

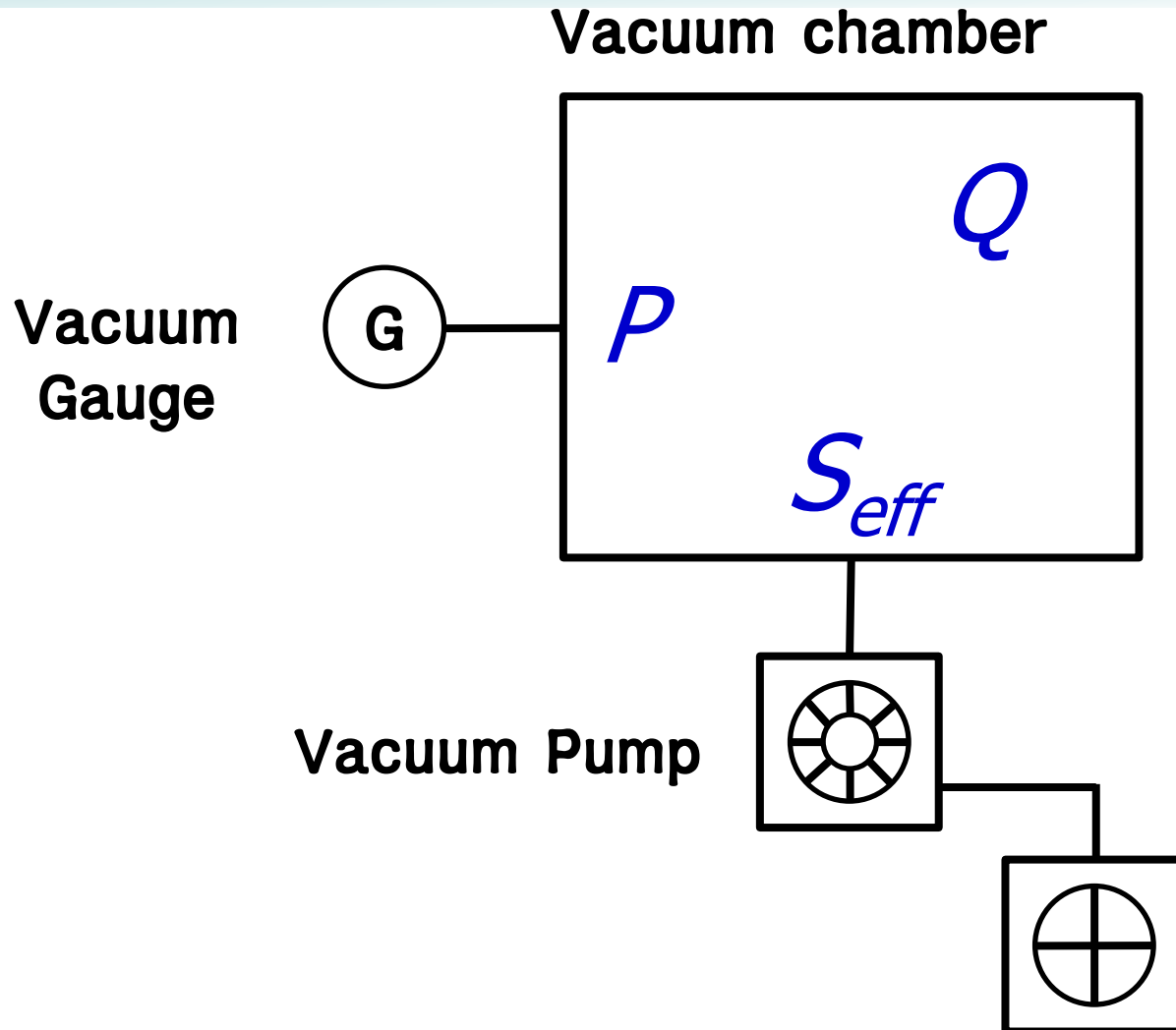
진공재료의 종류와 부품의 특성 및 제조 관리

2019.06.21

하태균
포항가속기연구소

1. 진공 재료의 특성 이해

2. 진공 재료의 성능 향상



왜 재료가 중요한가?

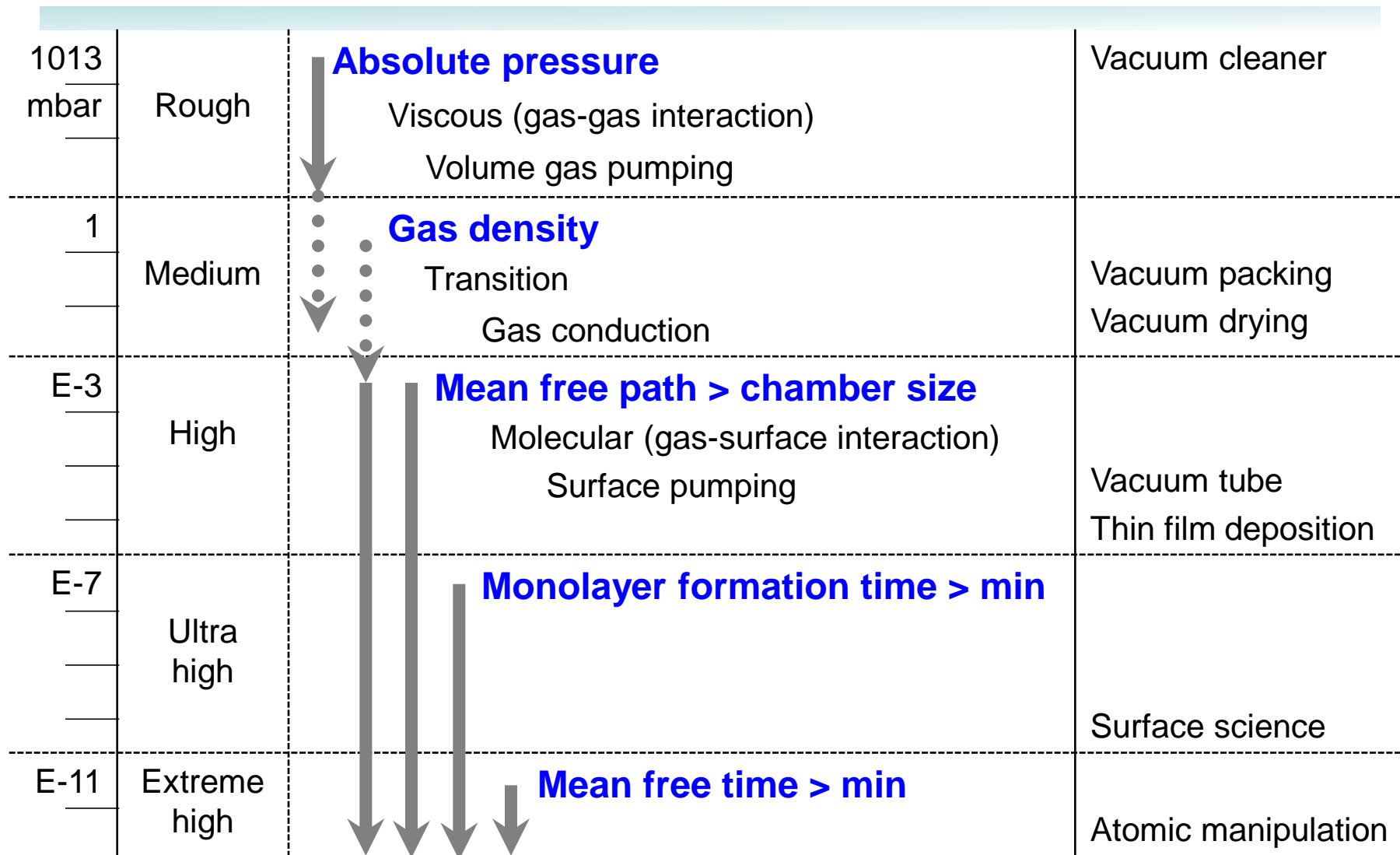
System base pressure


$$P \text{ (mbar)} = \frac{Q}{S_{eff}} \frac{(\text{mbar } \ell / \text{s})}{(\ell / \text{s})}$$



$$\frac{1}{S_{eff}} = \frac{1}{S} + \frac{1}{C}$$

목표진공도에 따라 재료가 달라짐



A person in a blue shirt is working on a complex vacuum technology setup in a laboratory. The setup includes various components like pumps, gauges, and tubing, all mounted on a large metal frame. The person is crouched down, adjusting a component. The background shows a laboratory environment with shelves and other equipment.

Materials for Vacuum technology

진공재료 선택

재료 선택시 고려할 내용

- 얻고자 하는 진공도는?

관련 항목

- 적절한 기체방출률
 - ✓ 전처리 필요여부
 - ✓ 세척 방법
- 고유 증기압
- 적절한 녹는점과 끓는점
- 재료의 누출률
- 적절한 투과율
- 불순물 기체 함유량
- 깨끗한 표면

진공재료 선택

재료 선택시 고려할 내용

- 강도는 충분한가?
- 제작은 용이한가?
- 기밀방법은?
- 사용환경과 수명은?

관련 항목

- 1 기압차 유지
 - ✓ 허용 응력과 허용 변형
- 기계가공성, 성형성
- 적절한 접합법
- 금속 또는 엘라스토머
- 내부식성(내화학적성)
- 내방사성
- 열변형(적절한 열팽창 거동)

진공재료 선택

재료 선택시 고려할 내용

- 사용온도와 수명은?
- 투자율은?
- 전기전도도는?
- 제작 비용은?
- 재료 수급성은?

관련 항목

- 금속 재료 또는 플라스틱?
- 극저온 또는 고온?
- 높은 내열피로성
- 전자기 특성에 영향
- 전도체 또는 부도체
- 국내 또는 국외 수입
- 구입 기간

Vacuum Materials

Steels

Stainless Steel

Aluminum (alloy)

Copper (alloy)

Other metals

Ceramics

Glass

Plastics



Steels

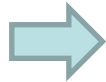
Steels

- Mild/structural steels
 - ✓ Carbon < 0.3 %
 - ✓ outgassing rate
 - ~~$q > (20 \sim 200) \times \text{STS}$~~
 - $q_{rd} > (20 \sim 200) \times \text{STS}$, $q_{H_2} \lesssim \text{STS}$
 - ~~HV compatible (10^{-6} mbar), endless emission of CO~~
 - UHV compatible, $\text{RegGas}_{\text{ST}} \sim \text{RegGas}_{\text{STS}}$
 - ✓ Weldable
 - ✓ Easy to corrode
 - ✓ Needs anti-corroding coating
 - ✓ Magnetic
 - ✓ Shielding material for magnetic field

Steels

- Mild/structural steels
 - ✓ S235, S355, S20C
 - ✓ UHV compatible
 - plate, pipe, rod
 - $\sim 10^{-11}$ mbar
 - $\lesssim 5 \times 10^{-12}$ (mbar l/s cm²) after bake
 - ✓ MV, RV compatible
 - Cast parts; pump and valve housing
 - $\sim 10^{-3}$ mbar

Steels



- Anti-corrosion
 - ✓ $\sim 300^{\circ}\text{C}$
 - ✓ Process

~~• Low q~~
~~— UHV~~
~~— XHV~~

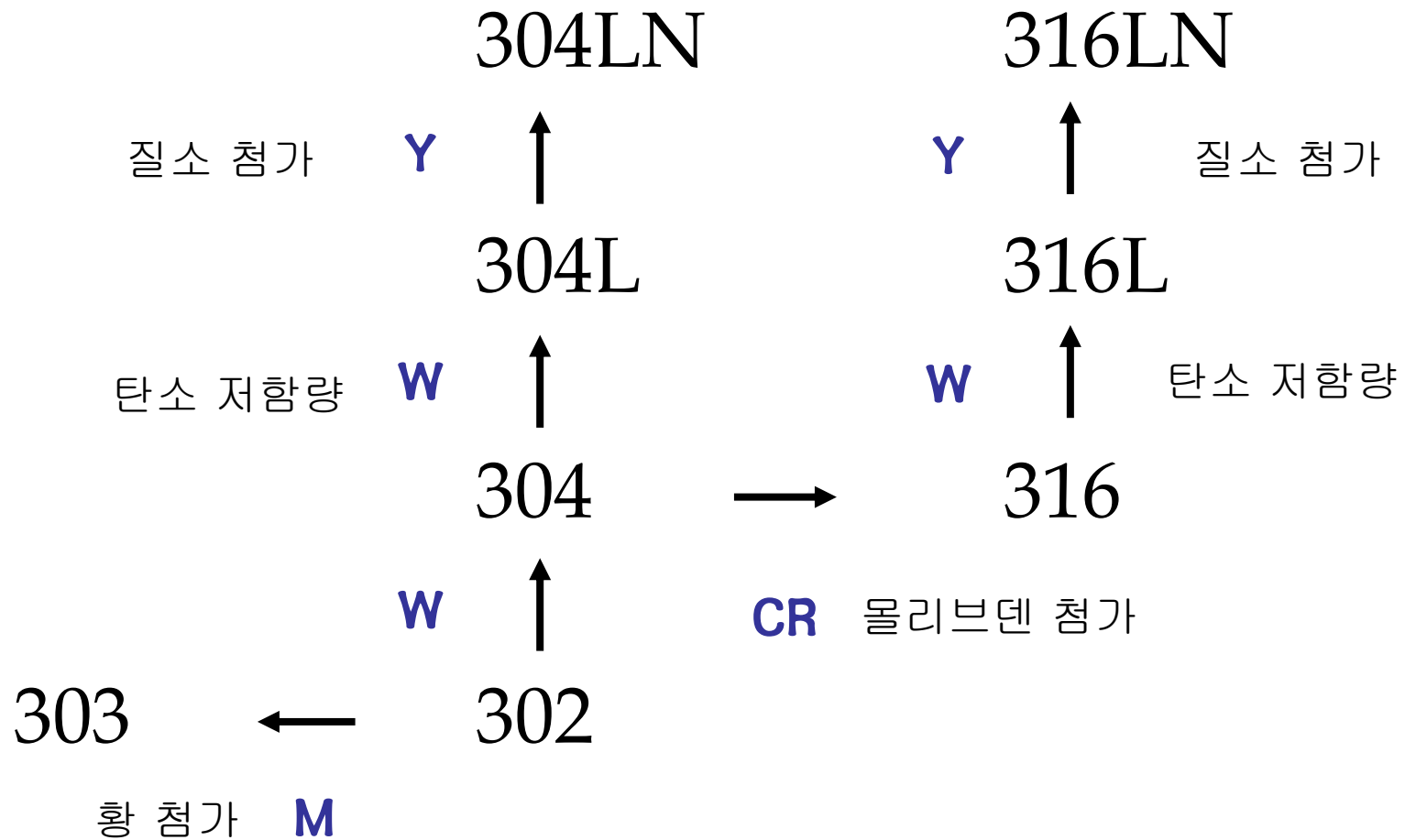


Stainless
steels

Stainless steel

- AISI 304/316
 - ✓ Austenitic
 - High strength
 - Non-magnetic (but, not entirely)
 - Good weldability
 - ✓ Corrosion resistance
 - During vacuum processing and bakeout ($\sim 300^{\circ}\text{C}$)
 - ✓ UHV/XHV compatible
 - $< 10^{-11}$ mbar
 - $(2\sim6) \times 10^{-12}$ (mbar l/scm²) after bake
 - $< 1 \times 10^{-13}$ (mbar l/scm²) after special treatment

Stainless steel



18-8 Steel Family

Stainless steel

- Role of ingredients
 - Cr(10%) Resistance to oxidation
 - Ni(8%) Austenitic structure/ Anticorrosion
 - Mo Accelerates passivating film formation
 - W Mechanical resistance at high temperature
 - Ti During welding and cycles stabilizes the austenitic structure
 - N Mechanical characteristics

Stainless steel

- 303

- ✓ 19% Cr, 10% Ni, 0.15% C, 0.15% S
- ✓ A free machining stainless
- ✓ Not suitable for UHV applications
 - ✓ Emission of sulfur at higher temperature
 - ✓ (The q of 303 can be lowered to 10^{-13} mbar l/s·cm² by a combination of fabrication and post treatments such as bakeout.)
- ✓ Welding is a problem due to the evolution of sulfur during welding causing porosity.

Stainless steel

■ 304

- ✓ 18% Cr, 8% Ni, 0.08% C
- ✓ Most common materials used in vacuum technology

■ 304L

- ✓ 18% Cr, 8% Ni, 0.03% C
- ✓ One of common steels used in vacuum technology
- ✓ Less carbide precipitation
- ✓ Cleaner machining and better welds than 304
- ✓ Lower mechanical properties than 304

Stainless steel

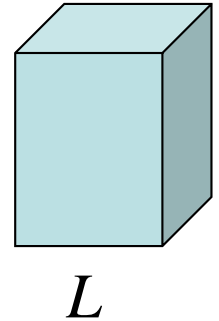
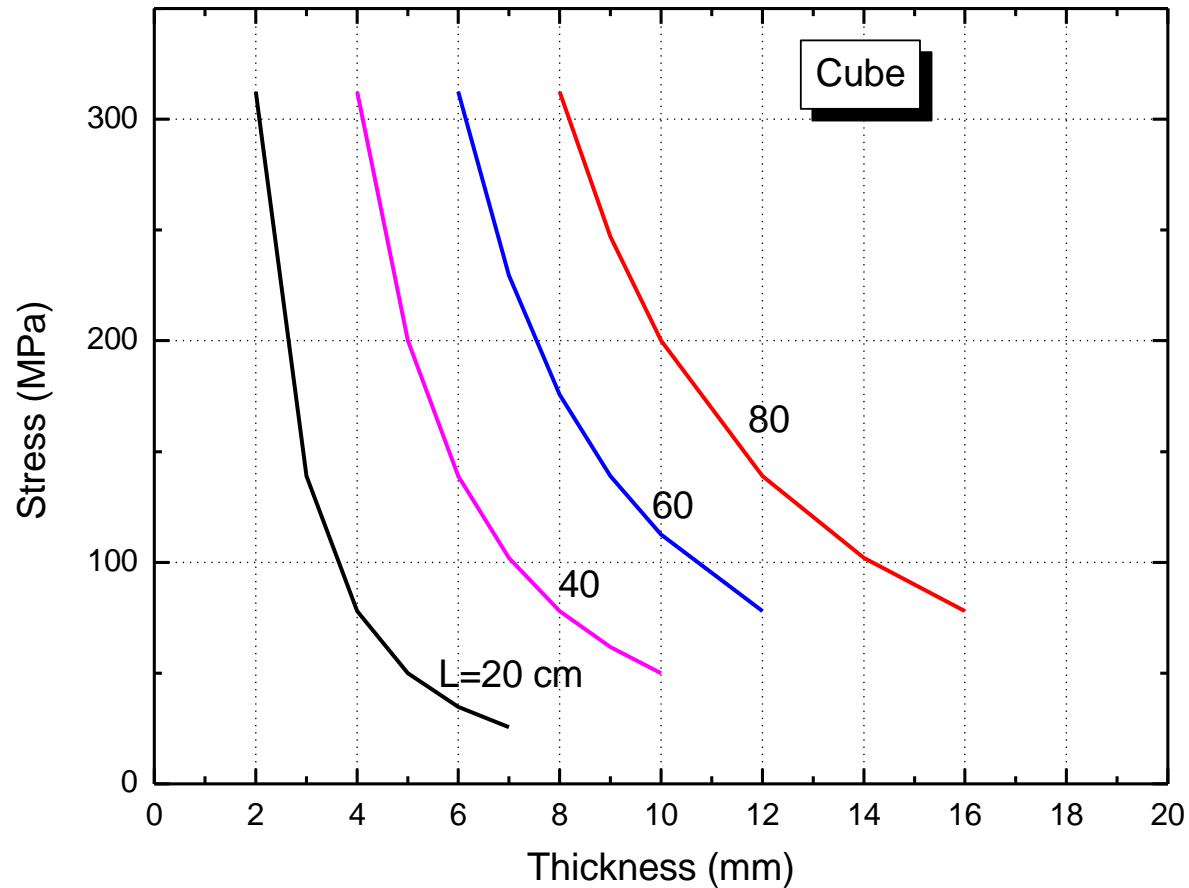
■ 316L

- ✓ 18% Cr, 14% Ni, 0.03% C, 3% Mo
- ✓ Stabilized with molybdenum to prevent carbide precipitation in the weld zone
- ✓ Lower q than 304(L)
 - $\sim 2 \times 10^{-12}$ (mbar l/scm²) after bake
 - $< 5 \times 10^{-14}$ (mbar l/scm²) after special treatments
- ✓ Used where chemical compatibility is a concern.
- ✓ Low-magnetic stainless steels ($\mu_r < 1.02$)
 - Suitable for analyzers, accelerators
 - Heat treatment; (750 ~ 1050°C)

Mechanical properties

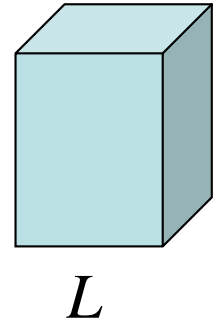
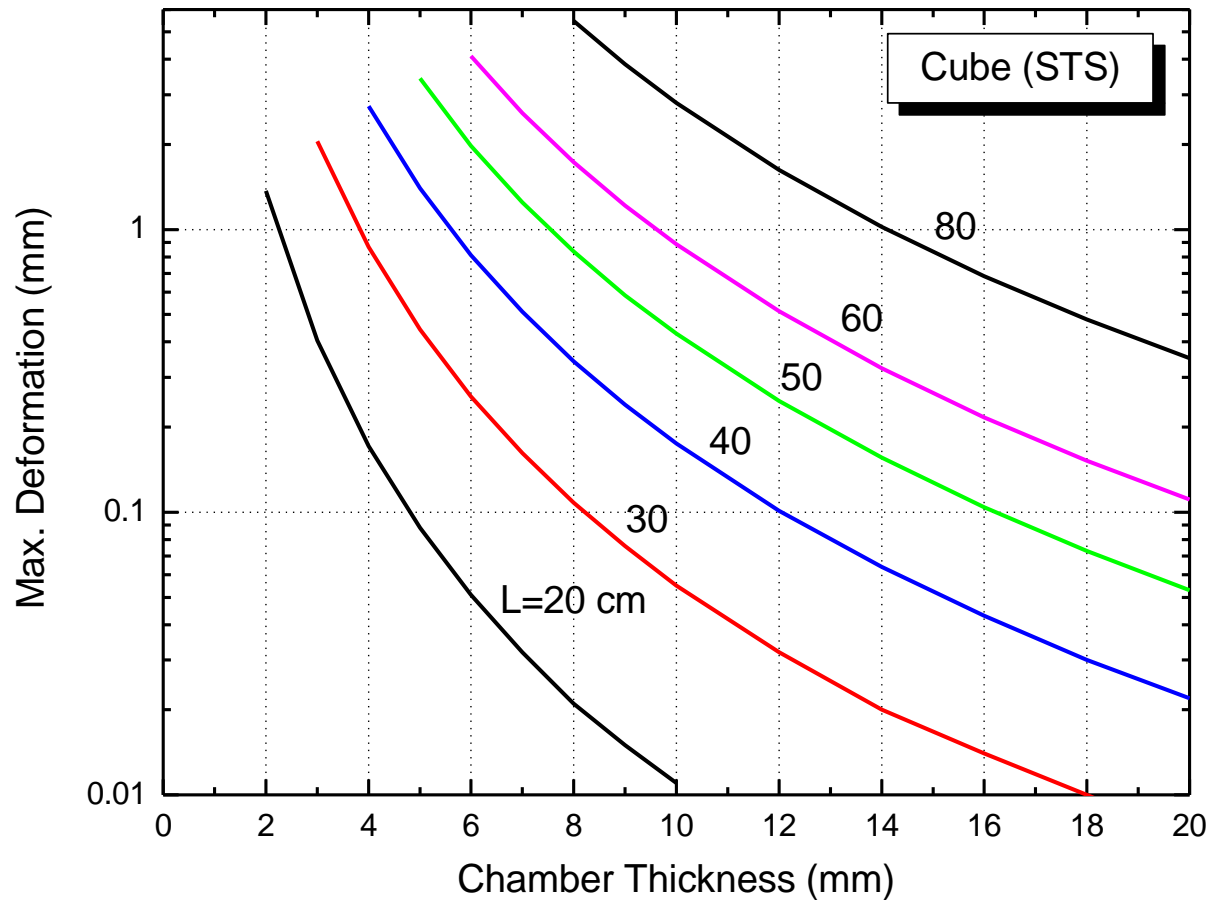
	Yield strength (0.2%)	Tensile strength
	MPa	MPa
316	206.8	517.1
316L	172.4	482.6
A6061-T6	241	289.4
A6063-T5 - T6	110	152

Stress : Cube



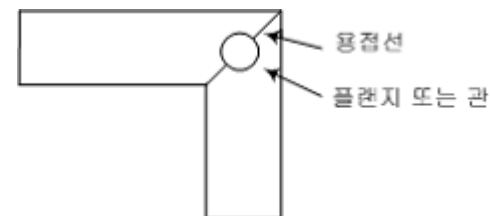
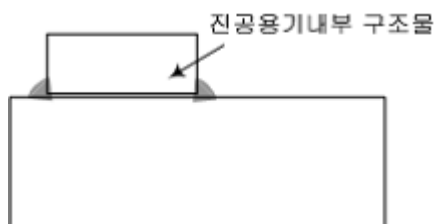
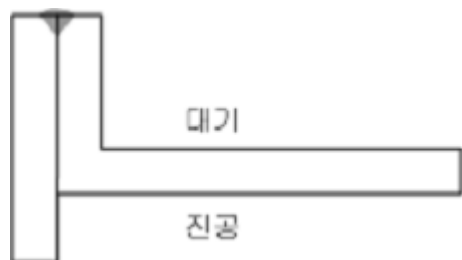
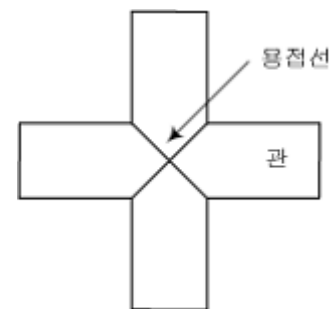
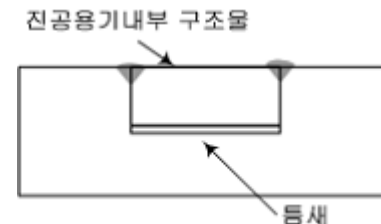
→ “진공공학(인상열 외)”

Deformation : Stainless steel



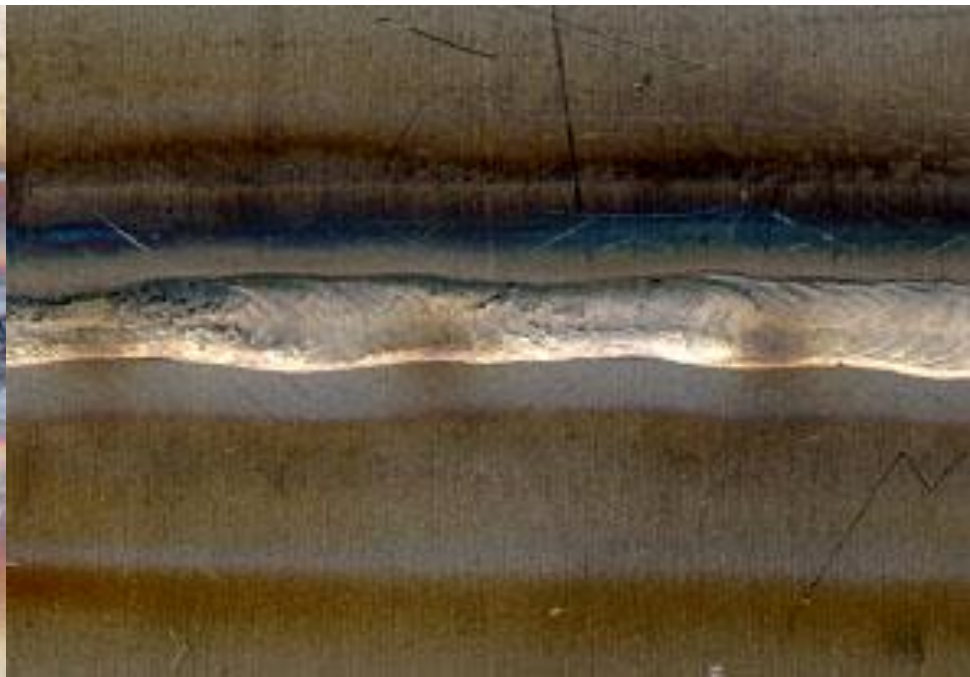
→ “진공공학(인상열 외)”

피해야 하는 용접

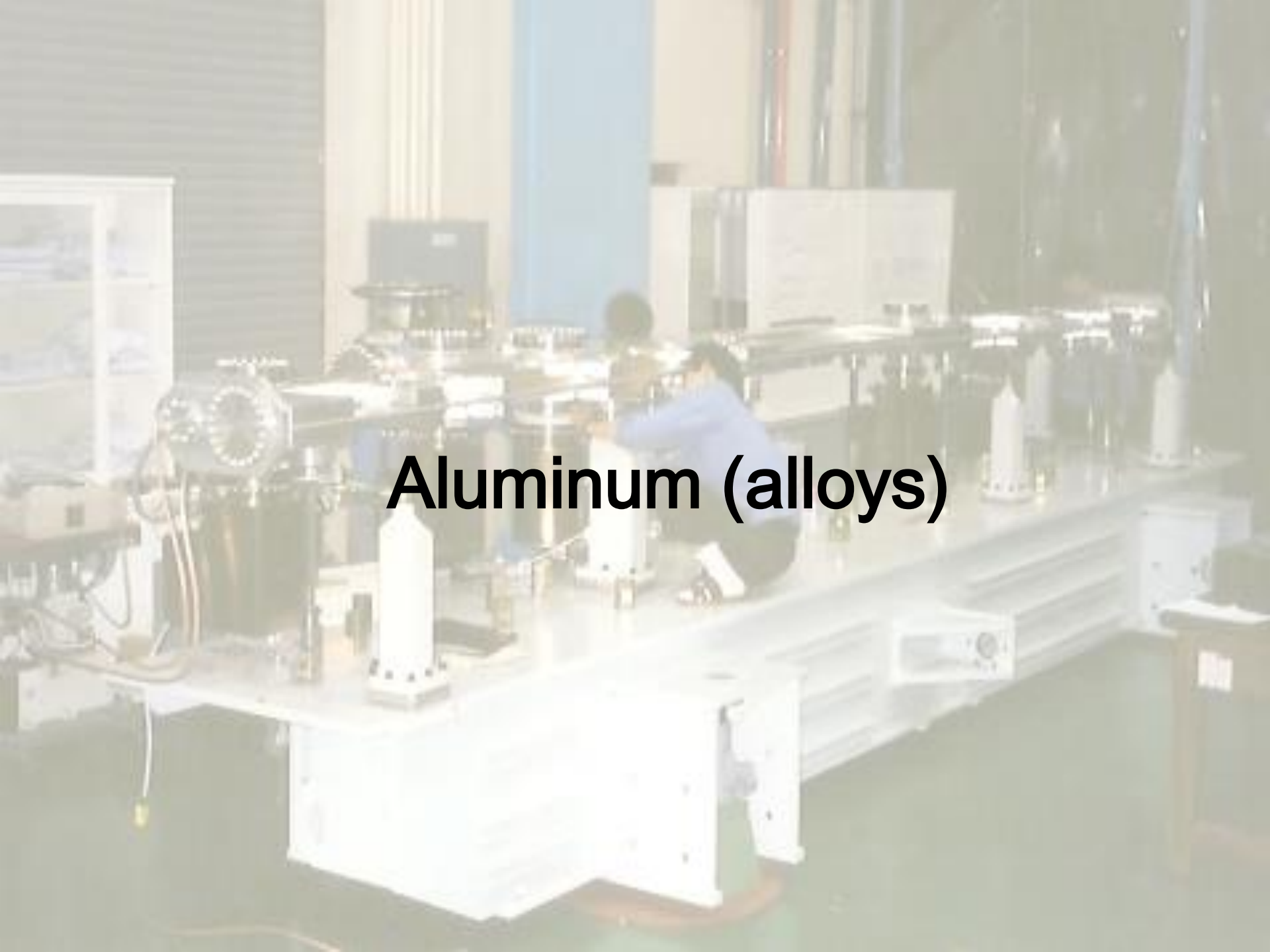


→ “초고진공 용접핸드북(KRISS)”

용접 품질



Aluminum (alloys)



Aluminum (alloys)

- 1xxx pure aluminum
 - ✓ > 99% Al by weight
 - ✓ A1050; suitable for metal gaskets
- 2xxx copper alloys
 - ✓ Duralumin; once the most common aerospace alloys (they were susceptible to stress corrosion cracking and are increasingly replaced by 7000 series in new designs.)
 - ✓ A2219; suitable for Conflat flange (weldable)

Aluminum (alloys)

- 3xxx manganese alloys
 - ✓ A3004; suitable for vacuum bellows
- 5xxx magnesium alloys
 - ✓ easy to machine, higher strength, good weldability
 - ✓ A5083; for a large scale chamber
- 6xxx magnesium and silicon alloys
 - ✓ Easy to machine and extrude
 - ✓ A6063; most common materials in vacuum technology
 - ✓ A6061; one of the most vacuum materials
 - ✓ A6060; extrusion

Aluminum (alloys)(진공용)

재 료	처 리	주요 합금성분 (%)		용 도
2219	T87, T852	Cu 5.8-6.8	Mn 0.2-0.4	플랜지
3004		Mn 1-1.5	Fe 0.25	벨로우즈
5052		Mg 2.2-2.8	Fe 0.4	진공용기, 벨로우즈
5083	H321	Mg 4-4.9	Mn 0.4-1	진공용기
6061	T5, T6	Mg 0.8-1.2	Si 0.4-0.8	진공용기, 벨로우즈
6063	T5, T6	Mg 0.45-0.9	Si 0.2-0.6	진공용기
6263		Mg	Si	진공부품 (티, 크로스, 엘보)
6951	T6	Mg 0.4-0.8	Si 0.2-0.5	벨로우즈

Tempering

- -F As fabricated
- -H Strain hardened (cold worked) with or without thermal treatment
 - -H1 Strain hardened without thermal treatment
 - -H2 Strain hardened and partially annealed
 - -H3 Strain hardened and stabilized by low temperature heating
- Second digit A second digit denotes the degree of hardness
 - -HX2 = 1/4 hard
 - -HX4 = 1/2 hard
 - -HX6 = 3/4 hard
 - -HX8 = full hard
 - -HX9 = extra hard
- -O Full soft (annealed)

- -T Heat treated to produce stable tempers
 - T5 Cooled from hot working and artificially aged (at elevated temperature)
A6061 160°C 18 h
 - (ex) 160°C for 18 h -> ambient cooling
 - 170°C for 12 h -> ambient cooling
 - 175°C for 8 h -> ambient cooling
 - 205°C for 3 h -> ambient cooling (1.5 h)
 - T6 Solution heat treated and artificially aged
 - (ex) ~ 180°C – 220°C for 6 h – 1h -> water cooling
- -W Solution heat treated only.

Aluminum (alloys)

- **A6063**-T5(T6)(T0)
 - ✓ The most widely used aluminum alloy
 - ✓ The most common materials in vacuum technology
 - ✓ Low outgassing rates
 - $< 5 \times 10^{-13}$ (mbar l/scm²) after bake ($< 180^{\circ}\text{C}$)
 - ✓ Easy to forming using **extrusion**, machining and welding
 - ✓ Alloy retains its strength after welding
 - ✓ ConFlat[®] flanges are made from A2219 with knife edge coated with TiC (TiN, CrN).
 - ✓ With A1050 metal gaskets

- Mainly used in UHV and forelines
- Bakeout temperature
 - Max 180°C
 - In-general 100 – 150°C
- Melting point; 660°C ($P_{\text{vapor}}=10^{-8}$ mbar)
- Entirely non-magnetic
- High thermal and electrical conductivity

Aluminum (alloys)

- **A6061**-T5(T6)

- ✓ The most widely used aluminum alloy
- ✓ Low outgassing rates
 - ✓ $< 5 \times 10^{-13}$ (mbar l/scm²) after bake ($< 180^\circ\text{C}$)
- ✓ Easy to forming using *machining* and welding
- ✓ Alloy retains its strength after welding
- ✓ ConFlat[®] flanges are made from A2219 with knife edge coated with TiC (TiN, CrN).
- ✓ Large chambers which would be difficult to heat treat in to T6 condition are often made from **5083**.

Aluminum (alloys)

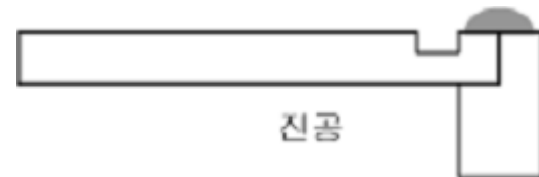
- Initial outgassing rate is higher($\sim 5x$) than that of SST.
 - ✓ The desorption rate of water vapor from the surface of aluminum is slower than stainless steels which gives it different initial pump down characteristics.
- **Anodizing**
 - ✓ a common surface treatment.
 - ✓ produces **hard inert** surface, but q becomes higher, $\sim x10$.
 - ✓ Not suitable for UHV applications
- ✓ Surface oxidation
 - ✓ Suitable for UHV and XHV applications

알루미늄 용접

- Al alloys require special attention to both weld design and weld technique.



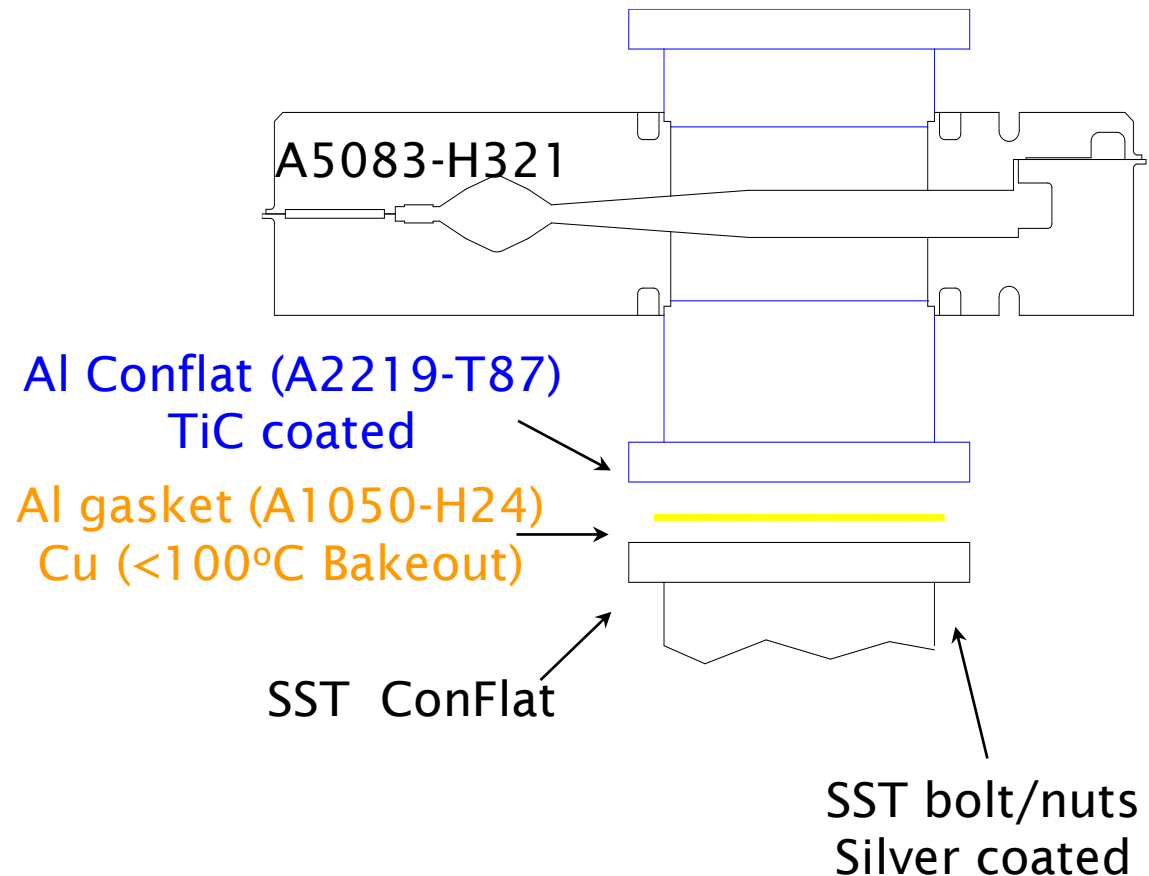
ac-TIG welding
(with filler metal)



→ “초고진공 용접핸드북
(KRISS)”

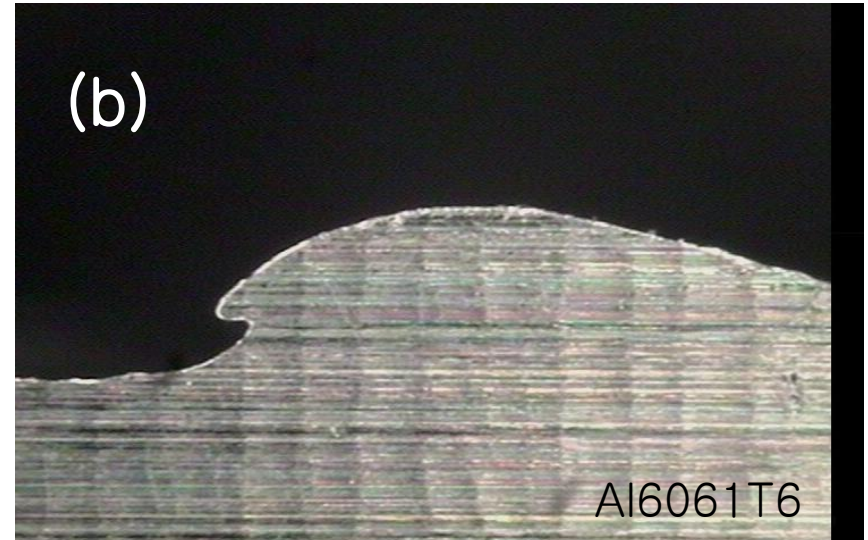
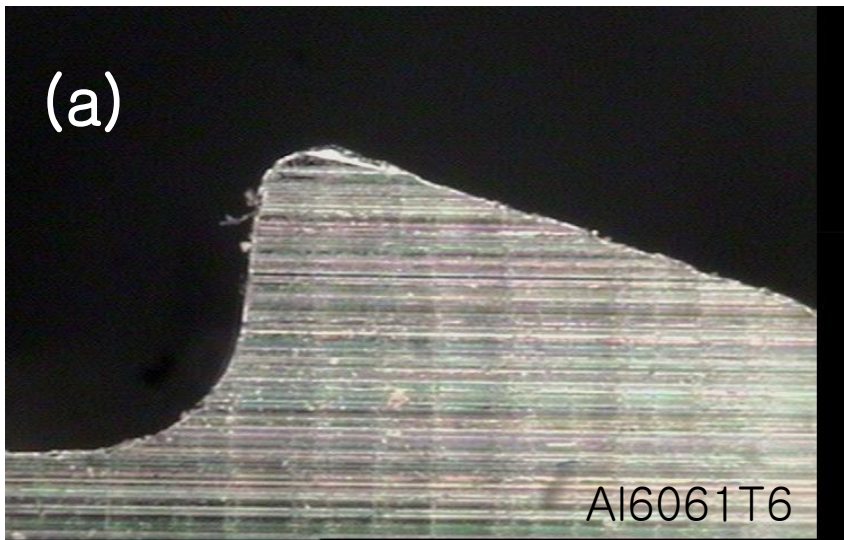
Vacuum Seals

- Al/SST hybrid ConFlat system



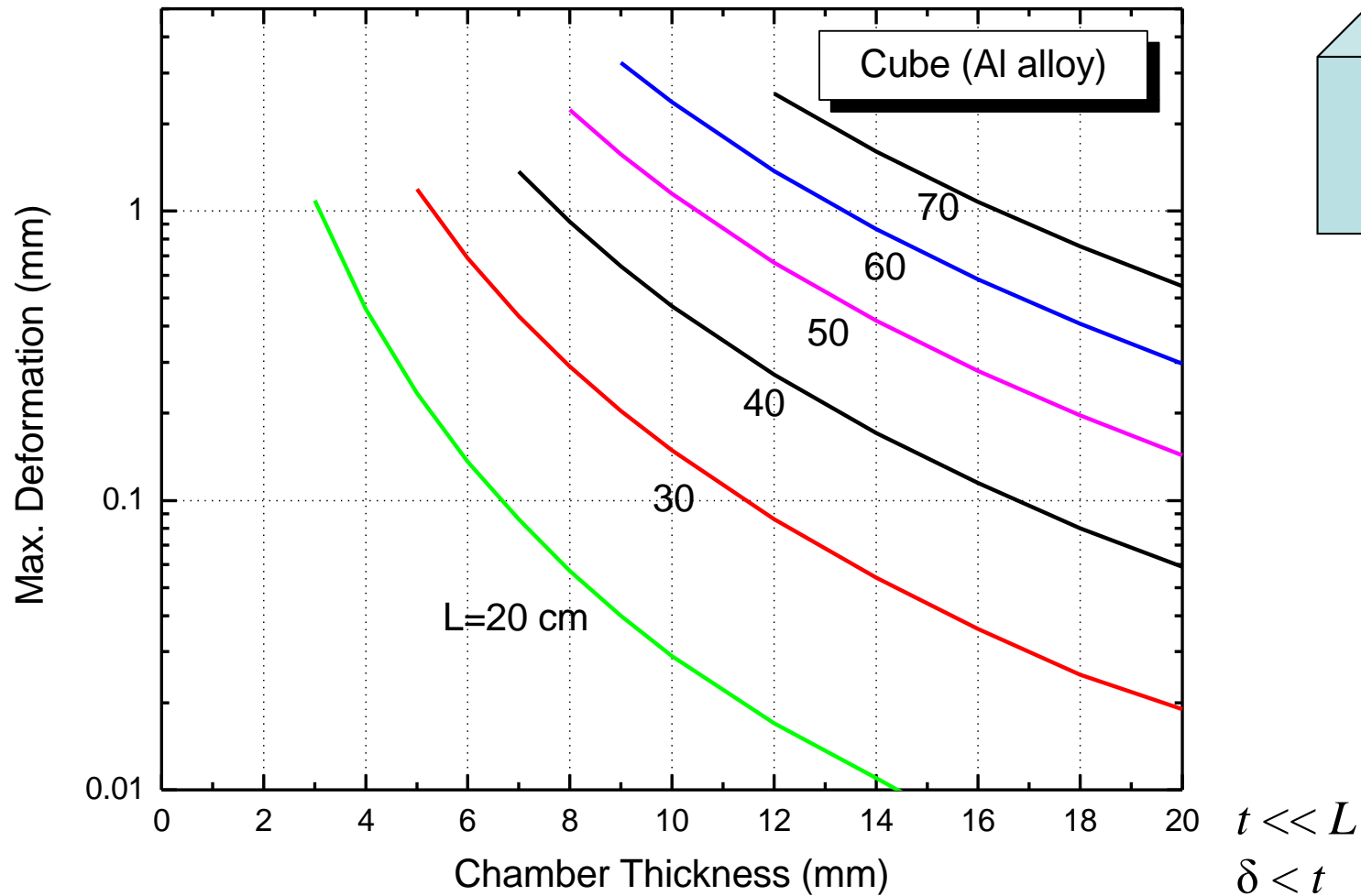
Vacuum Seals

- Al/SST hybrid ConFlat system

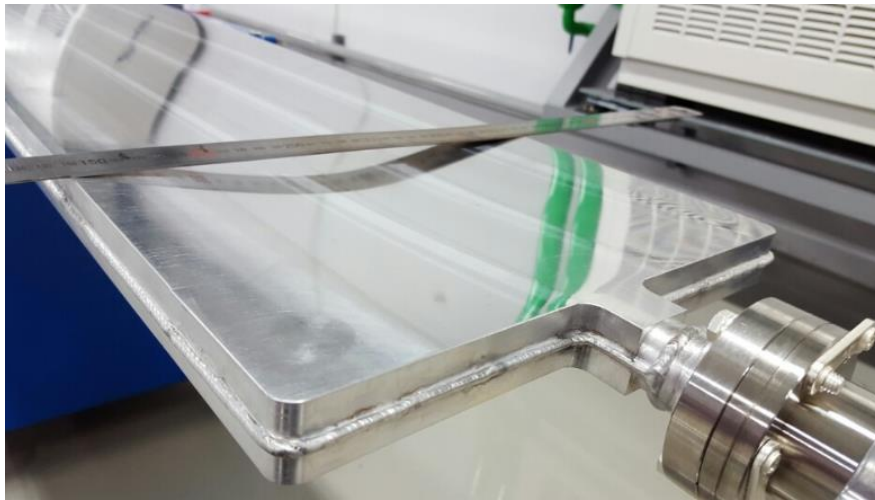


Knife edge of Al flange (before use (a) and after use (100 times) (b)]

Deformation : Al alloy



Deformation : Al alloy

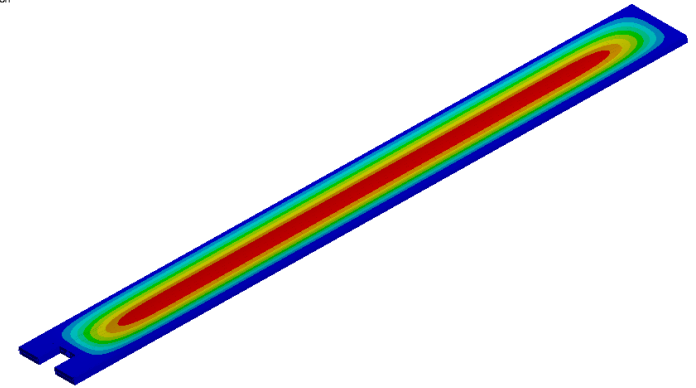


(Thickness=3 mm, width = 270 mm)

C: Static Structural
Total Deformation
Type: Total Deformation
Unit: mm
Time: 1

16.119 Max
14.326
12.537
10.746
8.955
7.164
5.373
3.582
1.791
0 Min

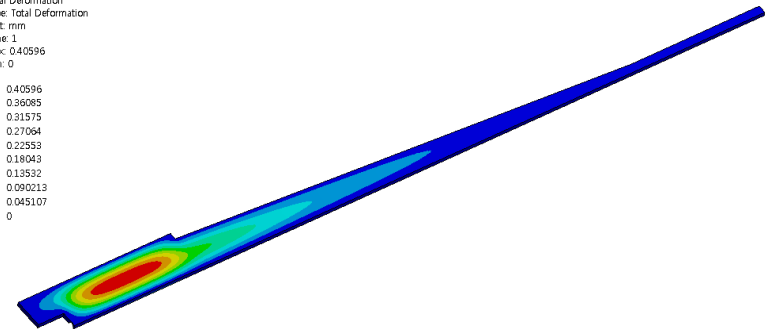
16 mm deformation (3T)



E: SI
Total Deformation
Type: Total Deformation
Unit: mm
Time: 1
Max: 0.40596
Min: 0

0.40596
0.36085
0.31575
0.27064
0.22553
0.18043
0.13532
0.090213
0.045107
0

0.4 mm deformation (4T)





SST - Al 비교

		Stainless steel	Aluminum alloy
Vacuum characteristics	Outgassing rate	depends on surface treatments	
	Preinstallation bakeout	950℃	< 210℃
	<i>In situ</i> bakeout	150 - 450℃	< 150℃
Mechanical characteristics	Mechanical strength	higher (×1.5)	
	Thermal expansion coefficient		higher (×2)
	Thermal conductivity		higher (×15)
	Shaping by machining		easier
	Shaping by extrusion		easier
	Shaping by welding	easier	
	Quality of demountable seals	better	
Other characteristics	Magnetic property	not entirely non-magnetic	completely non-magnetic
	Residual radioactivity		lower

Copper (alloys)

A photograph of a laboratory or industrial testing environment. In the center, a large, white, L-shaped machine is the focal point. It has various mechanical components, including a large circular gauge or motor on the left side. A person wearing a blue long-sleeved shirt and dark pants is crouched behind the machine, working on it. To the right of the machine, there is a control panel with a digital display and several buttons. The background shows a blue wall and some other equipment. The text "Copper (alloys)" is overlaid in the center of the image.

Copper (alloy)

- High **thermal and electrical conductivity**
 - Suitable for electrical feedthrough
 - Suitable for thermal/radiation absorber
 - Suitable for cryogenic applications
- Hydrogen embrittlement
 - For HV and UHV, coppers(alloys) with oxygen free or reduced oxygen contents are required.
- Bakeout; Up to 300°C in vacuum
- Cold welding; **OFHC gasket** for ConFlat flanges
- Joining techniques; brazing, soldering, welding
- Outgassing rate; $\sim 10^{-9}$ mbar liter/sec cm²

Copper alloy

Brass(Zinc alloy) and **Bronze**(Tin alloy)

- High vapor pressure at high temperatures
(Used in vacuum systems where $T < 100^{\circ}\text{C}$)
 - Easy to machine
 - Cheap
- Commonly used in rough and high vacuum chambers and fixtures.
- Common joining techniques: Soldering
- Outgassing rate
 $\sim 10^{-7}$ mbar liter/sec cm^2

Copper alloy

- To increase its strength
 - OFC + Al_2O_3 (0.1-0.5%) GlidCop
 - Yield strength(at 0.2% offset) > 200 MPa
(OFC < 100 MPa)
 - OFC + Ag or $\text{Au}_{0.2\%}$ Expensive (x 4)
 - OFC + Zr High outgassing rate
 - OFC + Be Brazing(X), EBW(O)
 - OFC + Cr



Outgassing rates

$q(\text{mbar l/s cm}^2)$	@ 10 hour	Baked
Aluminum alloy	5×10^{-8}	$< 5 \times 10^{-13}$
Aluminum(anodized)	3×10^{-7}	5×10^{-10}
Stainless steel	1×10^{-8}	$(2-6) \times 10^{-12}$
Mild Steel	2×10^{-7}	$< 1 \times 10^{-12}$
Brass	6×10^{-7}	
Copper	5×10^{-9}	1×10^{-12}
Copper (OFHC)	2×10^{-9}	$< 1 \times 10^{-12}$

A photograph of a laboratory or workshop. A person in a blue shirt is working at a large white table. The table is covered with various scientific instruments, including a large circular device on the left, several smaller devices, and a large white container. The background shows a blue wall and some equipment. The text "Other metals" is overlaid on the image.

Other metals

Gold and Silver

- Gold
 - ✓ Very low vapor pressure
 - ✓ Used as **metal gaskets**, surface seals in valves
as coating for electrical conductors
 - ✓ Used as brazing **filler** alloys
 - ✓ Cu/Au or Cu/Au/Pd
- Silver
 - ✓ Very low vapor pressure
 - ✓ Silver **plated** bolts/nuts to reduce friction/cold weld.
 - ✓ High oxygen permeation rate through silver at high temperature.

Titanium

- Very active metal
- Easily react with O_2 , N_2 at $> 150^\circ C$
 - ✓ Weld should be done with inert gas environment
- Used as *metalizing materials* for brazing
- *TSP*(Titanium sublimation pump)
 - ✓ Sublimation at $\sim 1,350^\circ C$
 - ✓ $\sim 5 \mu m$ coating for 1 hr
- *Ion pump*: *Cathode* material
- *NEG alloy* (Ti-Zr-Fe)
- *Chamber materials*: Very low outgassing rate

Indium

- Melting point; 156°C
 - Not suitable for bakeable UHV applications
- Very low vapor pressure
- Very soft
- High thermal conductivity
- Thus indium may be used as
 - ✓ *vacuum seal* for UHV at cryogenic applications
 - ✓ *thermal conductors* between two different metals

Ceramics



Ceramics

- Ceramics
 - ✓ Non-metal, inorganic materials
 - ✓ Mainly used as **insulators** in vacuum technology
- Three types of ceramics
 - ✓ Pure-oxide ceramics
 - ✓ Silicate ceramics
 - ✓ Glass-ceramics

Ceramics

- Pure oxide ceramics
 - ✓ Alumina, Zirconia, Beryllium oxide,...
 - ✓ Alumina (Al_2O_3)
 - Mostly used ceramics
 - Max temperature; 1,800°C
 - > 92% in vacuum technology
 - Can be brazed
 - Mainly used as electrical *feedthroughs/insulator*
 - Bakable upto 350-550°C
 - Tensile strength 25 kpsi (96% density)
 - ✓ Sapphire (monocrystalline Al_2O_3)
 - UV and IR transparent
 - Used as *vacuum window*

Ceramics

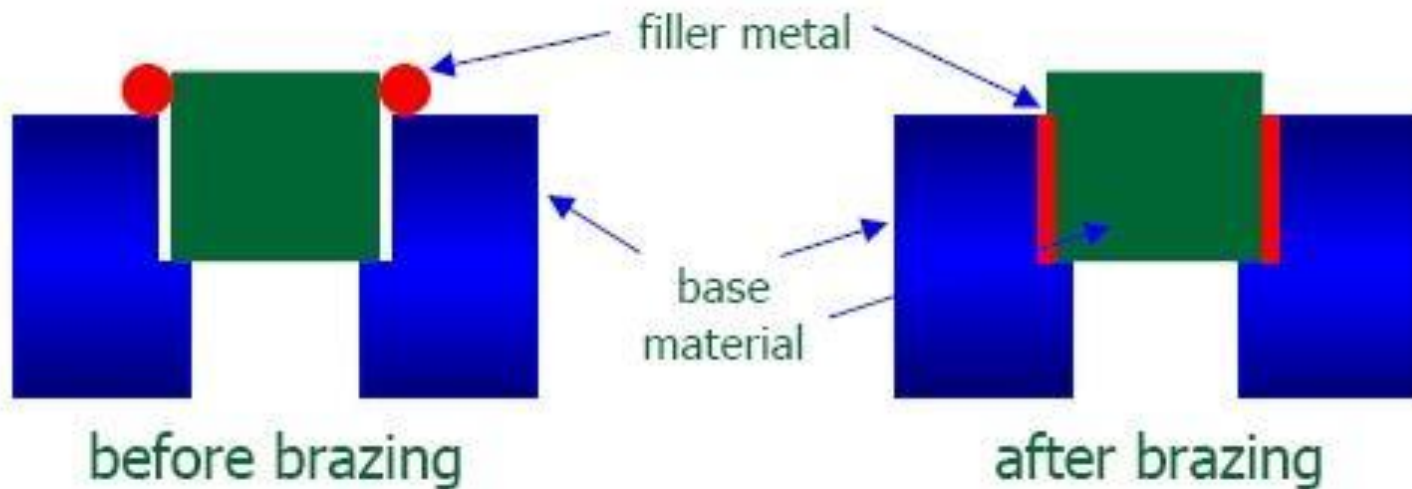
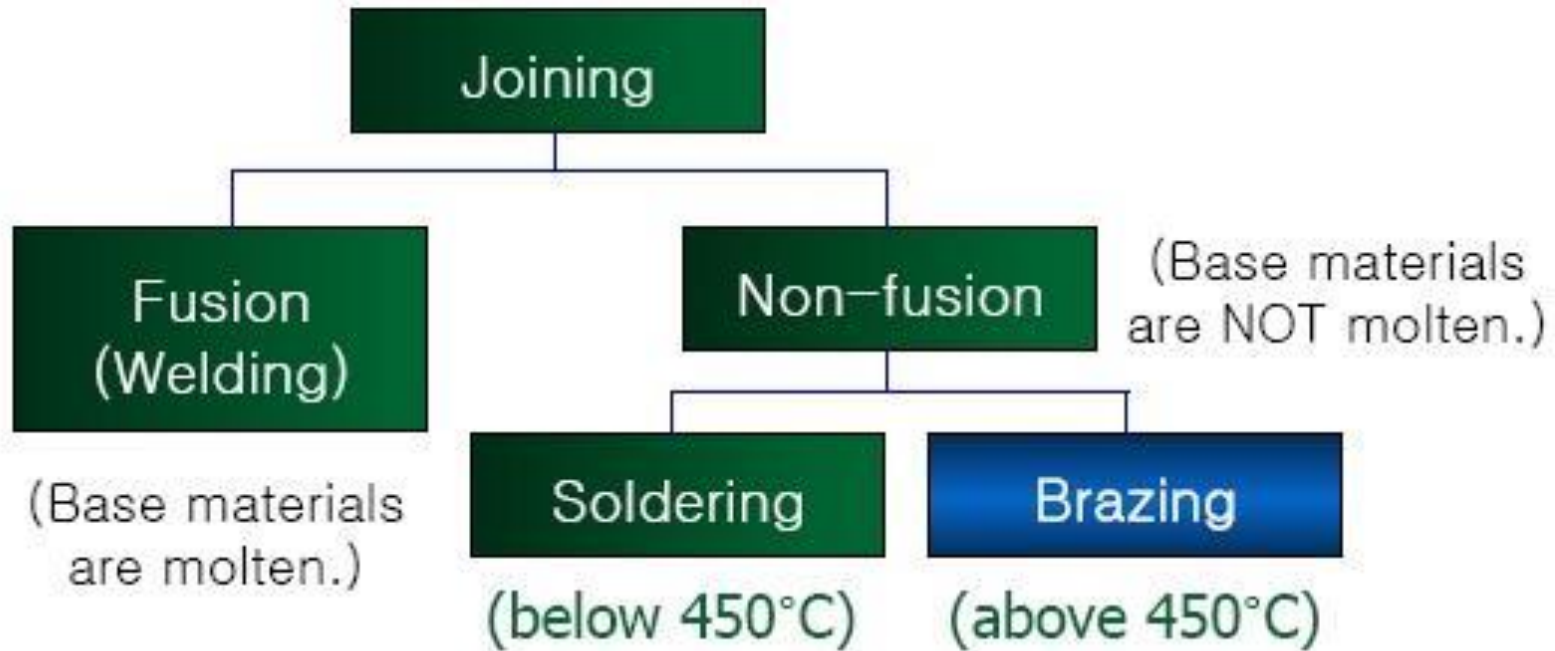
- Silicate ceramics
 - ✓ Steatite (MgO-SiO_2)
 - Max temperature $1,000^\circ\text{C}$
 - Tensile strength 15 kpsi
- Glass-ceramics
 - ✓ Crystalline ceramic
 - Can be **machined** with standard tools
 - Macor®, Corning 9658

Ceramics

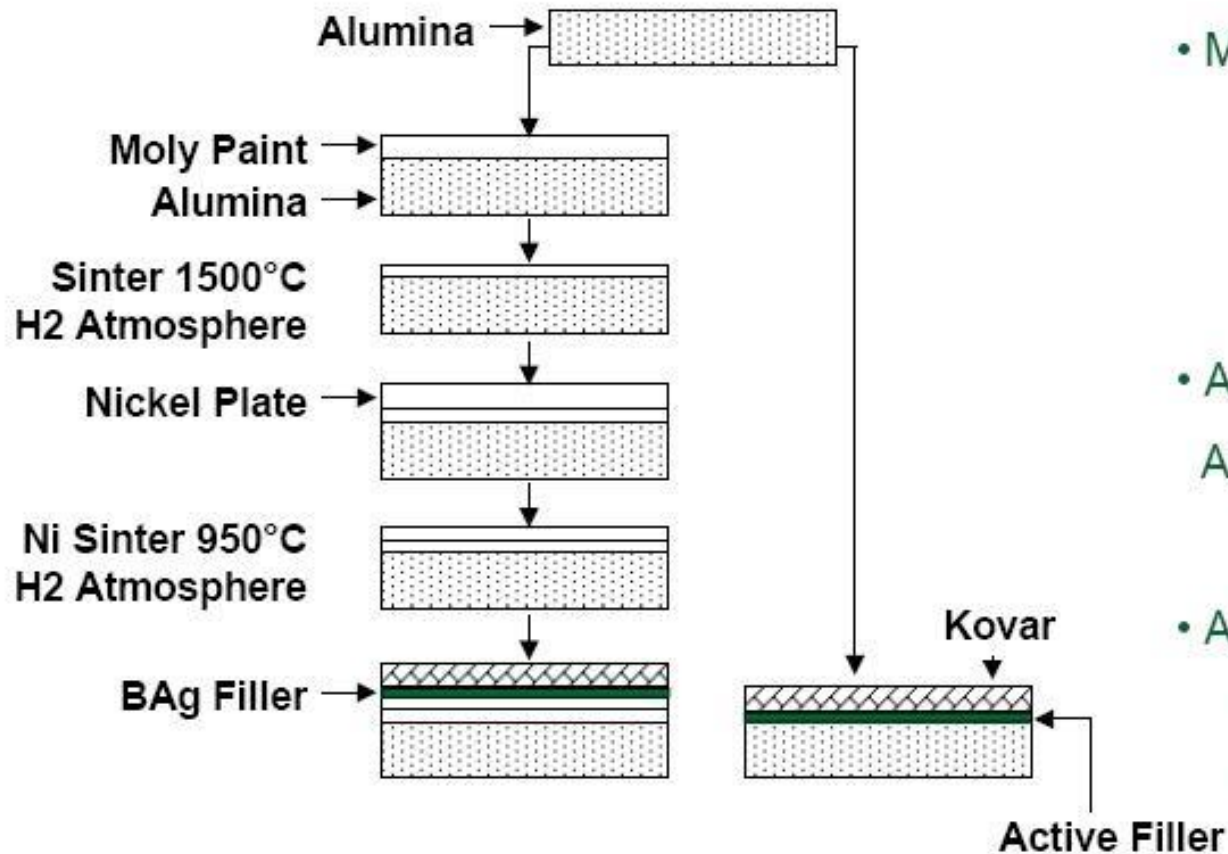
- Kovar

- ✓ Fe-Ni-Co alloy (thermal expansion ~ glass)
- ✓ Magnetic
- ✓ Intermediate material between ceramic and metal for brazing joint

Permanent vacuum joints



Brazing



- Metallization Method
 - Mo/Mn Paint Sintering
 - Vapor Deposition
- Active Brazing
$$\text{Al}_2\text{O}_3 + 3\text{Ti} \rightarrow 2\text{Al} + 3\text{TiO}$$
- Active Soldering
 - Brushing, Vibration required
 - No Spreading

Zeolite

- Very porous alumina silicates with alkali metals
- Mainly used as **drying** agents
- Mostly used as vapor adsorbing agents in vacuum
- *Sorption pump*
 - Liquid nitrogen cooled
 - Regeneration by heating

A photograph of a laboratory or industrial setting. A person in a blue shirt is kneeling at a large white table, working with a piece of glass. The table is equipped with various glass processing equipment, including a large circular furnace or annealing oven on the left, several glass bottles, and other specialized tools. The background shows a blue wall and some equipment. The word "Glass" is overlaid in the center of the image.

Glass

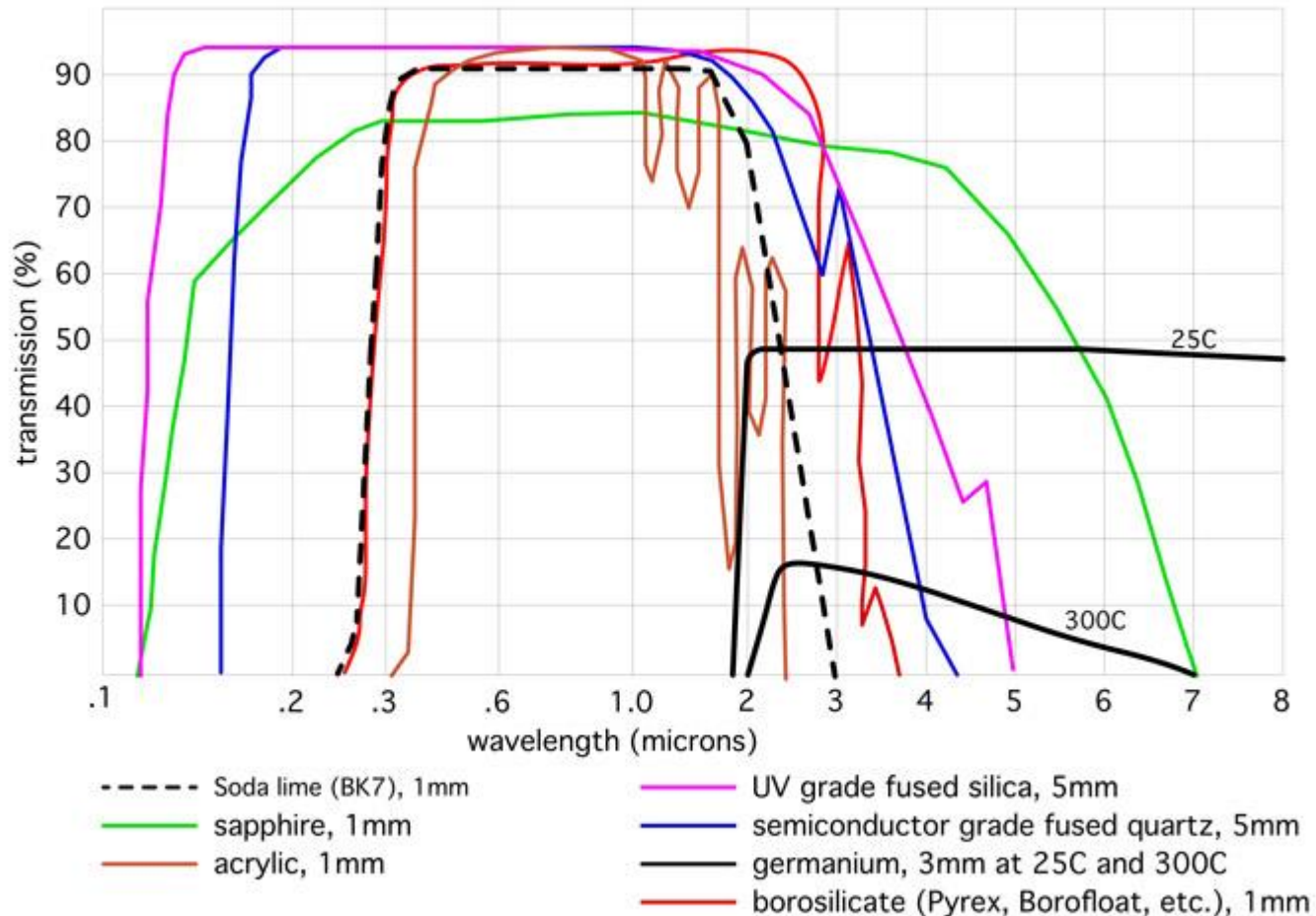
Glass

- Glasses
 - ✓ Non-metal, inorganic materials
 - ✓ Mainly used as *vacuum windows*
 - ✓ Also used in helium permeation leaks
- Three types of glasses
 - ✓ Soft glass: $60 \cdot 10^{-7} \text{ K}^{-1} - 120 \cdot 10^{-7} \text{ K}^{-1}$
 - ✓ Hard glass: $< 50 \cdot 10^{-7} \text{ K}^{-1}$
 - ✓ Quartz glass: $\sim 5 \cdot 10^{-7} \text{ K}^{-1}$

Glass

- Soft glass(SiO_2 + Alkali)
 - 65–70% SiO_2 , 2.5–15% Na_2O , 5–15% CaO
- **Hard** glass ($\text{SiO}_2 > 70\%$ + Boron)
 - Corning 7056, Duran, Pyrex
 - Most common glass in vacuum tech.
- Quartz (SiO_2 100%)
 - Used as optical vacuum window

Optical transmission



Thickness affects: Making the material thicker only slightly affects transmission %, but can significantly affect how deep the transmission goes into the UV and IR

[RAYOTEK^{inc.}]

Plastics



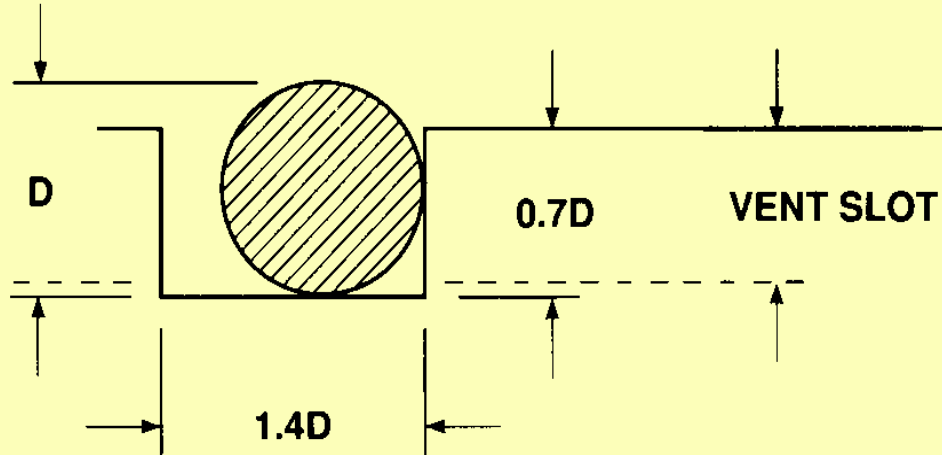
Plastics

- Elastomers
 - Elastic , Rubber-like materials
 - Used as reusable vacuum **gaskets** for RV, MV, HV, or (UHV)
 - Buna-N, Viton, Kalez, ...
- Thermoplastics
 - Thermally reversible
- Duroplastics
 - Thermally irreversible
 - **Epoxy**
 - Good adhesion with metals, glasses, ceramics

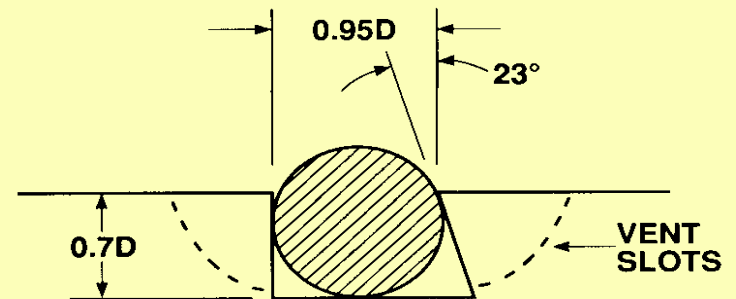
Outgassing rates

Outgassing rate (TorrLs⁻¹cm⁻²)	@ 10 h	Baked
Viton (FKM)	5×10^{-8}	5×10^{-10}
Buna N (NBR)	2×10^{-6}	4×10^{-8}
Epoxy (Shell Epon)	1×10^{-6}	8×10^{-8}
Teflon (poly'fluoro'lene/PTFE)	8×10^{-8}	8×10^{-9}
Nylon (polyamide)	3×10^{-7}	6×10^{-9}
PVC	3×10^{-7}	8×10^{-8}

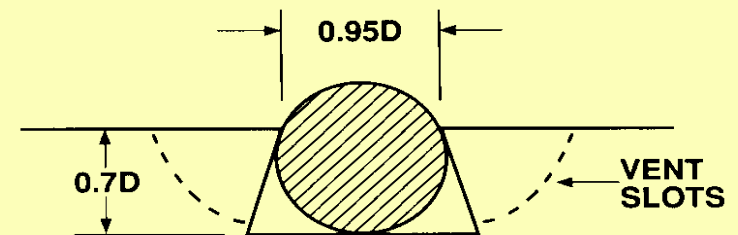
O-ring seal



Retaining the O-ring



A. DOVETAILED O-RING GROOVE



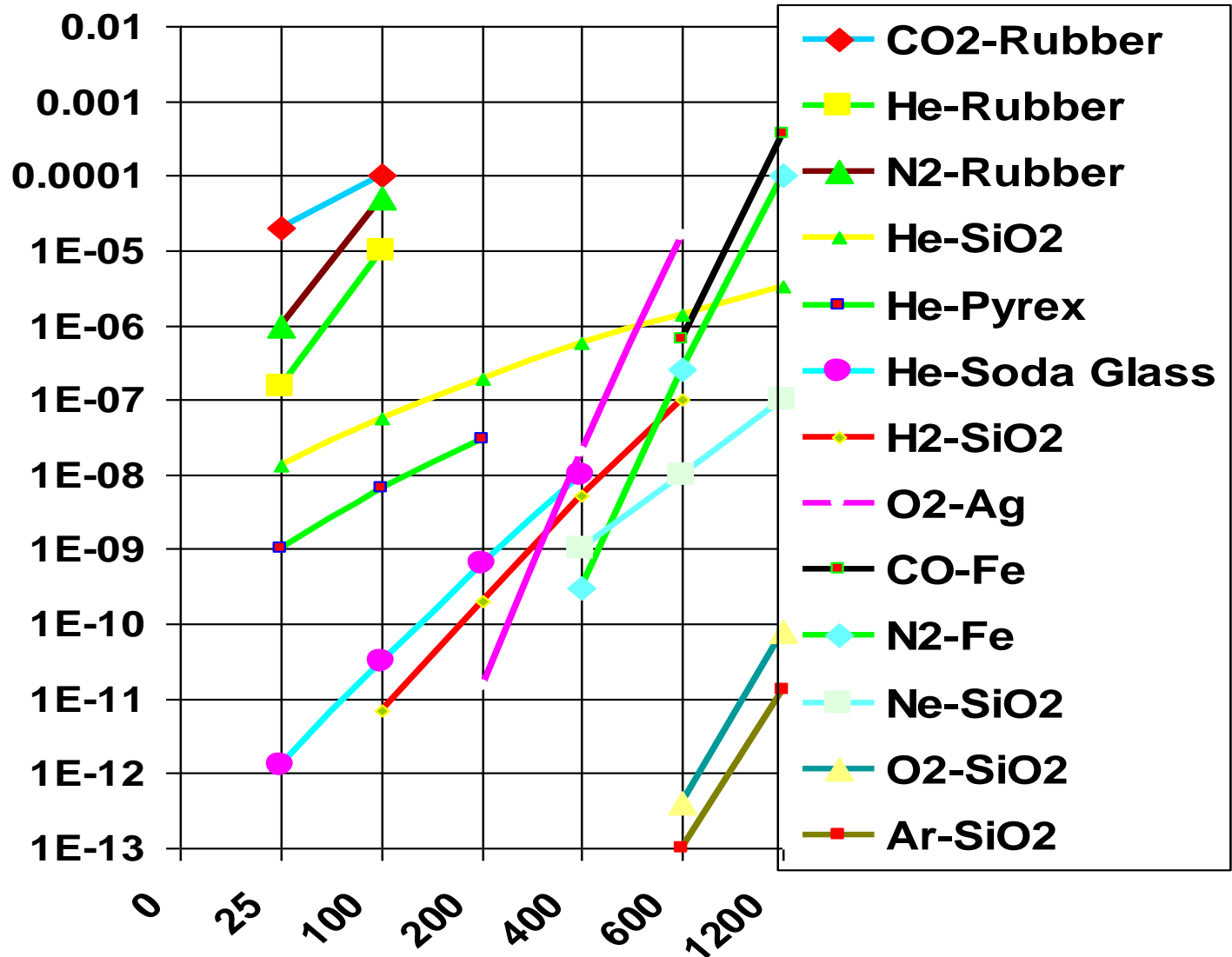
B. DOUBLE DOVETAILED O-RING GROOVE

DOVETAILING OF O-RING GROOVES

Permeation

- Permeation through solid materials involves
 - sorption, diffusion and desorption.
- Materials have permeation rates for different gases specific to that material.
 - steels have higher permeation rates with higher carbon content;
 - copper has low permeation for all gases;
 - aluminum has low permeation for hydrogen;
 - silver has high permeation rate for oxygen at high temperature
 - palladium has high permeation rate for hydrogen
 - polymers are permeable to all gases.

Permeation



Permeation

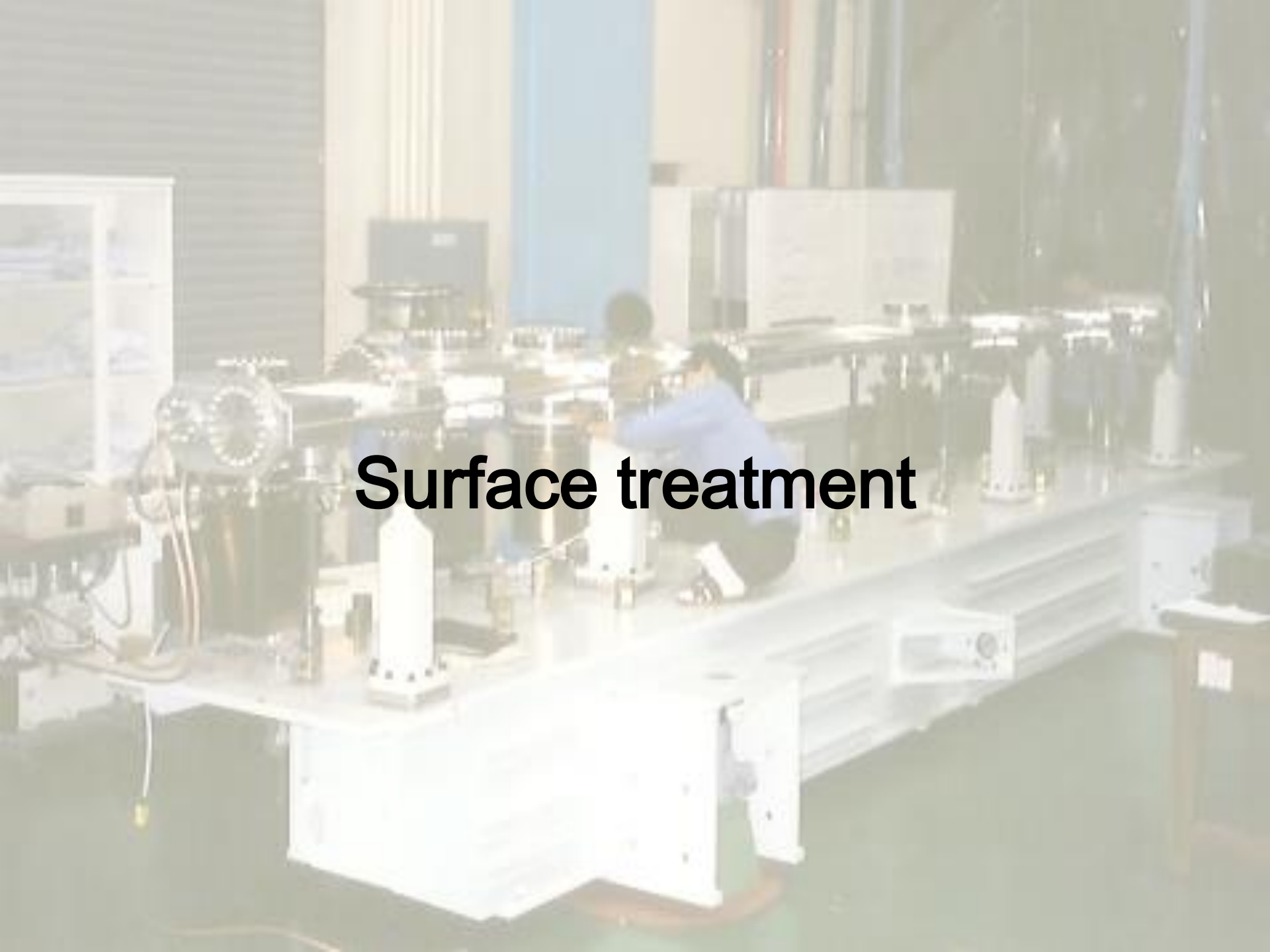
- Permeation is a strong function of temperature.
- Permeation gives additional gas loads.
 - Modifies the chamber environment(residual gases)
 - Affecting chemistry in vacuum process
 - Limits the ability to reach ultra high vacuum.
- Calibrated leaks
 - helium permeation through pyrex or quartz.

Helium permeation through elastomers

Polymer	Permeation rates	
	std-cc/s cm ² (×10 ⁻⁷) at 1 atm	
	25°C	150°C
Viton	1.3	49
Buna N	0.8	25.2

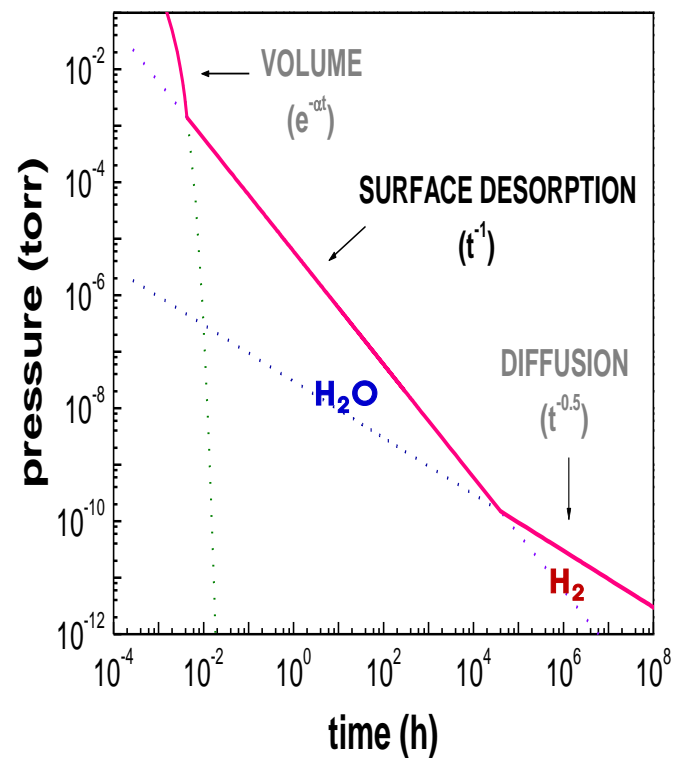
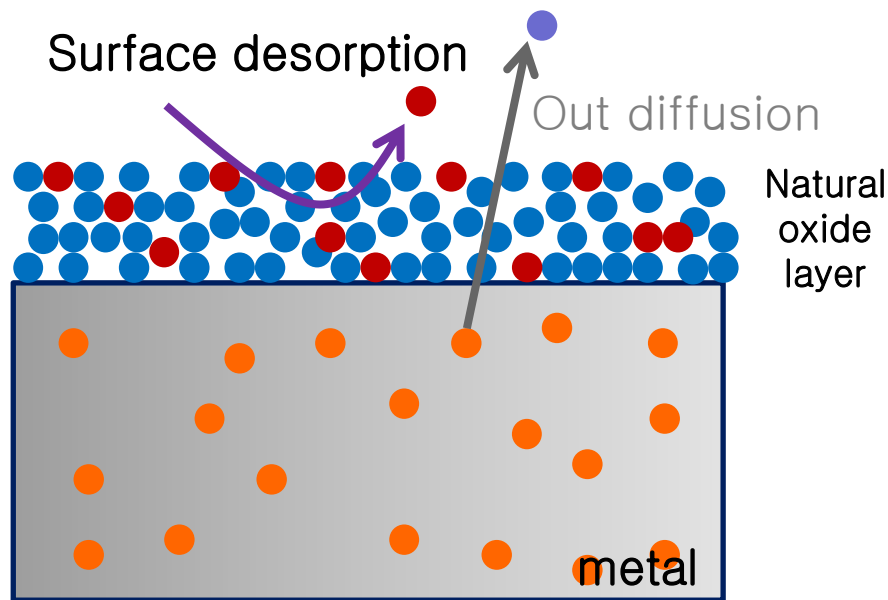
After several minutes helium will begin to permeate through the elastomer O-ring and show up as a leak.

Surface treatment

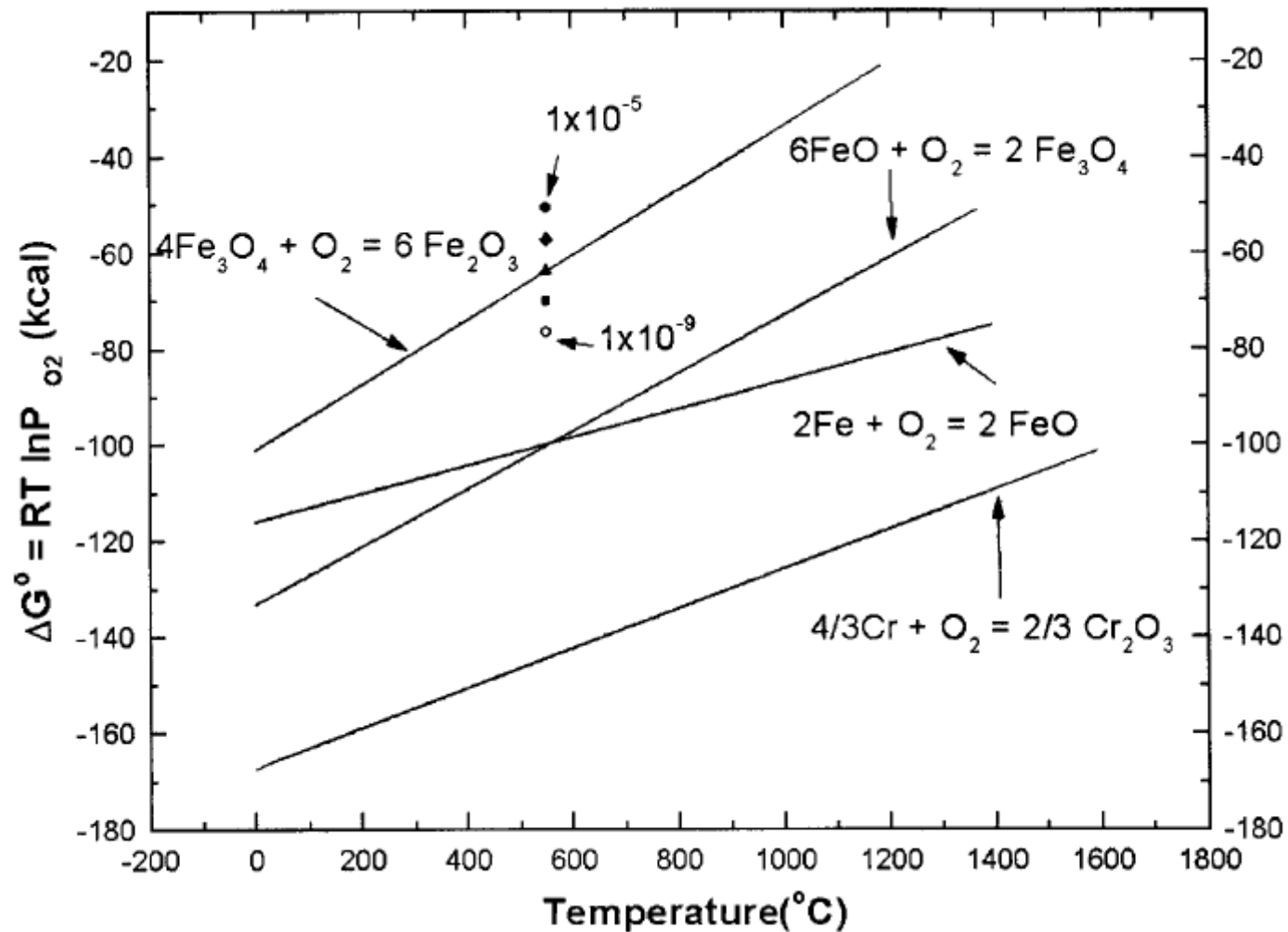


일반 스테인리스강의 진공특성

- ✓ 일반 금속(스테인리스강)의 표면은 다공질의 산화막이 형성되어 있음.
- ✓ 공기 노출 시 **다공질 표면**에 물과 같이 흡착성이 강한 입자가 다량으로 흡착됨.
- ✓ 진공 배기하면 흡착된 **물 분자가 서서히 방출**되므로 압력은 시간의 역수에 비례하며 ($p \propto 1/t$) 느리게 떨어짐.

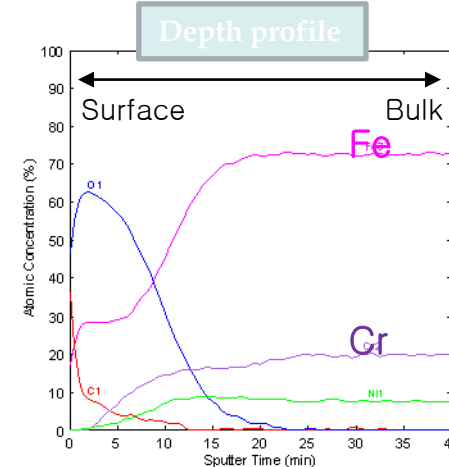
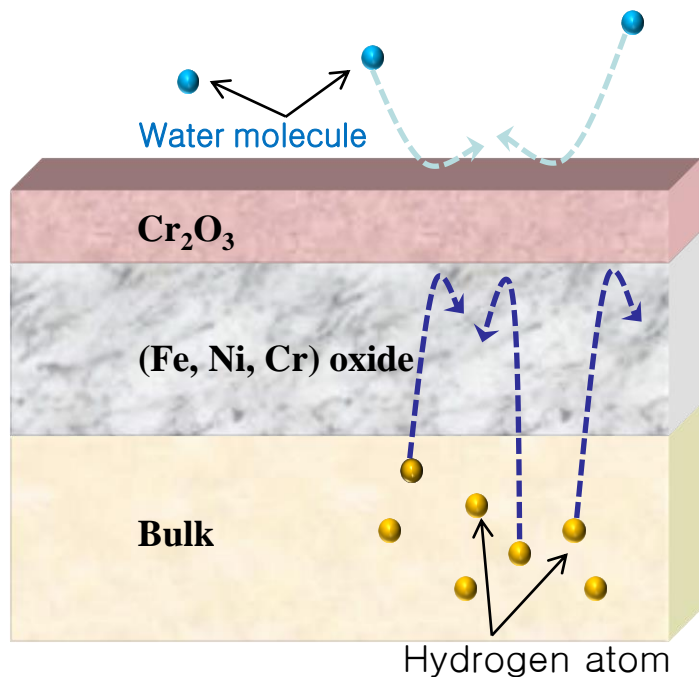


Ellingham Diagrams

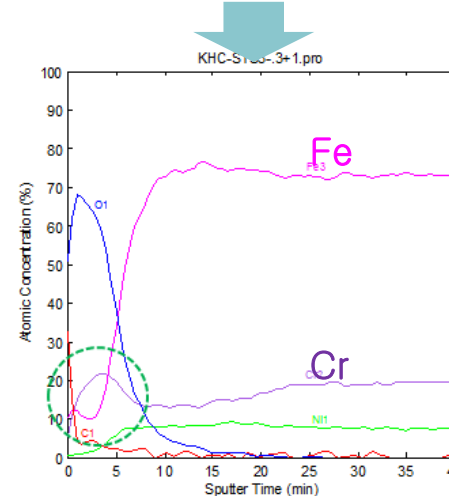


Vacuum Thermal Oxidation (VTO)

- ✓ VTO 표면 처리를 하면 다공질 산화막 위에 **순수한 크롬산화막이 형성됨.**
- ✓ 크롬산화막에 의해 공기 노출 시 물 분자의 흡착량이 크게 감소하고 흡착된 물 분자도 비교적 쉽게 떨어져나가므로 **급속 배기** 가능함.

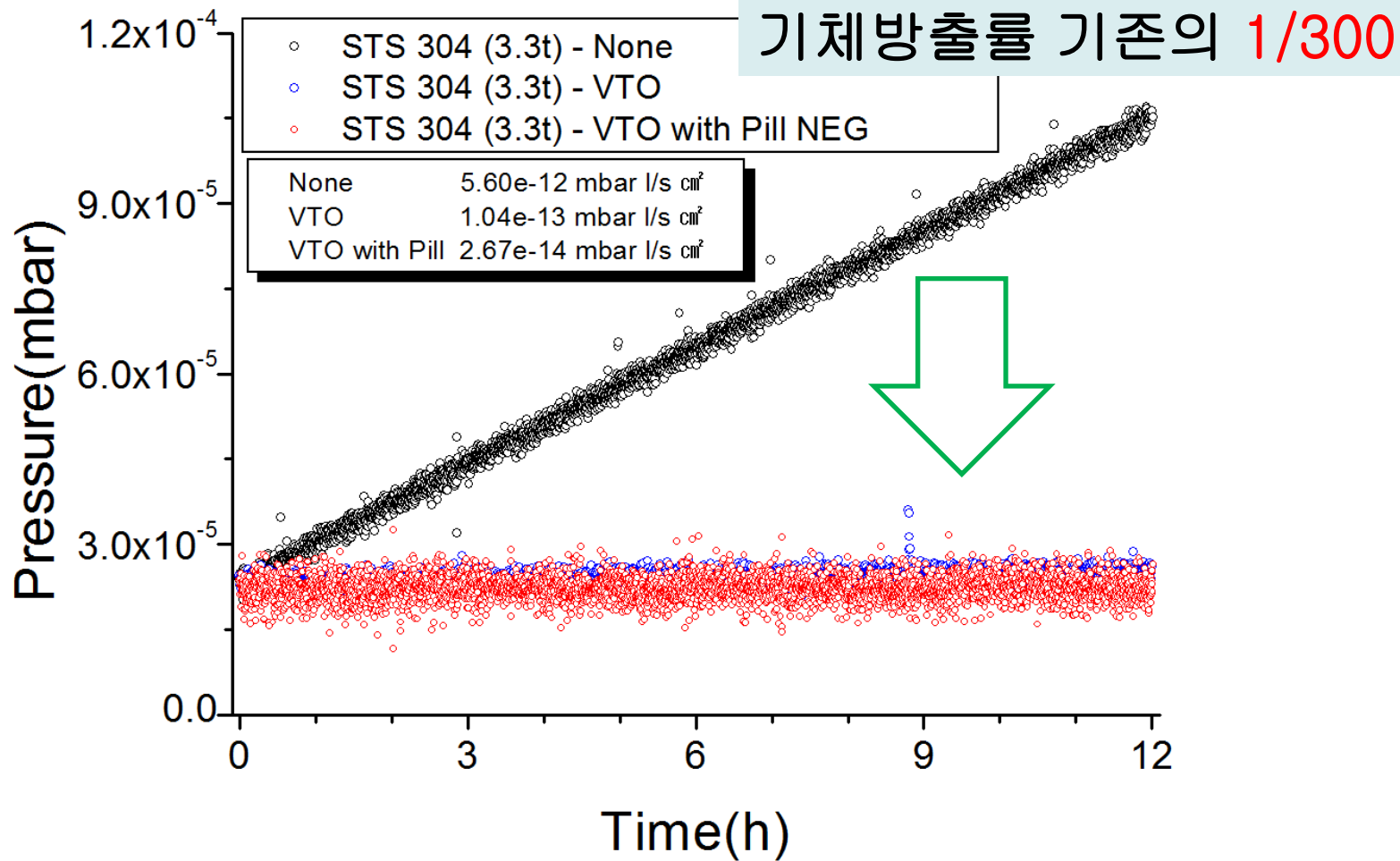


일반 (STS304)
:표면에 철산화물 많음



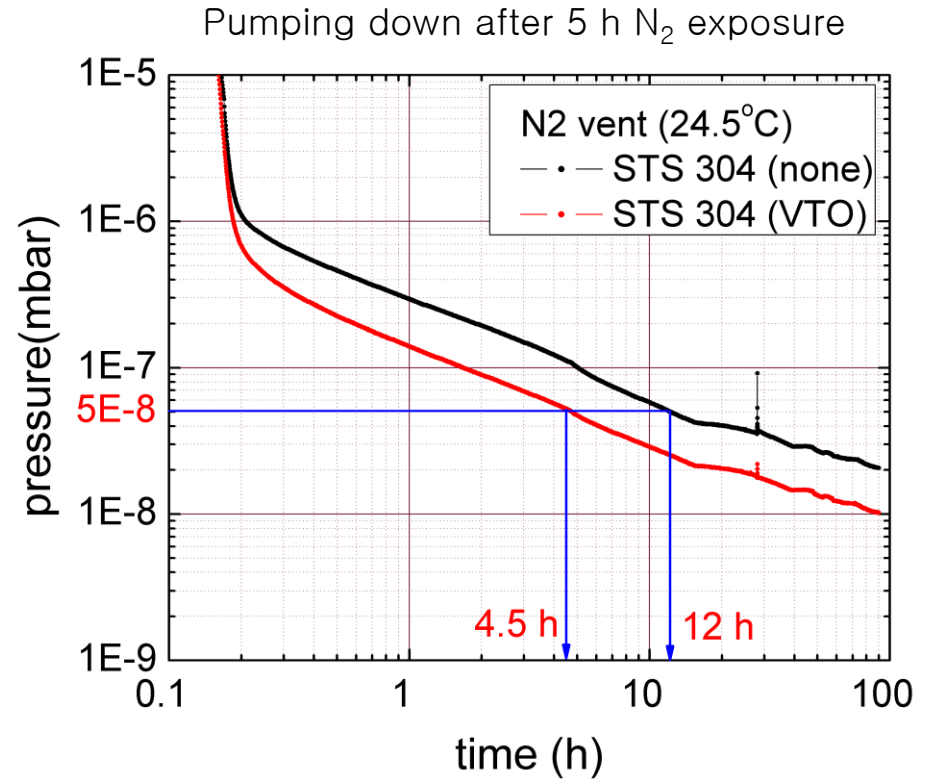
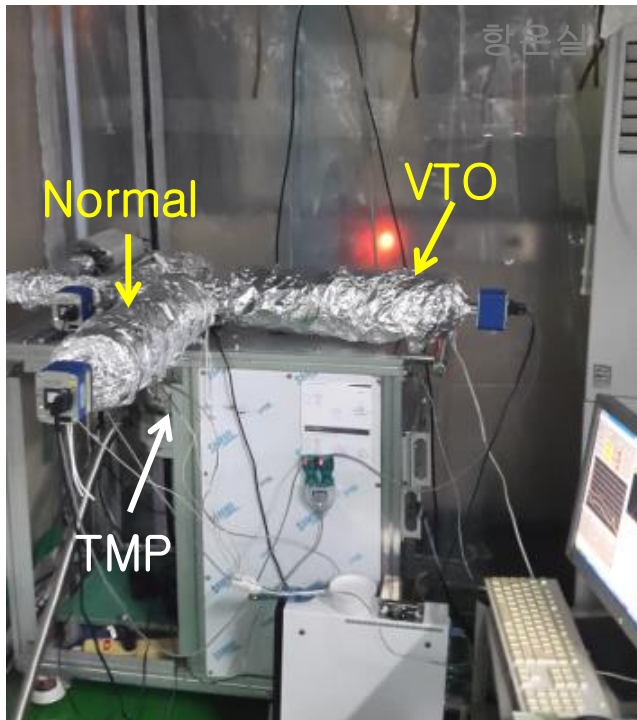
VTO (STS304)
:표면이 순수한 크롬산화물

기체방출률 비교



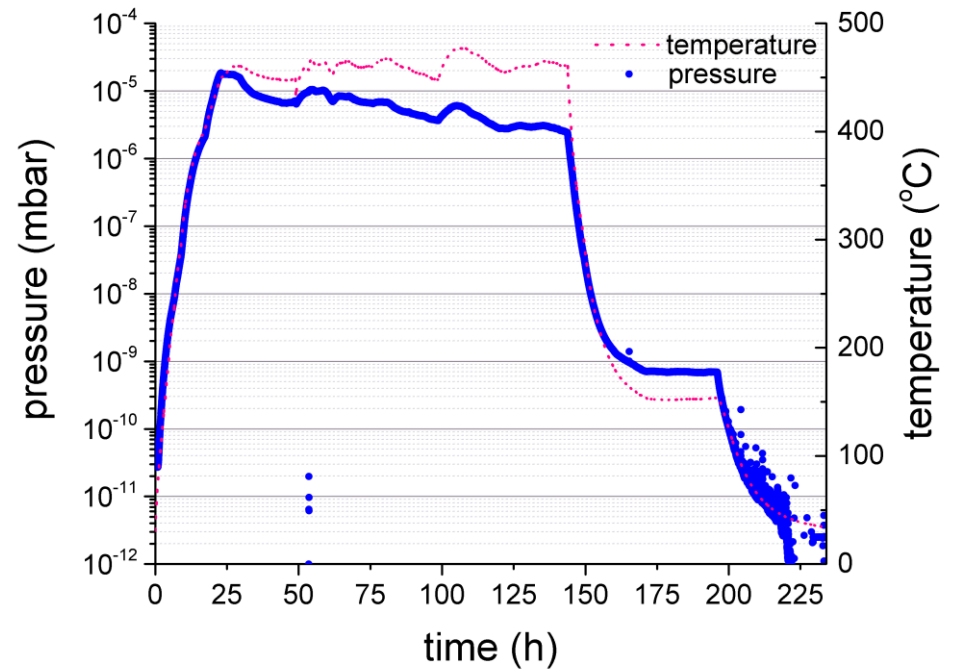
배기 시간 비교

- ✓ 동일한 STS304 진공용기(6 리터) 2 set 제작하여 1 set는 VTO 표면처리 함.
- ✓ 동시에 같은 조건으로 baking 후 질소 벤팅하여 5시간 유지한 이후 진공배기 함.
- ✓ 일반 STS304 진공용기와 VTO 처리한 진공용기의 배기시간 비교
: 고진공 ($5E-8$ mbar) 도달 시간이 일반 STS의 1/3



극고진공 챔버

- 1000 리터 급 극고진공 시스템: 상용 진공게이지의 측정 하한 도달.



- Literature survey
 - Do not believe those data
- Test and evaluation
- Make your own data