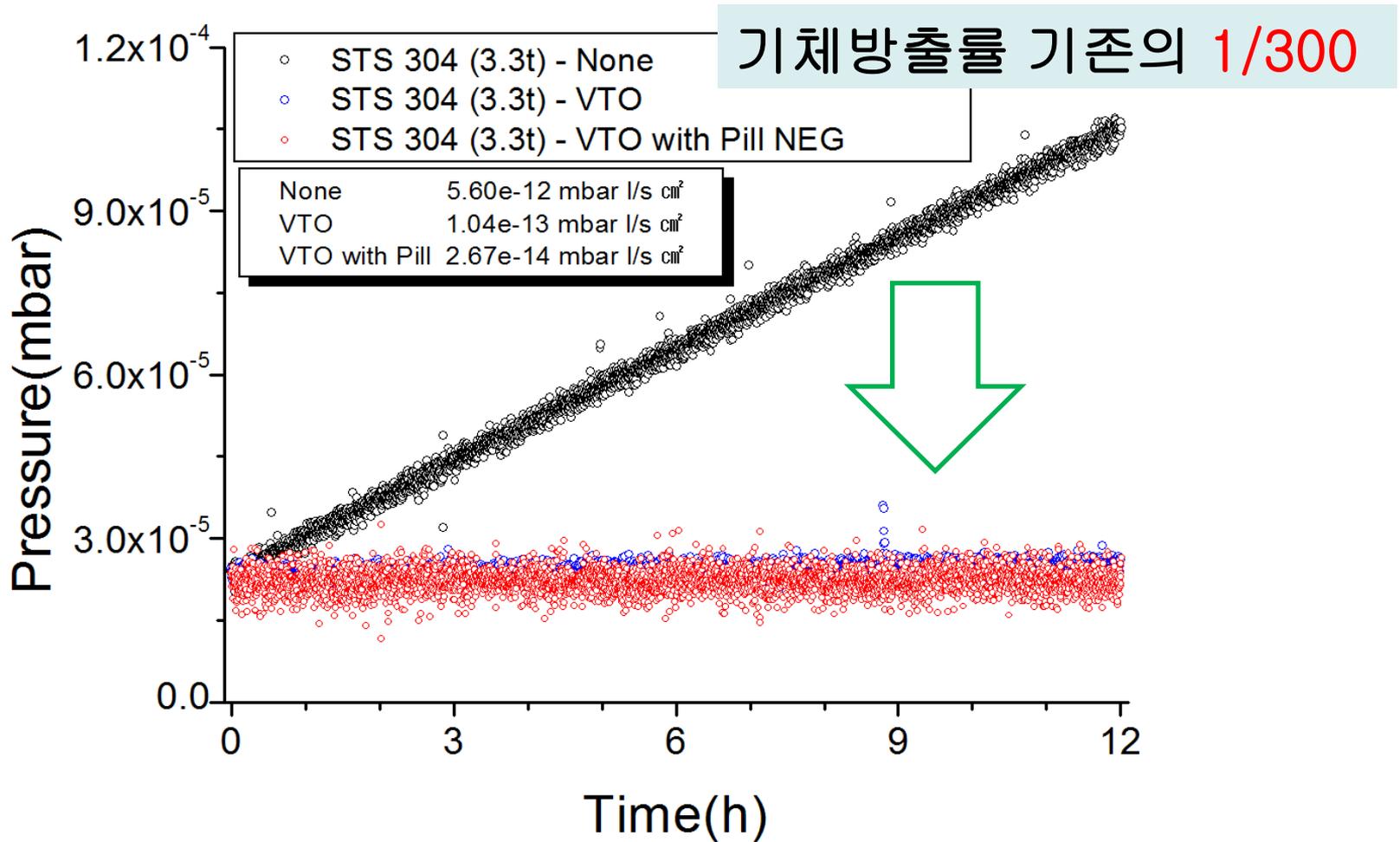


일반 (STS304)
:표면에 철산화물 많음



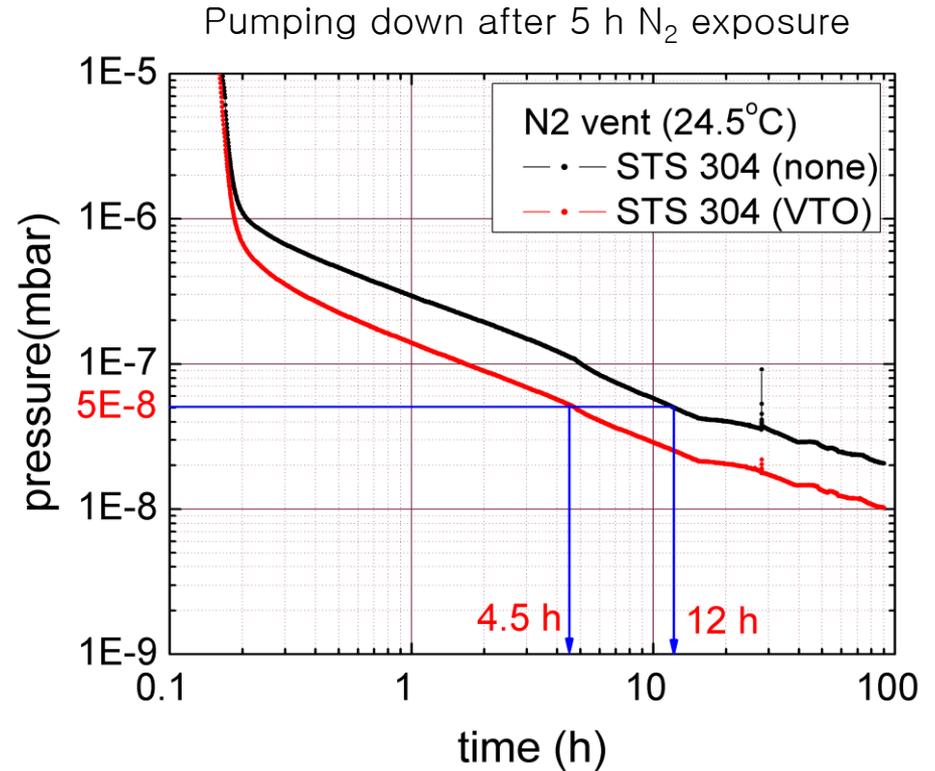
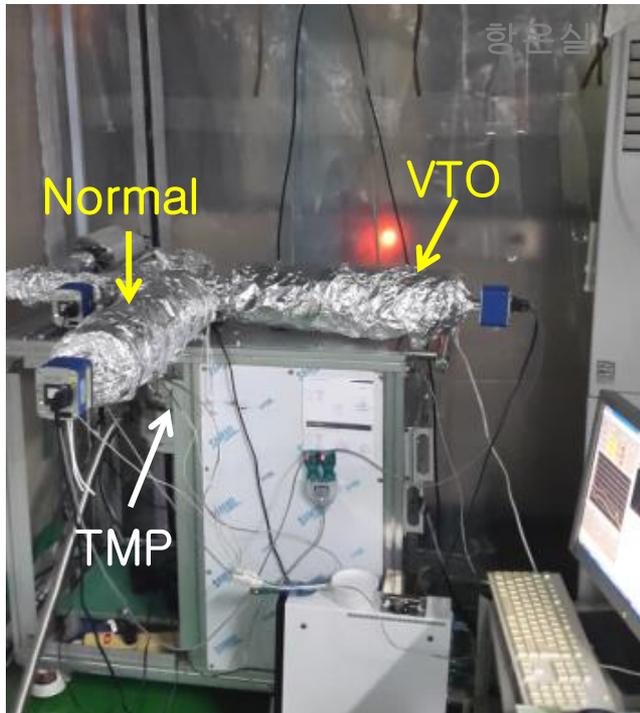
VTO (STS304)
:표면이 순수한 크롬산화물

기체방출률 비교



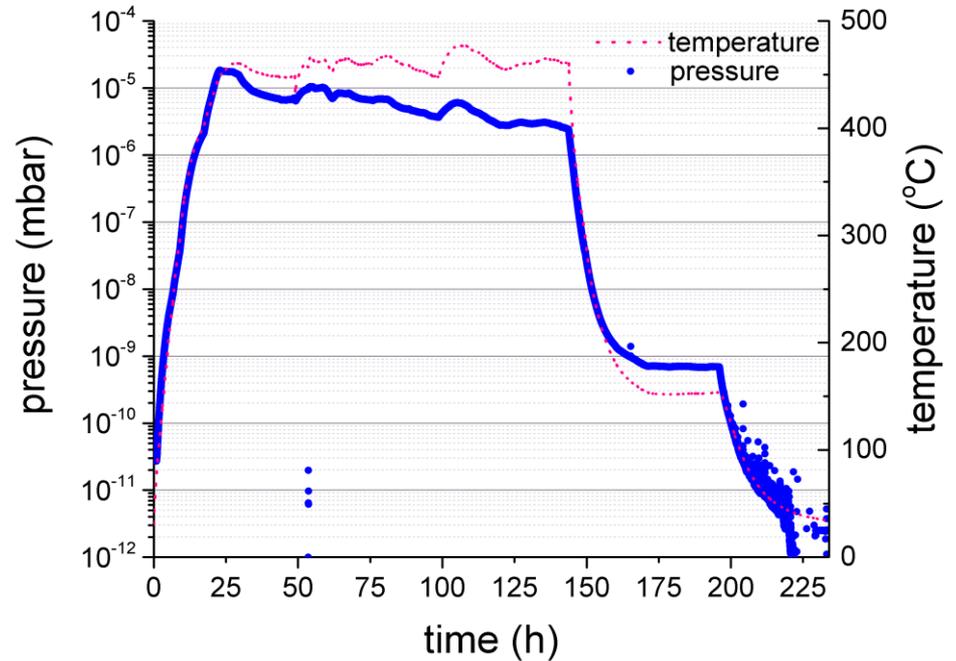
배기 시간 비교

- ✓ 동일한 STS304 진공용기(6 리터) 2 set 제작하여 1 set는 VTO 표면처리 함.
- ✓ 동시에 같은 조건으로 baking 후 질소 벤팅하여 5시간 유지한 이후 진공배기 함.
- ✓ 일반 STS304 진공용기와 VTO 처리한 진공용기의 배기시간 비교
 : 고진공 (5E-8 mbar) 도달 시간이 일반 STS의 1/3



극고진공 챔버

- 1000 리터 급 극고진공 시스템: 상용 진공게이지의 측정 하한 도달.



- Literature survey
 - Do not believe those data
- Test and evaluation
- Make your own data